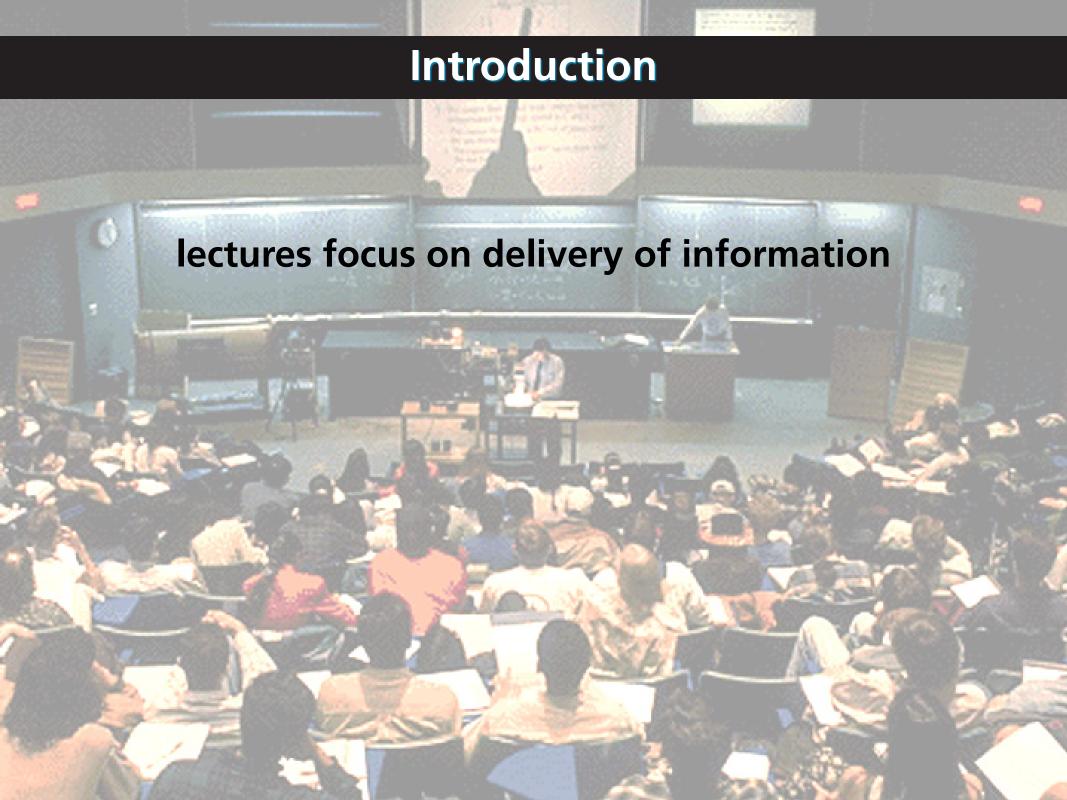
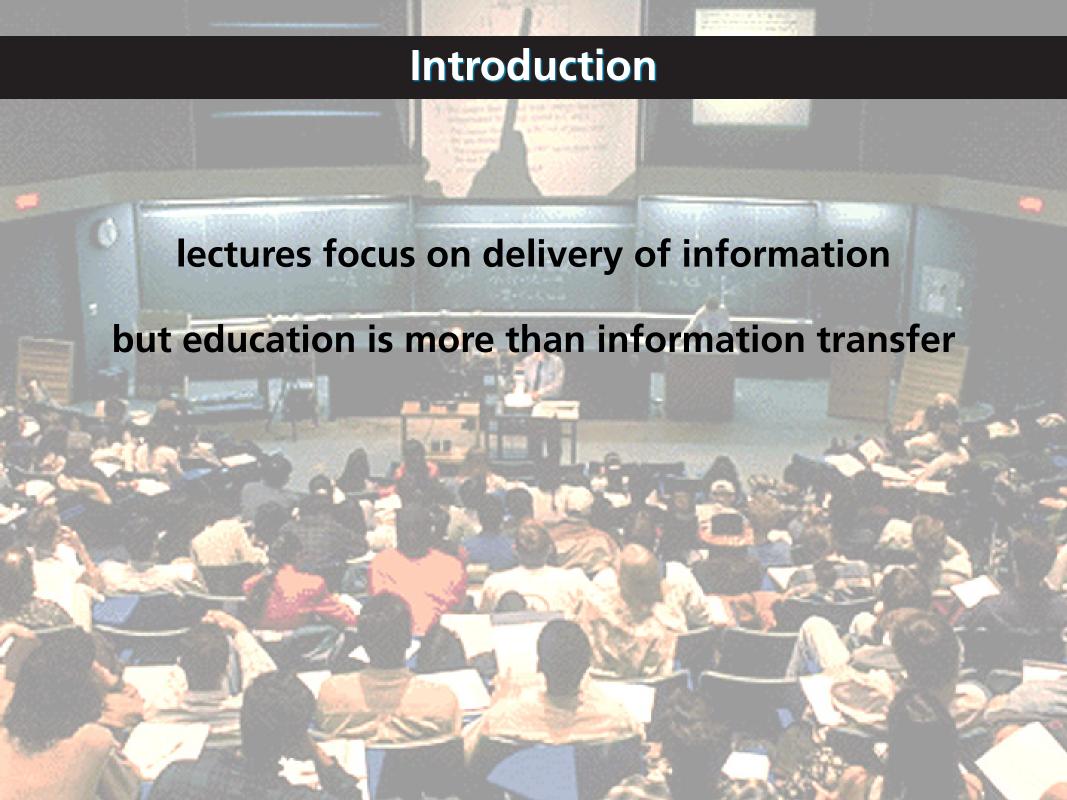
Peer Instruction: Why? And how?

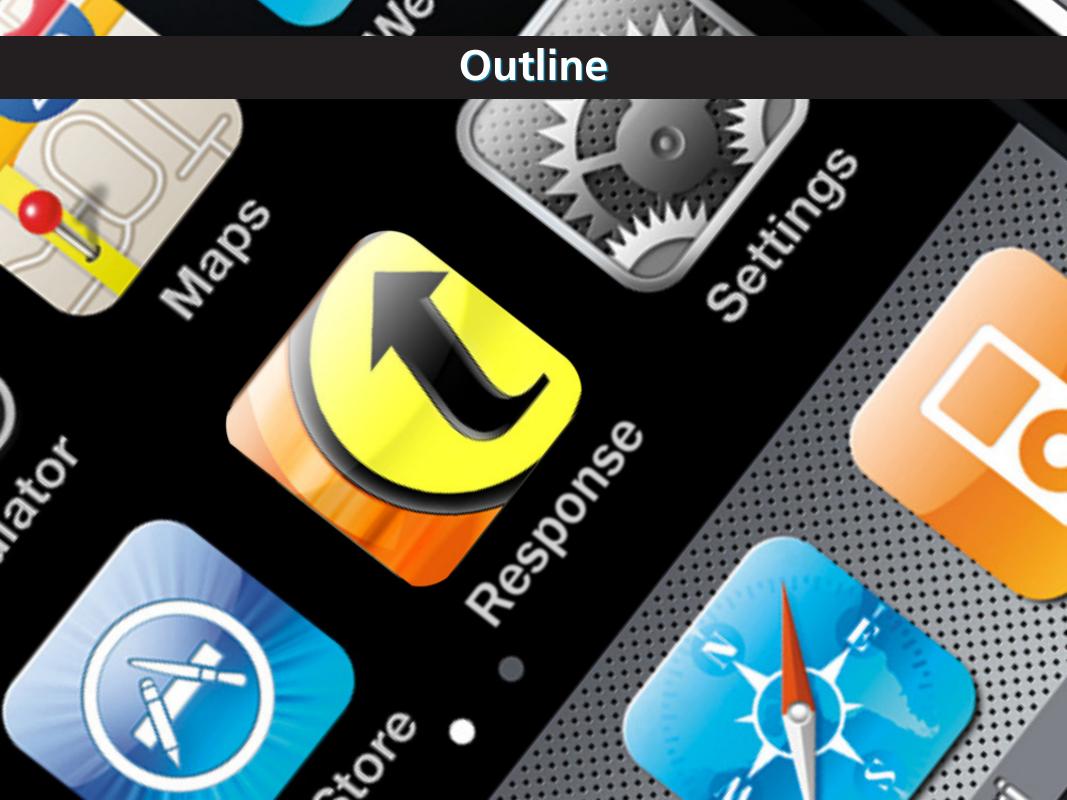


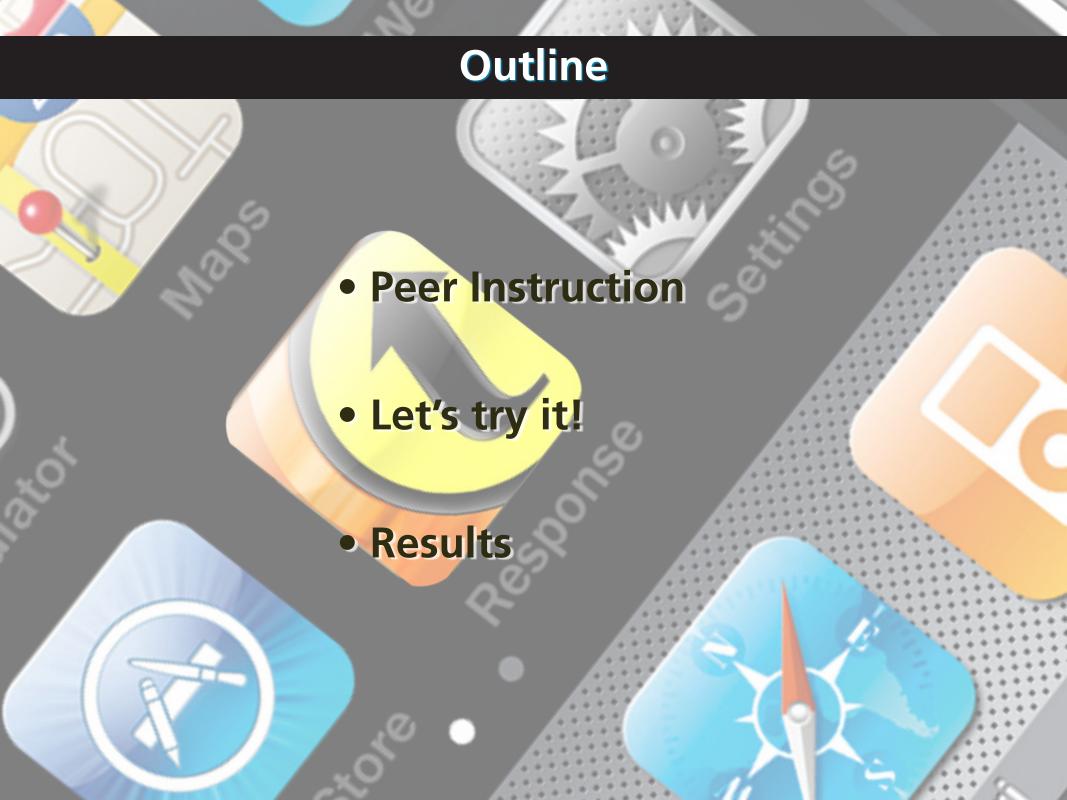


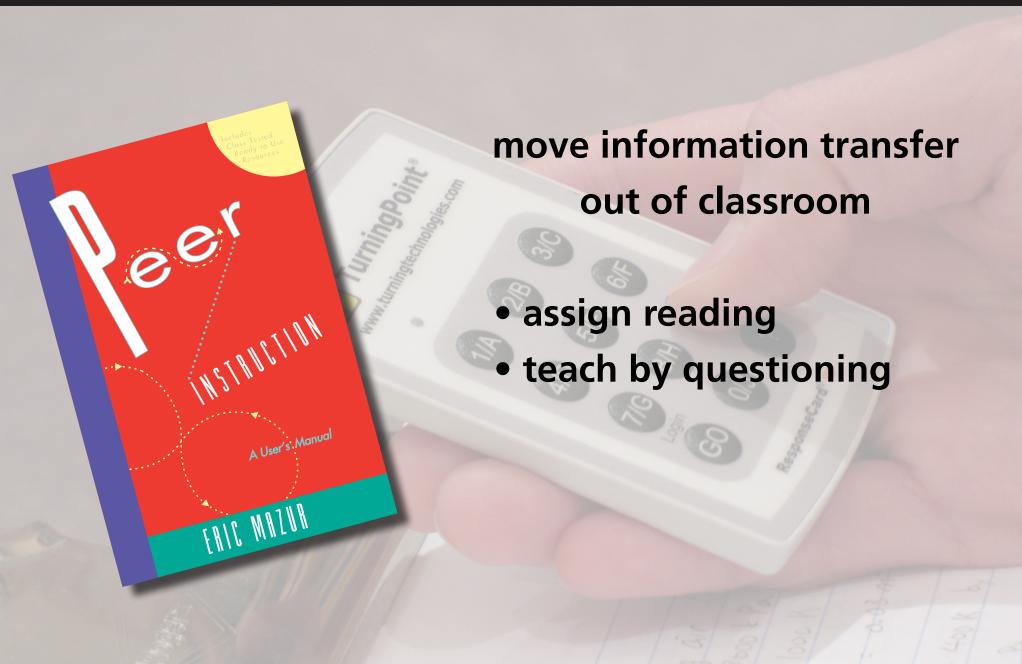
Introduction



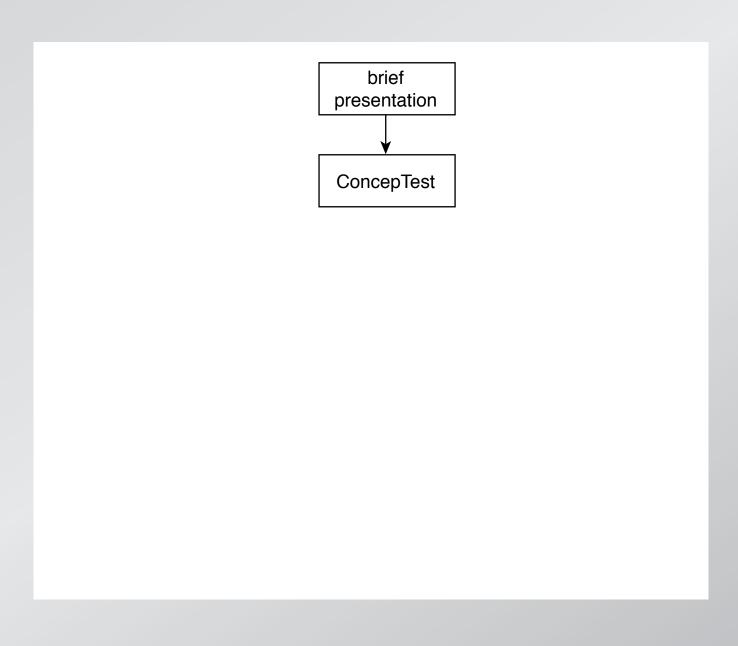


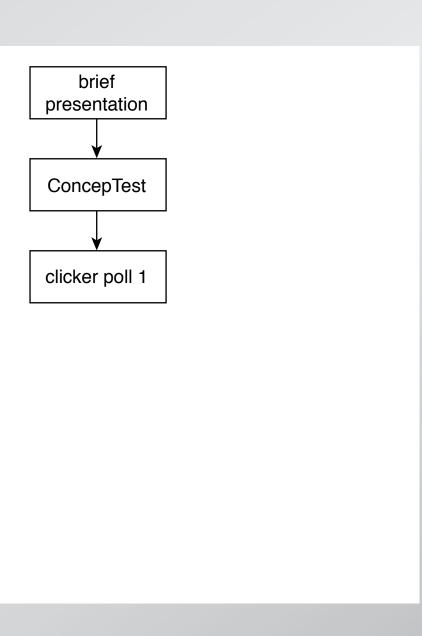


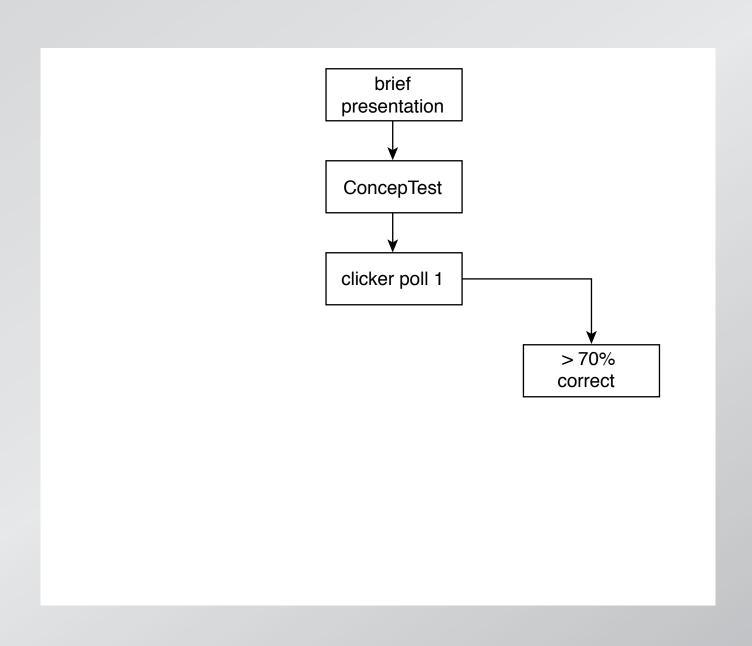


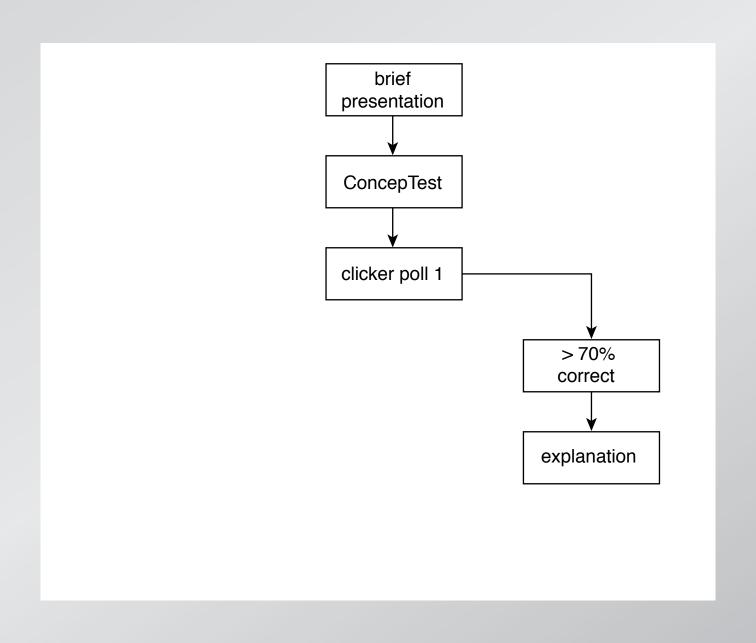


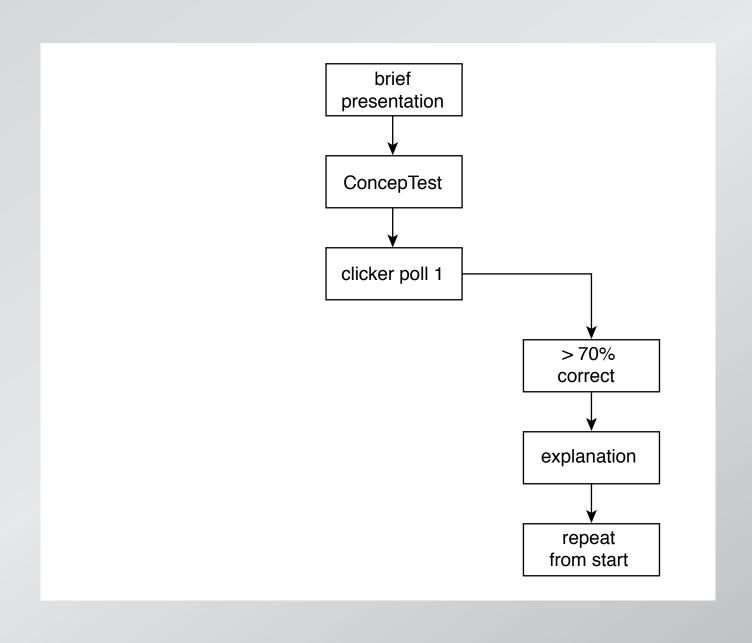
brief presentation

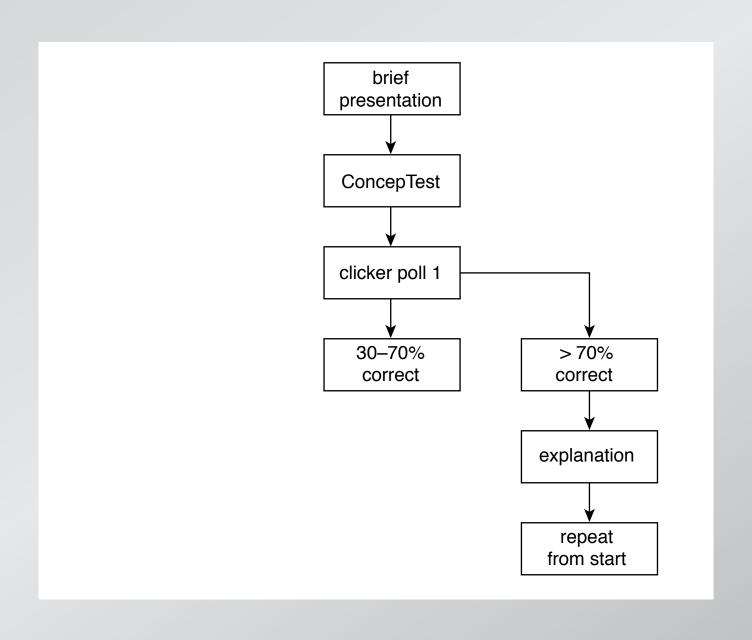


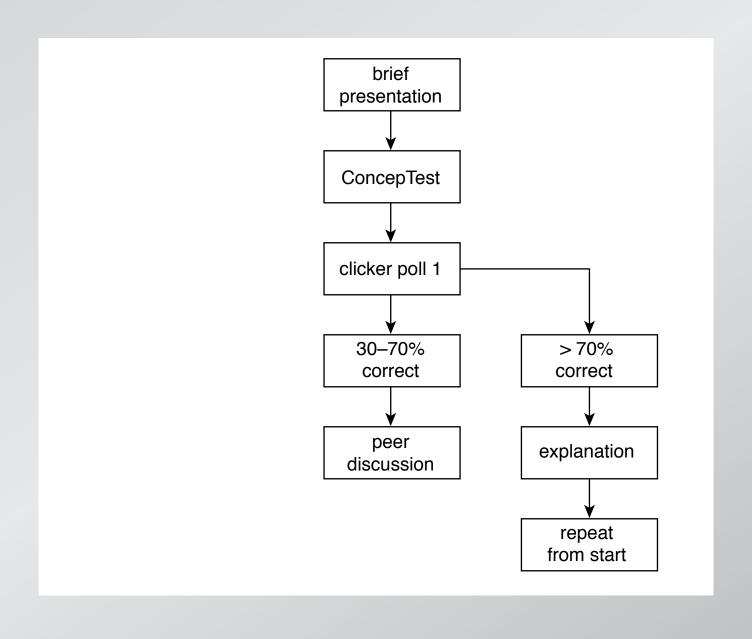


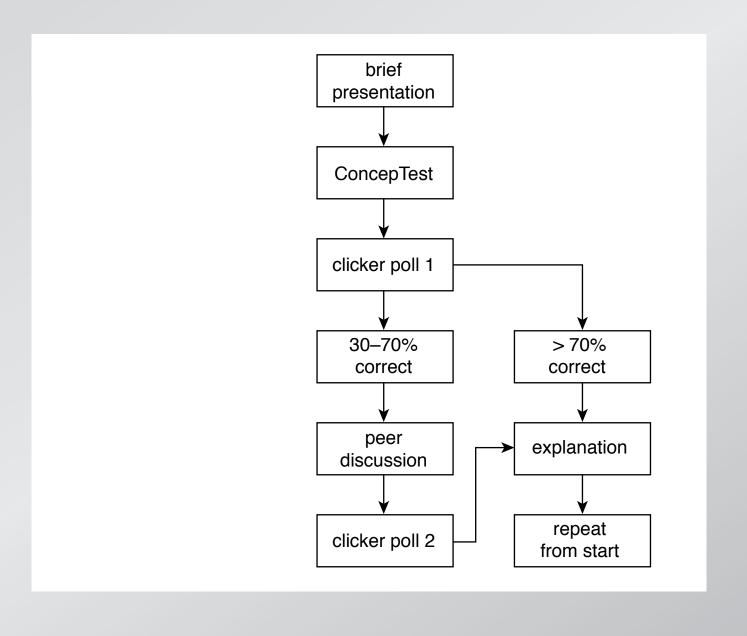


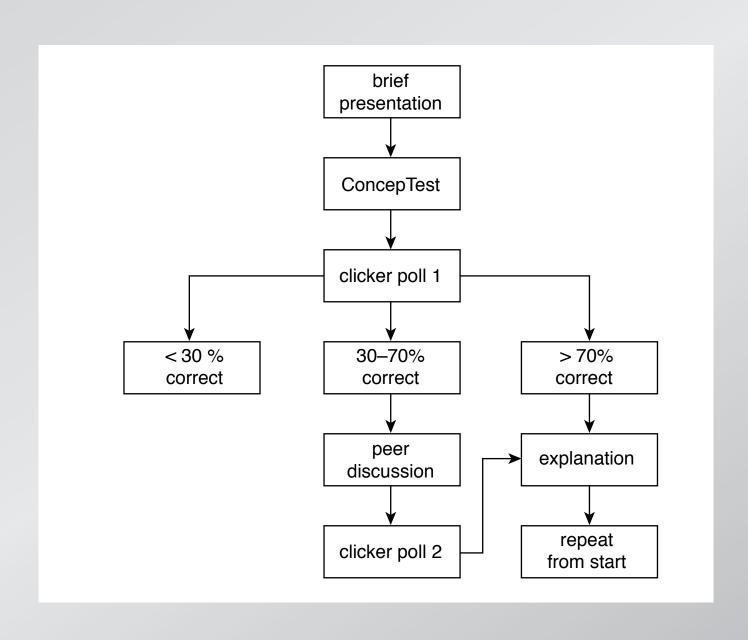


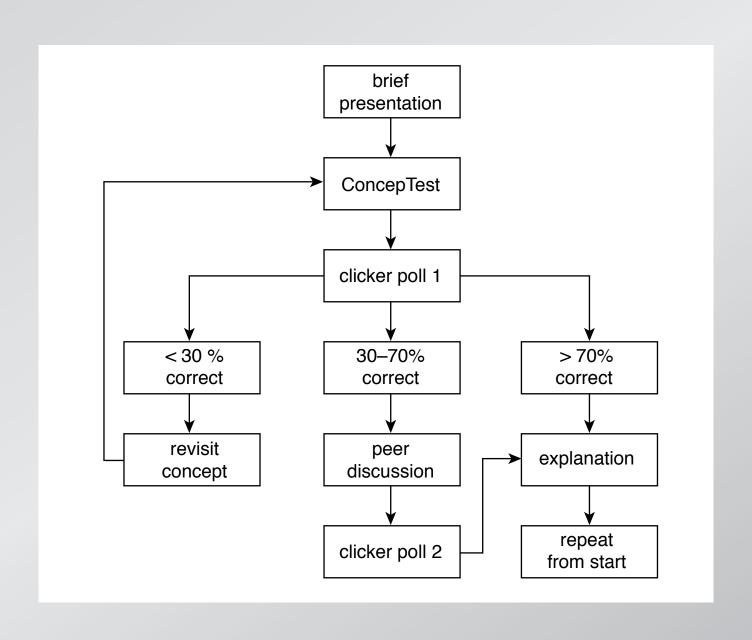


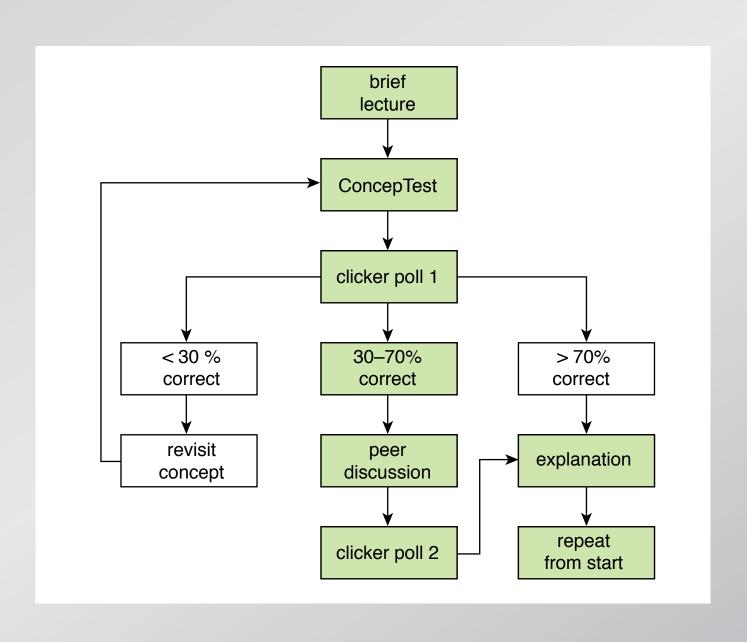


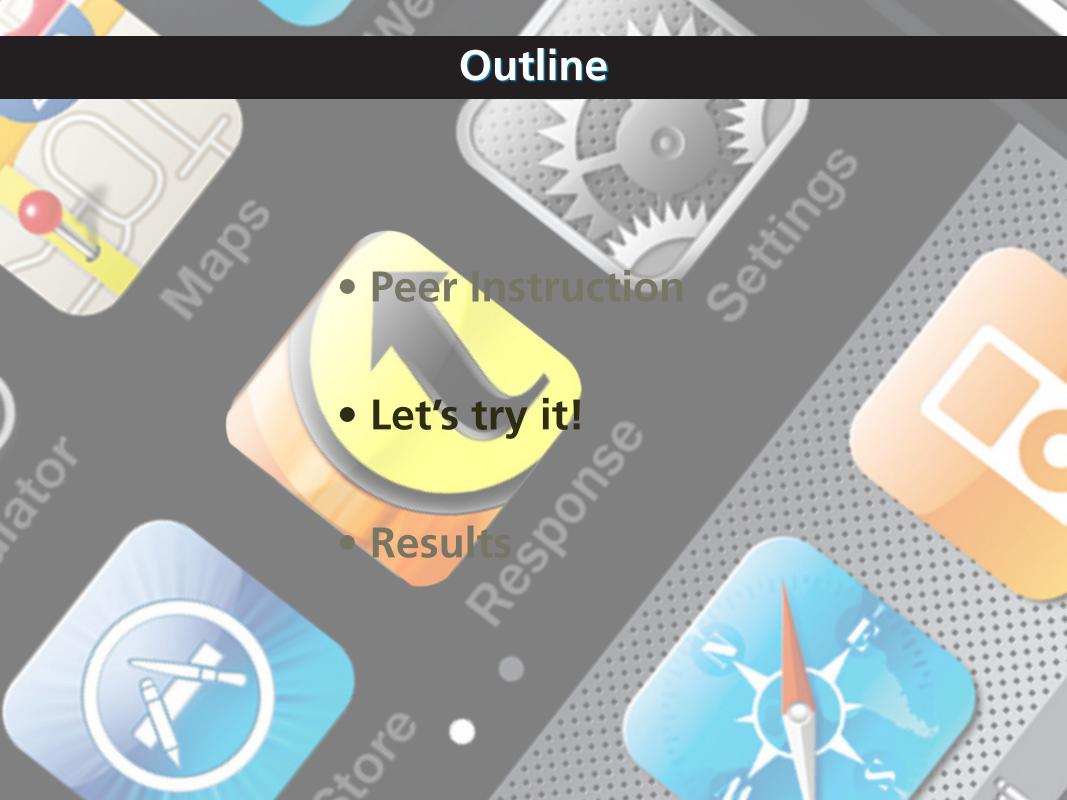












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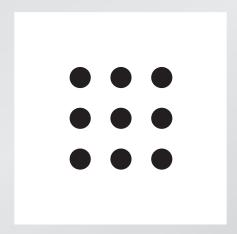


unique ID on back of clicker

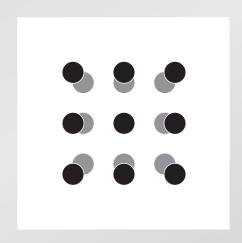
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When metals heat up, they expand because all atoms get farther away from each other.

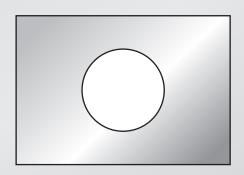
When metals heat up, they expand because all atoms get farther away from each other.



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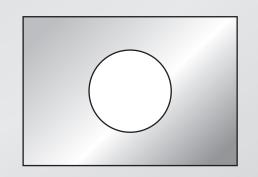
Consider a rectangular metal plate with a circular hole in it.



Consider a rectangular metal plate with a circular hole in it.

When the plate is uniformly heated, the diameter of the hole

- 1. increases.
- 2. stays the same.
- 3. decreases.



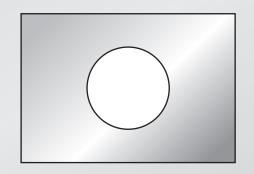


It's easy to fire up the audience!

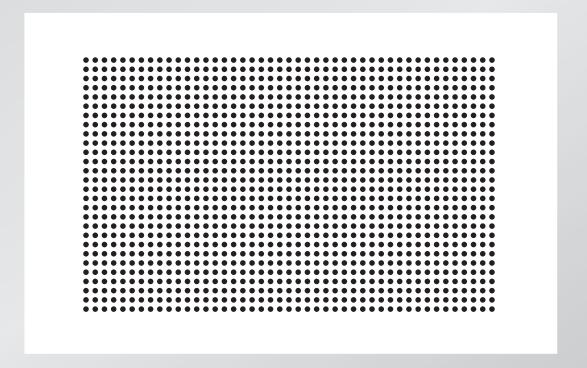
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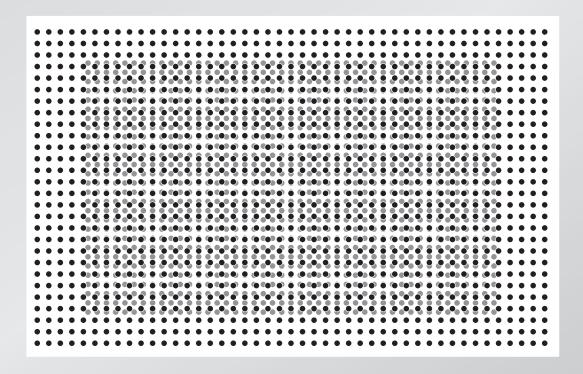
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- 2. stays the same.
- 3. decreases.



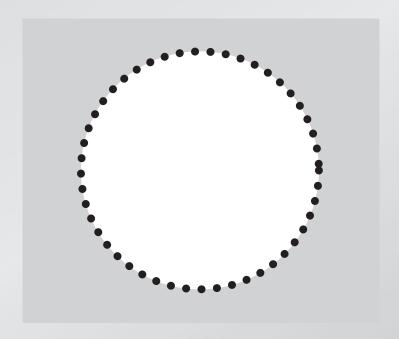
remember: all atoms must get farther away from each other!



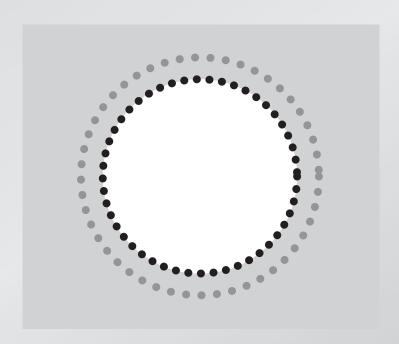
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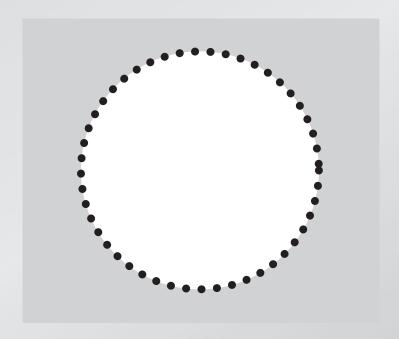
consider the atoms at the rim of the hole



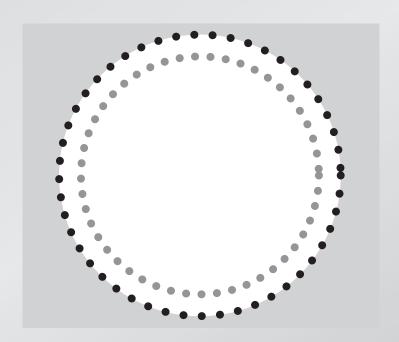
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consider the atoms at the rim of the hole



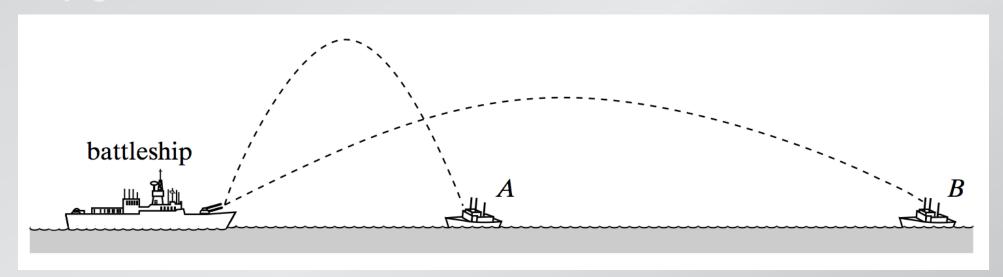
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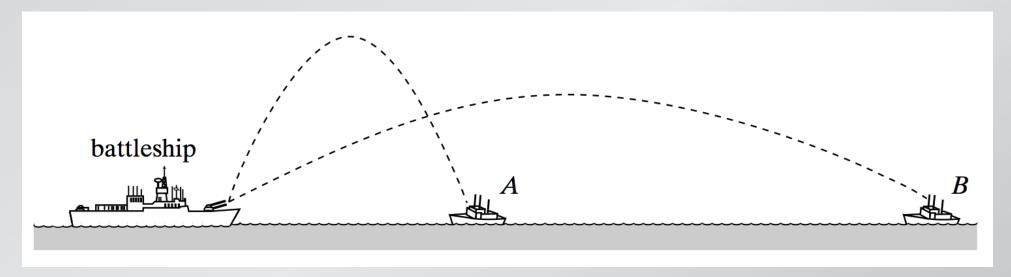
projectile trajectories:

time up = time down horizontal/vertical motion decoupled

A battleship simultaneously fires two shells at enemy ships. If the shells follow the parabolic trajectories shown below, which ship gets hit first?



A battleship simultaneously fires two shells at enemy ships. If the shells follow the parabolic trajectories shown below, which ship gets hit first?

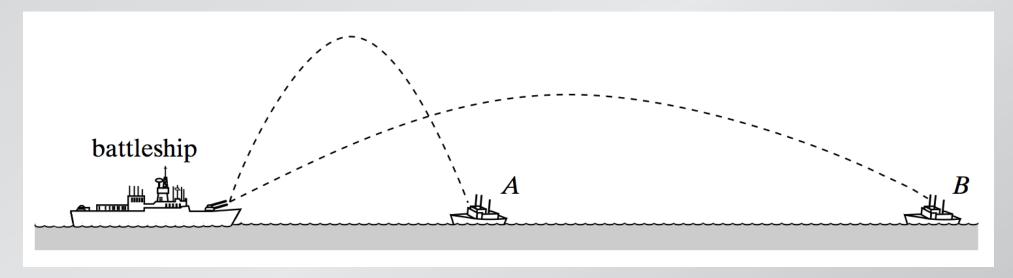


- 1. A
- 2. both at (nearly) the same time
- 3. B
- 4. need more information



You all got worked up again!

A battleship simultaneously fires two shells at enemy ships. If the shells follow the parabolic trajectories shown below, which ship gets hit first?



- 1. A
- 2. both at (nearly) the same time
- 3. B 🗸
- 4. need more information

time projectile spends in air:

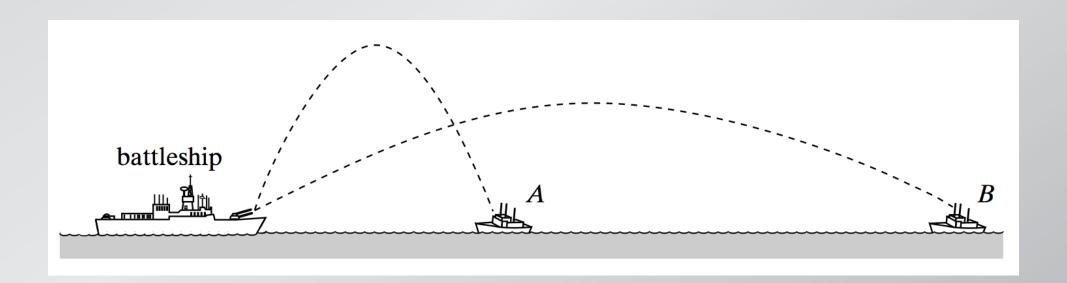
 $2t_{\rm h}$

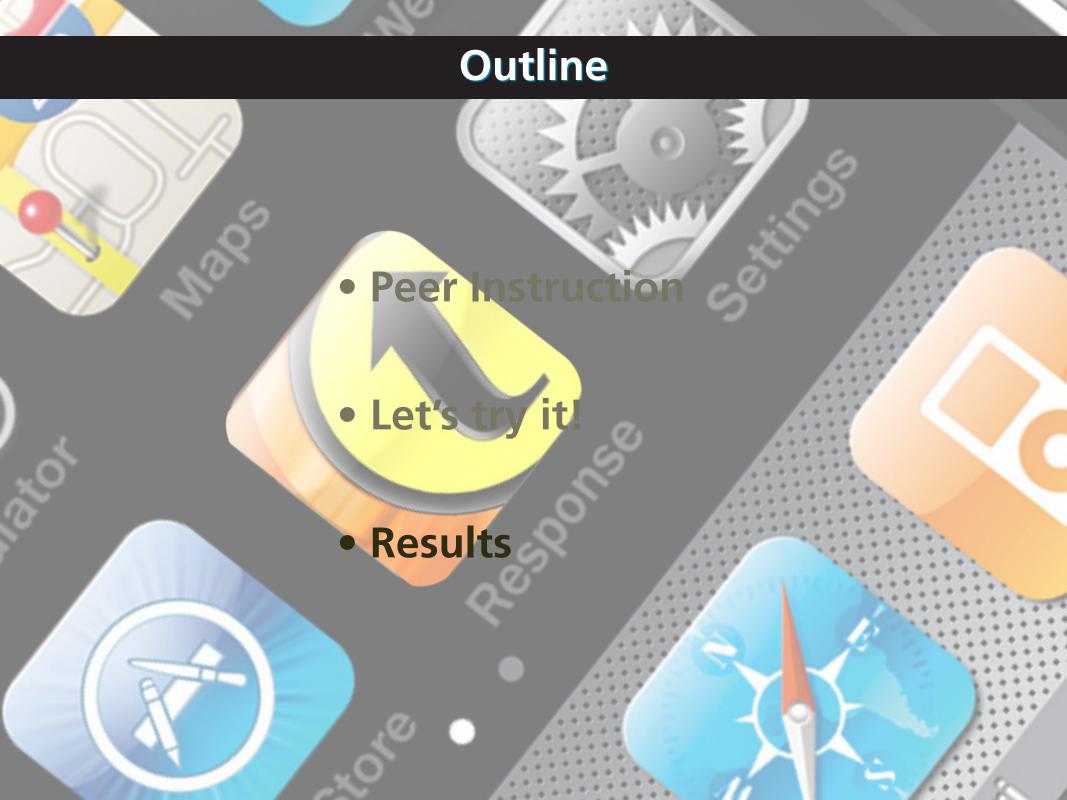
(twice time to fall from maximum height)

time projectile spends in air:

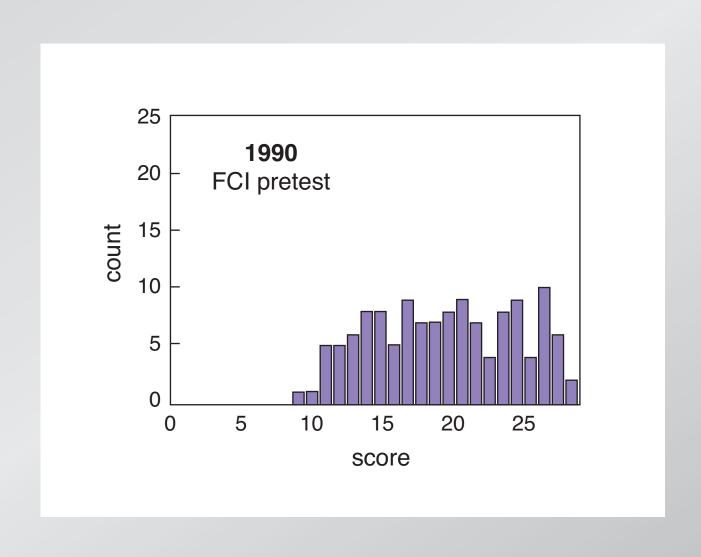
 $2t_{\rm h}$

(twice time to fall from maximum height)

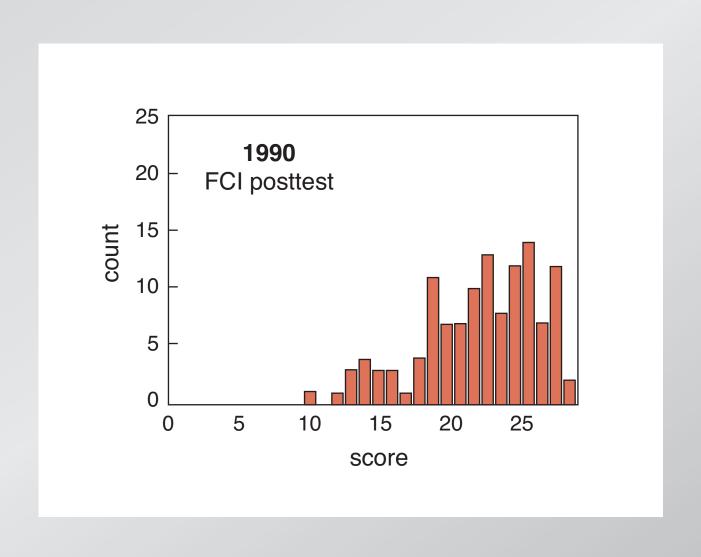




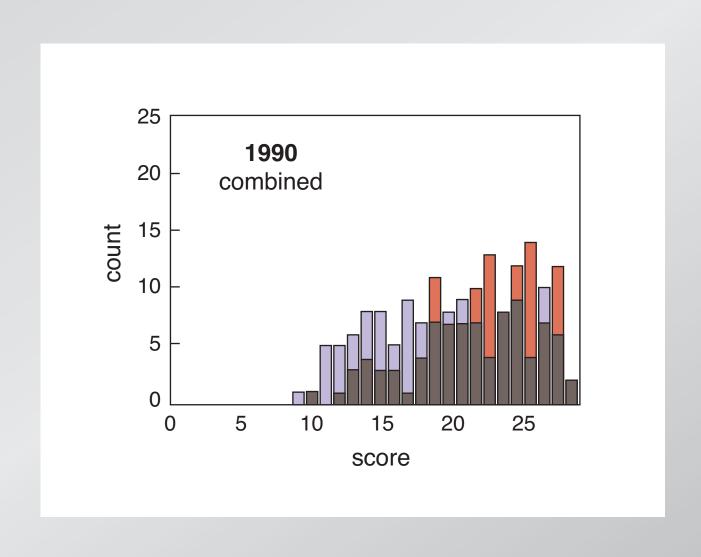
traditional instruction



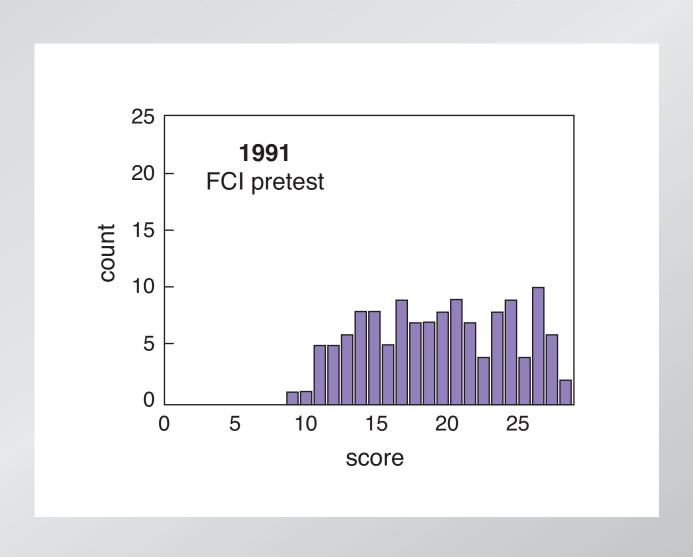
traditional instruction



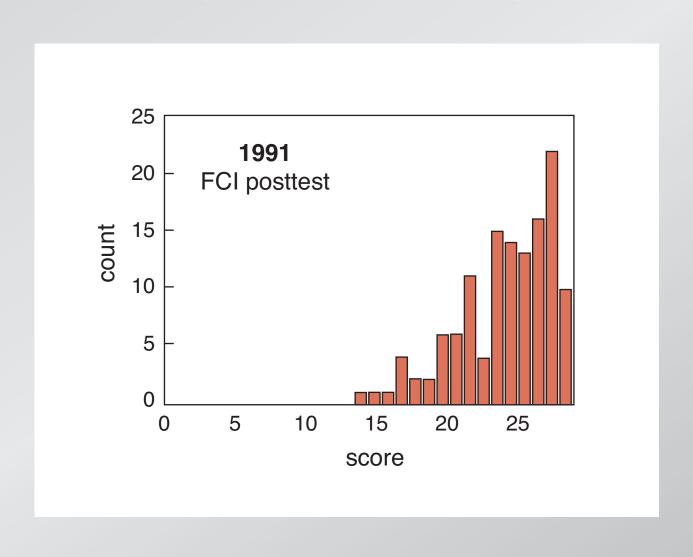
traditional instruction



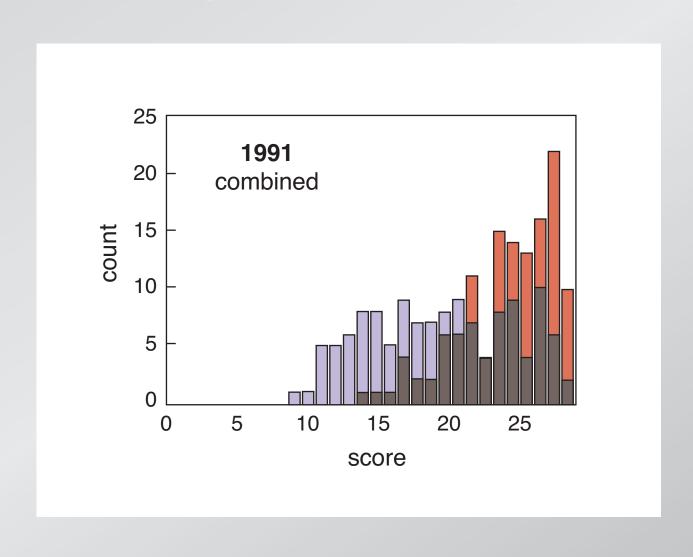
first year of implementing PI



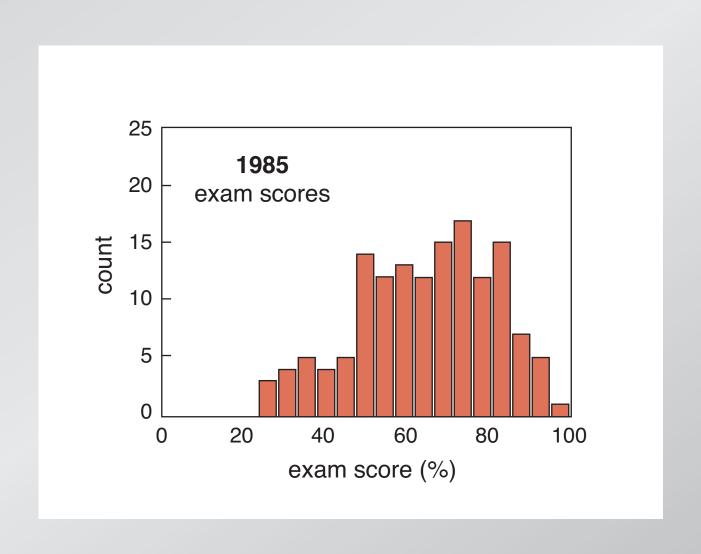
first year of implementing PI

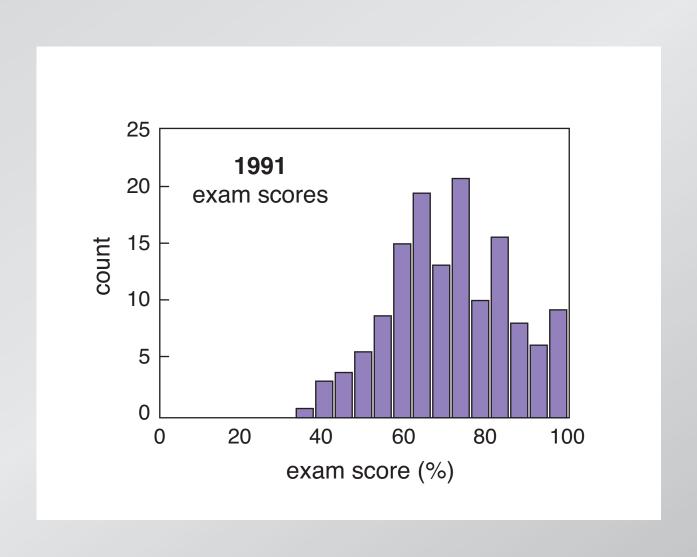


first year of implementing PI

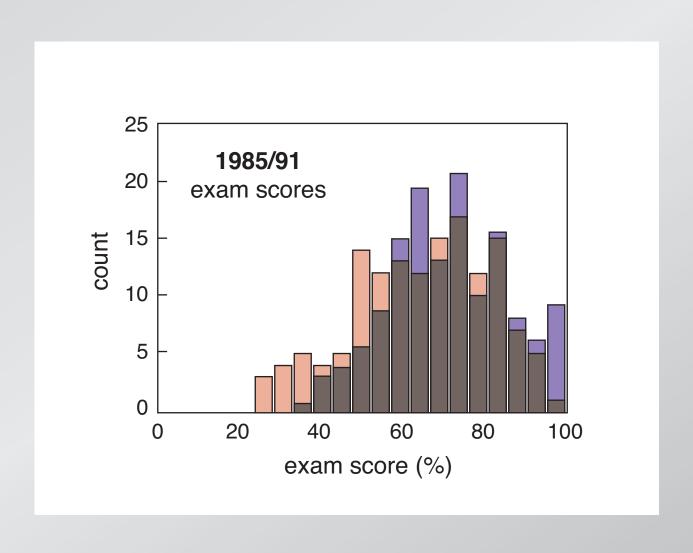


what about problem solving?





better understanding leads to better problem solving



Conclusion active engagement greatly improves learning gains

Research Funding:

Pew Charitable Trust, Pearson/Prentice Hall, Davis Foundation, Engineering Information Foundation, Derek Bok Center for Teaching and Learning, National Science Foundation

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Peer Instruction: Why? And how?





How do we learn?

Think of something you are really good at — something that you know you do well.

How do we learn?

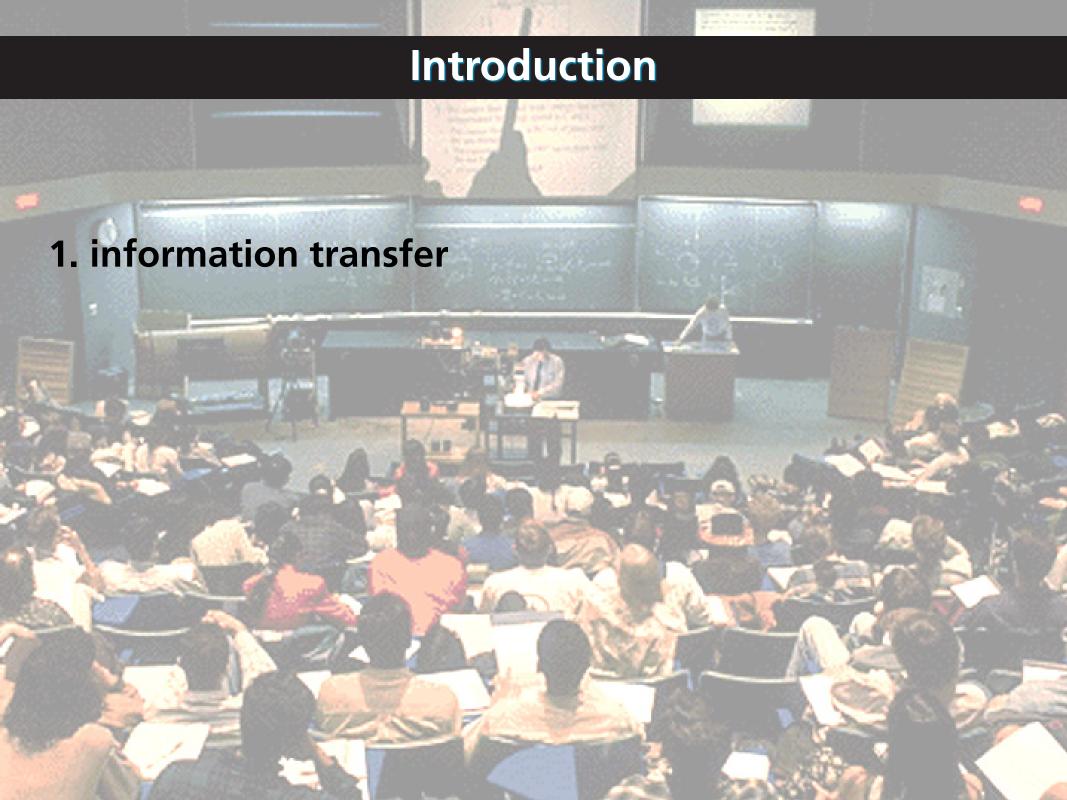
Think of something you are really good at — something that you know you do well.

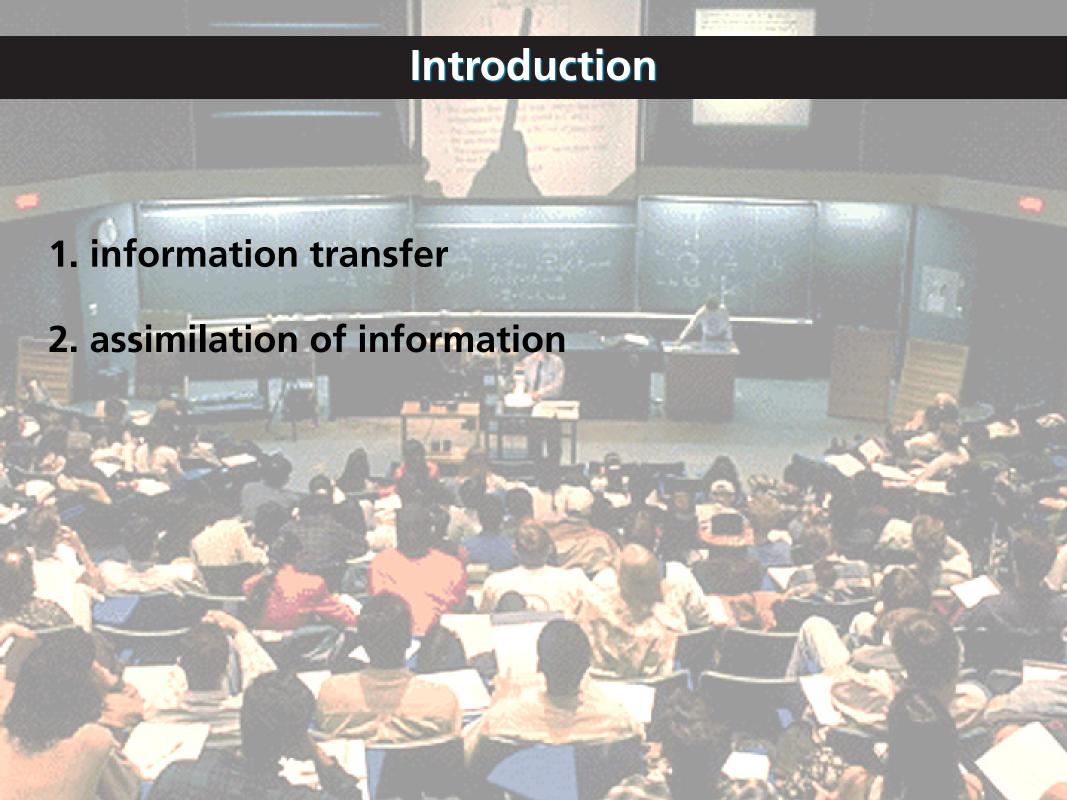
How did you become good at this?

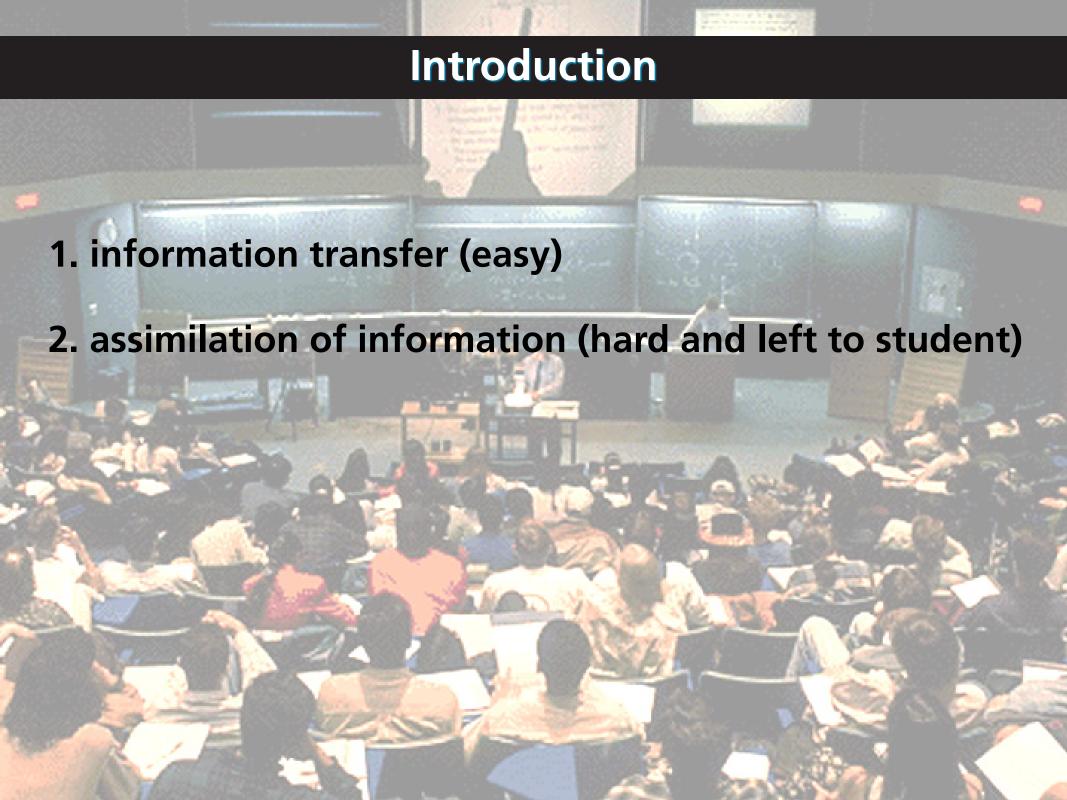
How do we learn?

I learned it:

- A. by practicing
- **B.** in lectures
- C. by doing it (trial and error)
- D. other







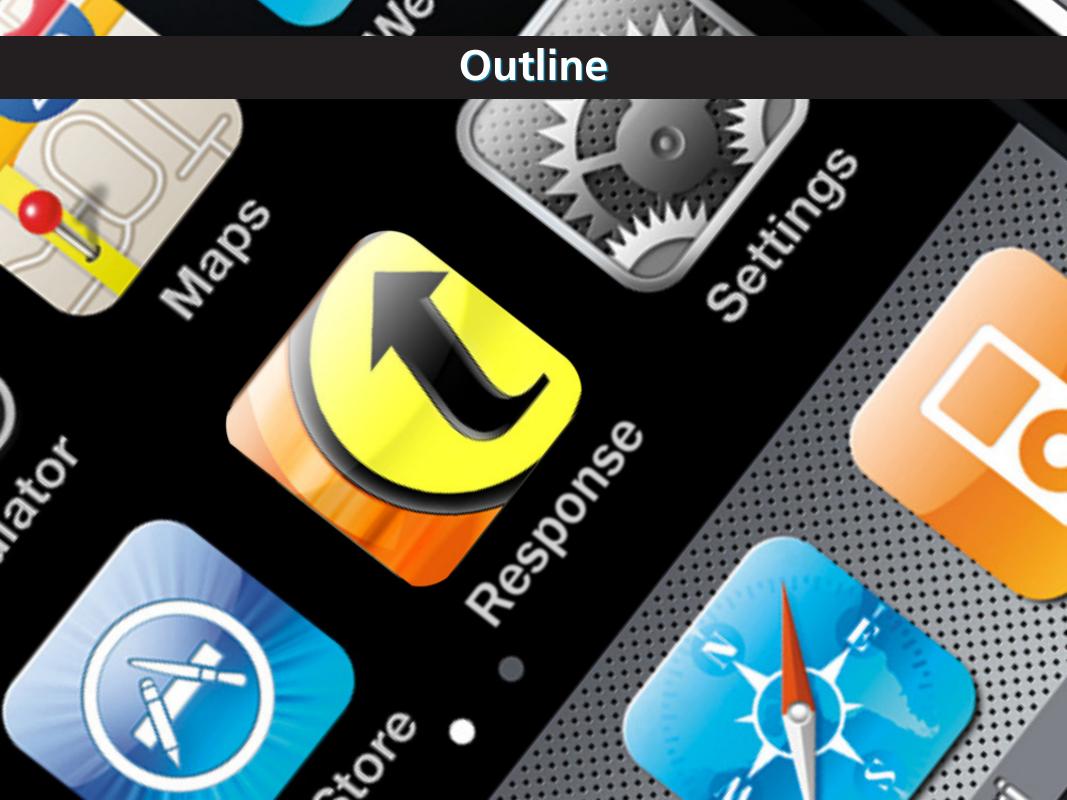
Introduction

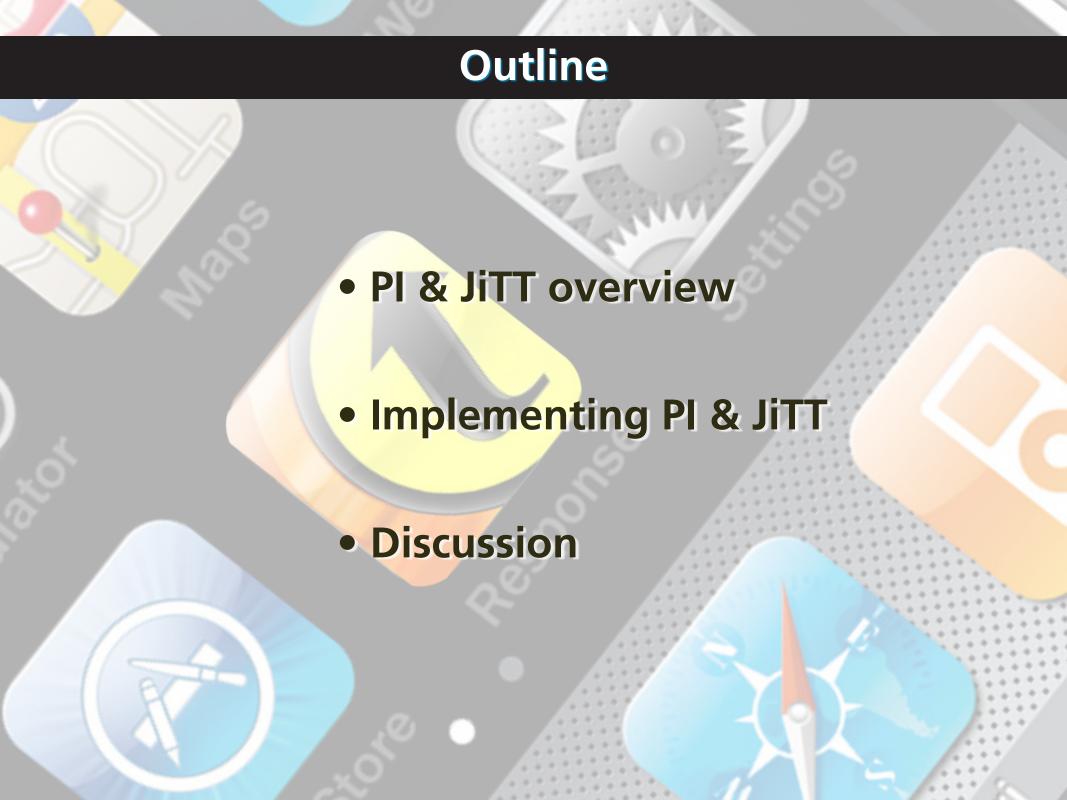
Solution: move information transfer out of classroom!

Introduction

Solution: move information transfer out of classroom!

(so we can help students assimilate the information in class)





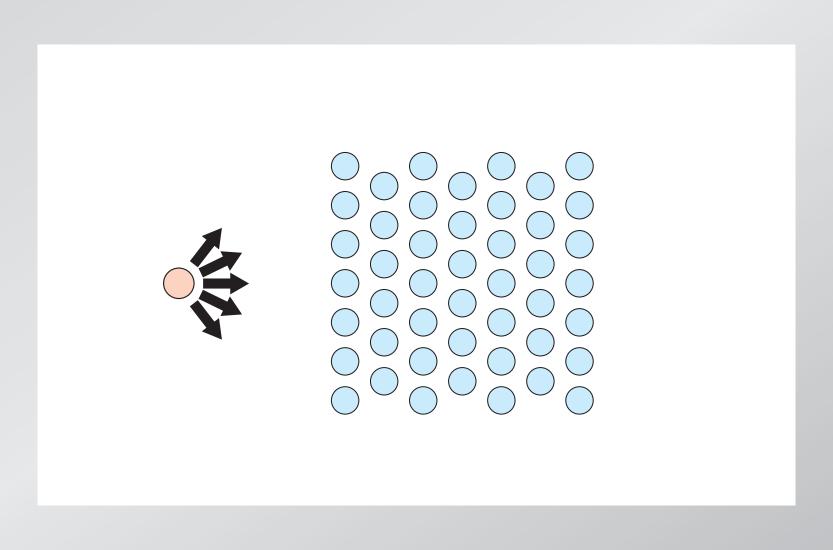
Implementing PI & JiTT

"I am very interested to learn more about the systematic use of peer instruction"

"Can peer instruction be organized in traditional courses? and how?"

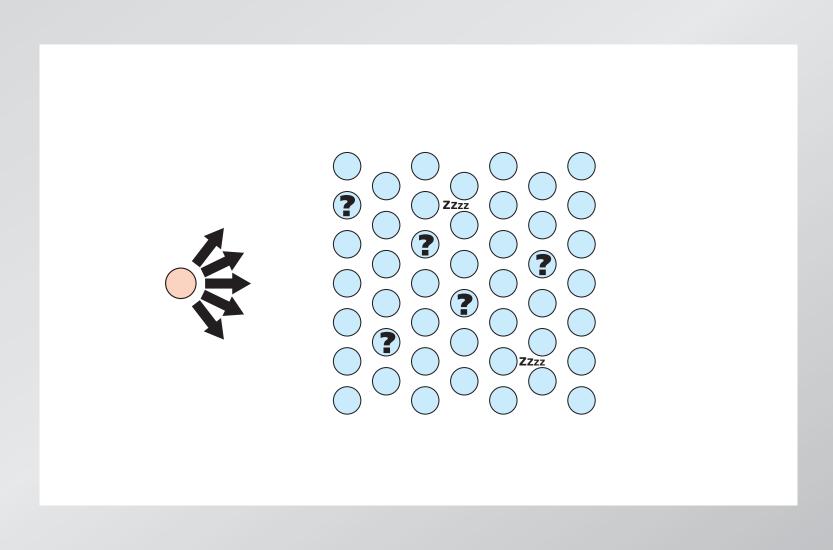
PI & JiTT Overview

traditional approach: one-way flow of information

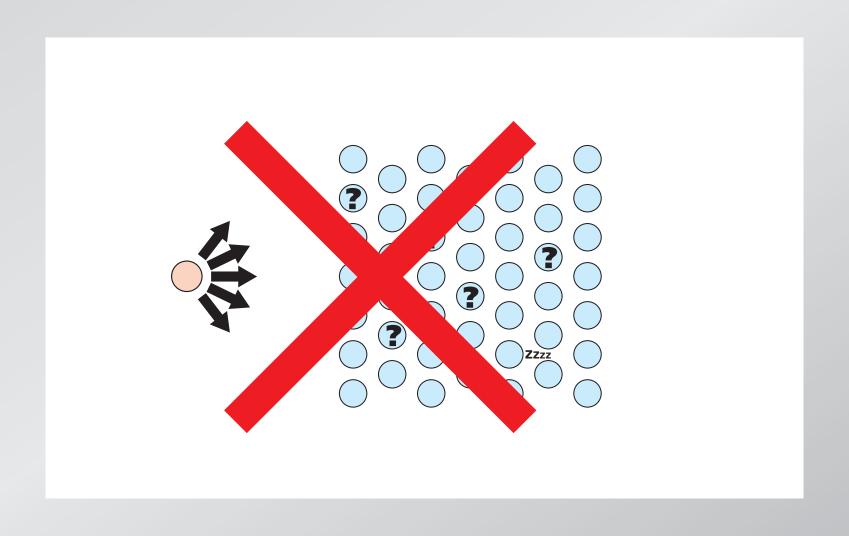


PI & JiTT Overview

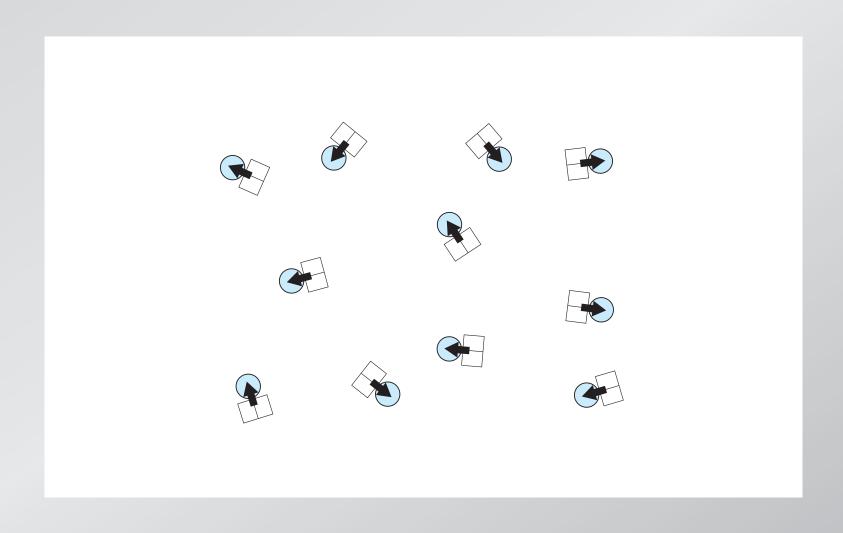
traditional approach: one-way flow of information



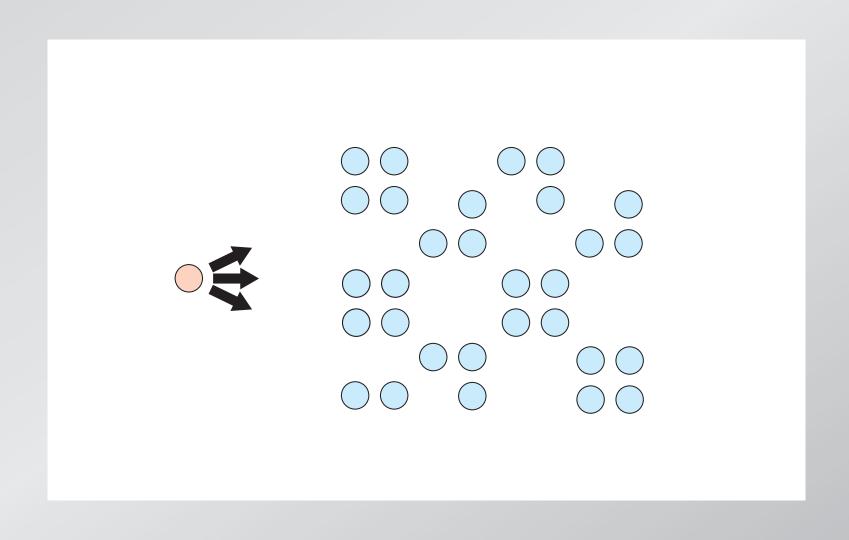
no learning



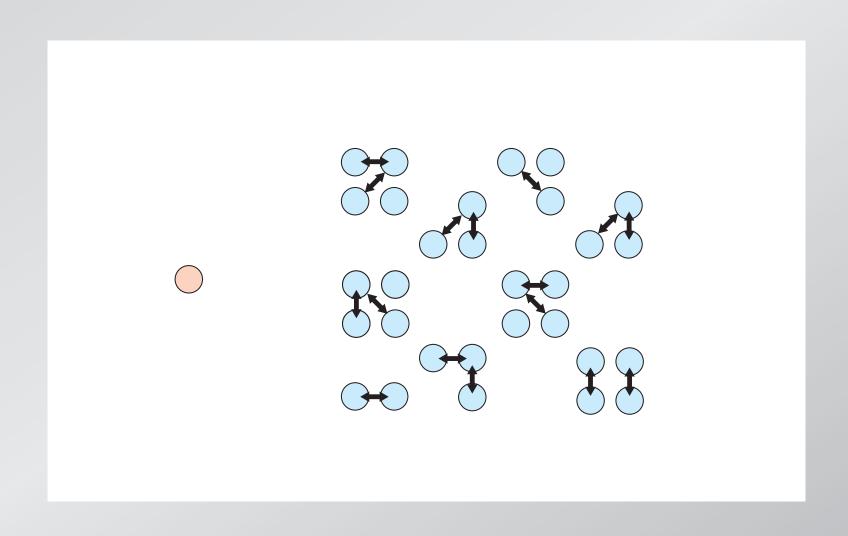
instead: assign reading (JiTT)



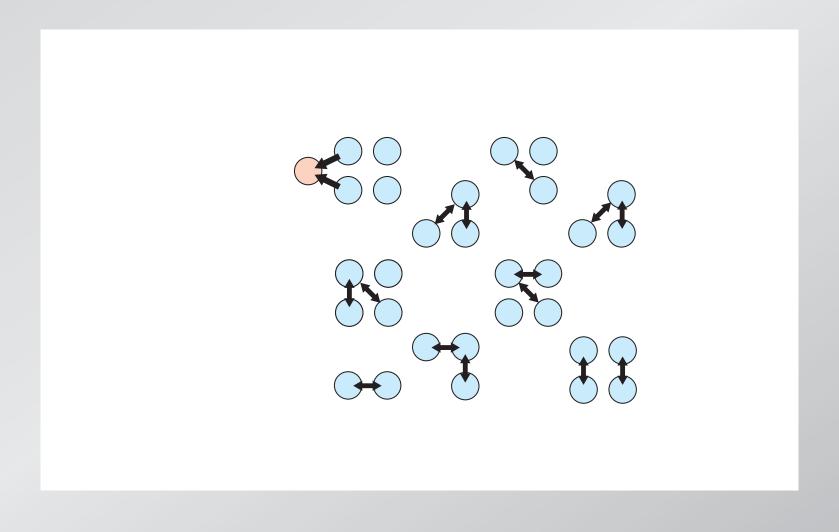
lecture a little...



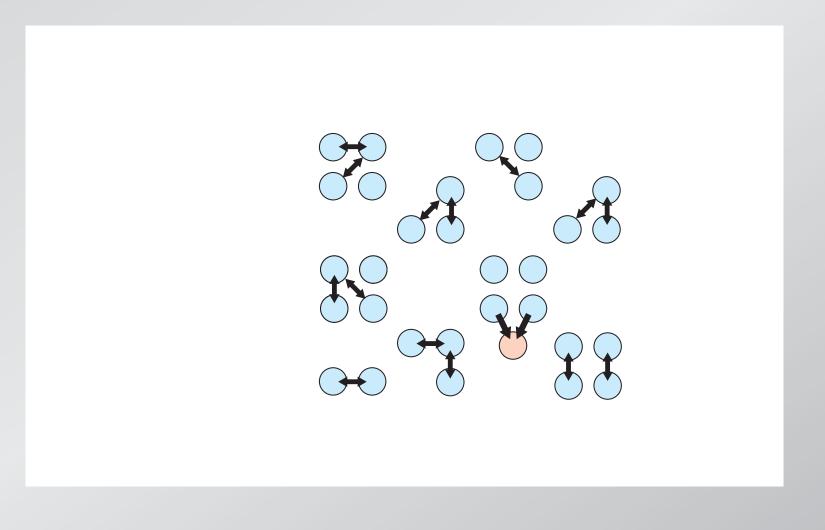
...and have students discuss and learn (PI)



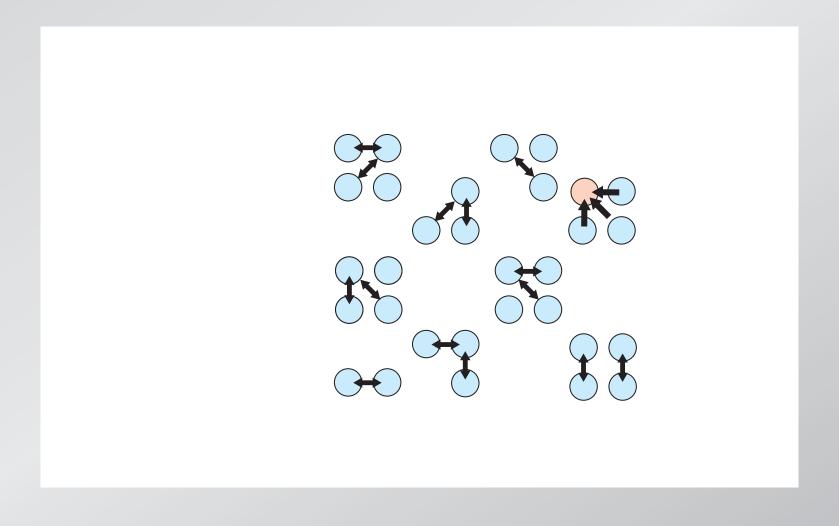
information flows two ways



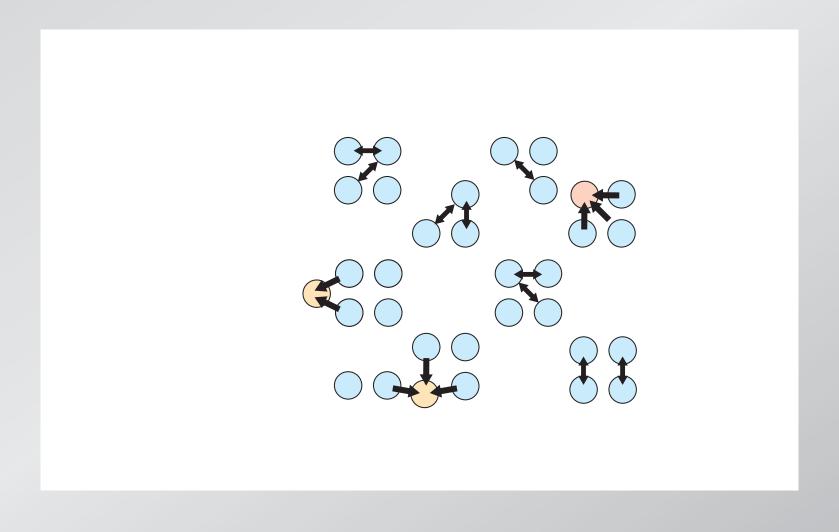
information flows two ways



information flows two ways



(bring in your assistants too!)

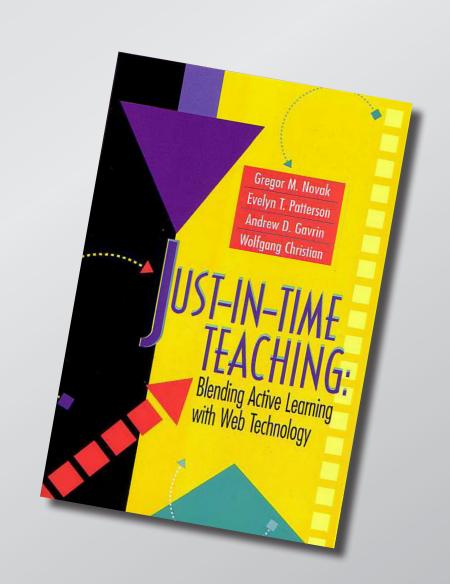


"[How do I] prepare learning materials in a way that allows students to self study and prepare beforehand to have efficient communication in the class?"

Just-in-time-Teaching (JiTT)

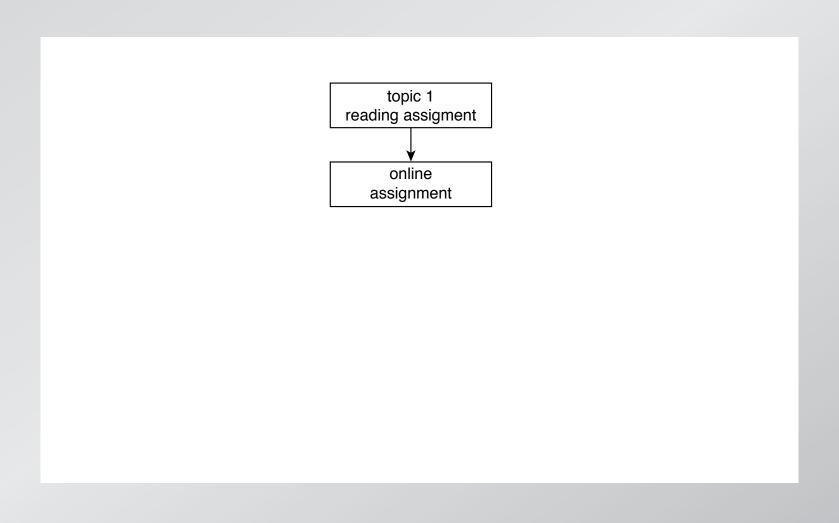
www.jitt.org

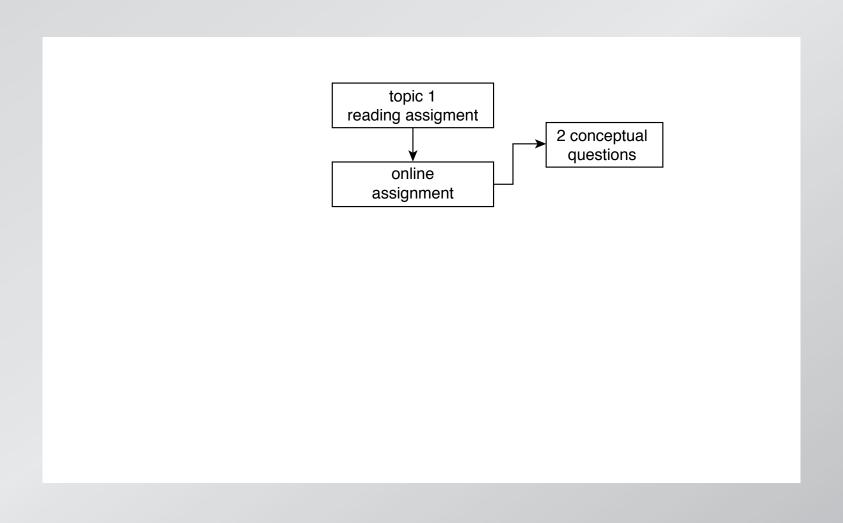


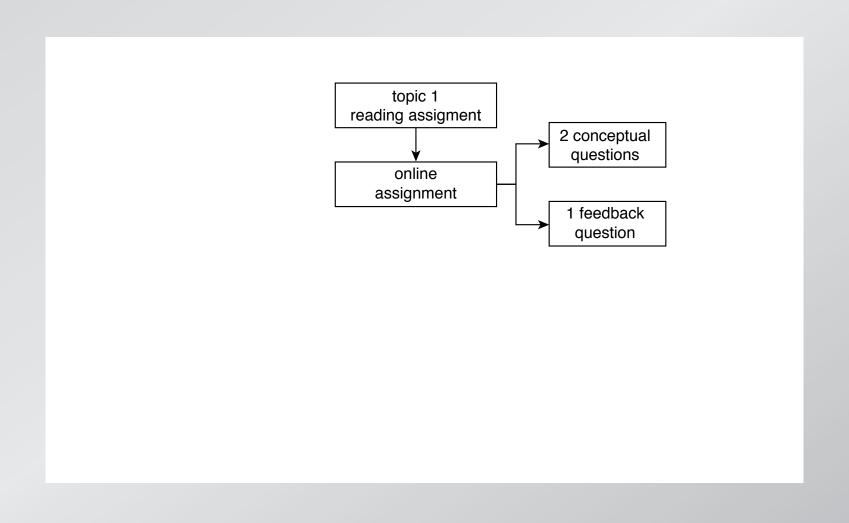


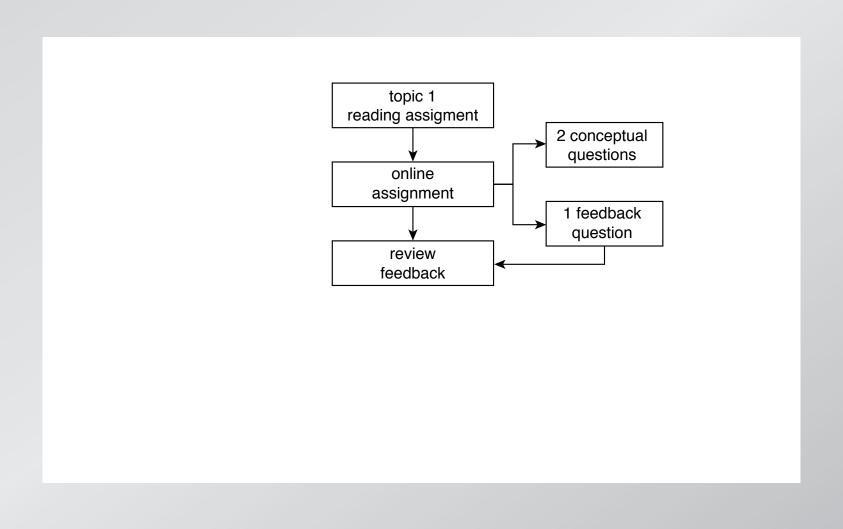
JiTT workflow

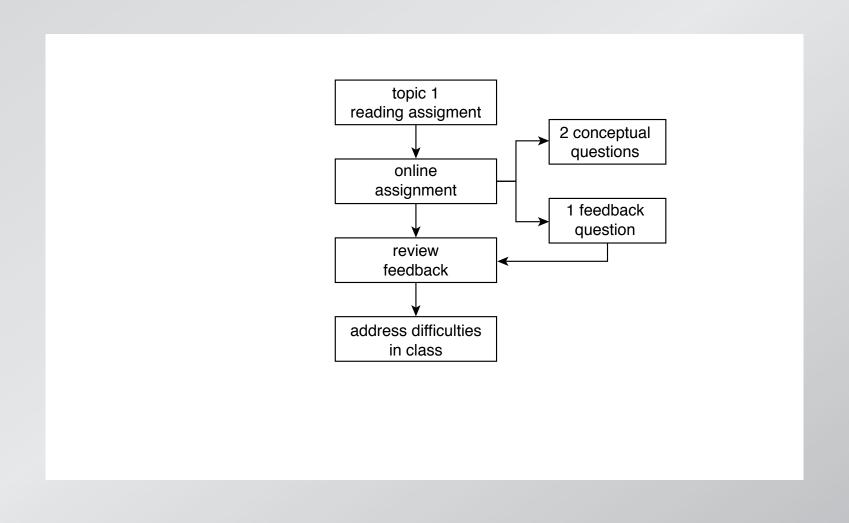
topic 1 reading assigment

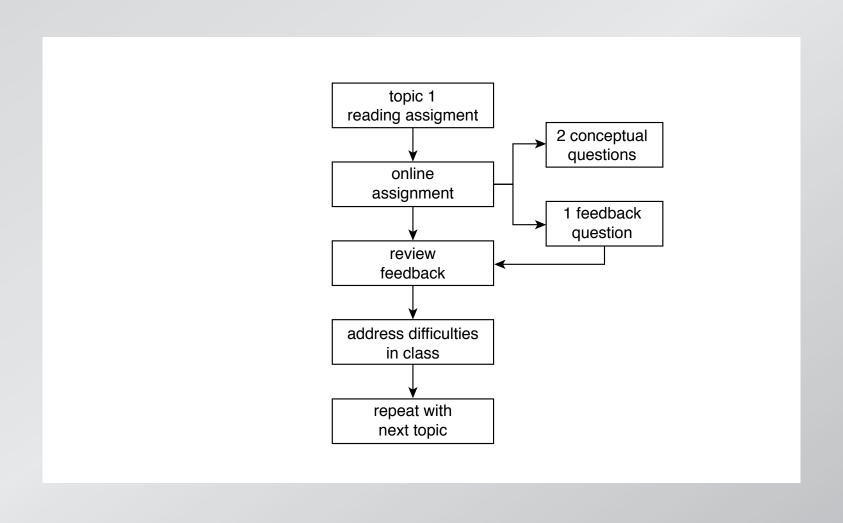








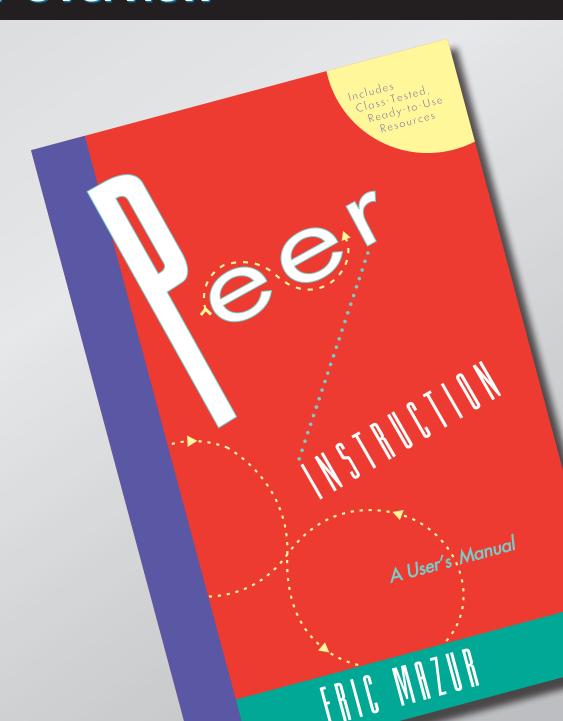


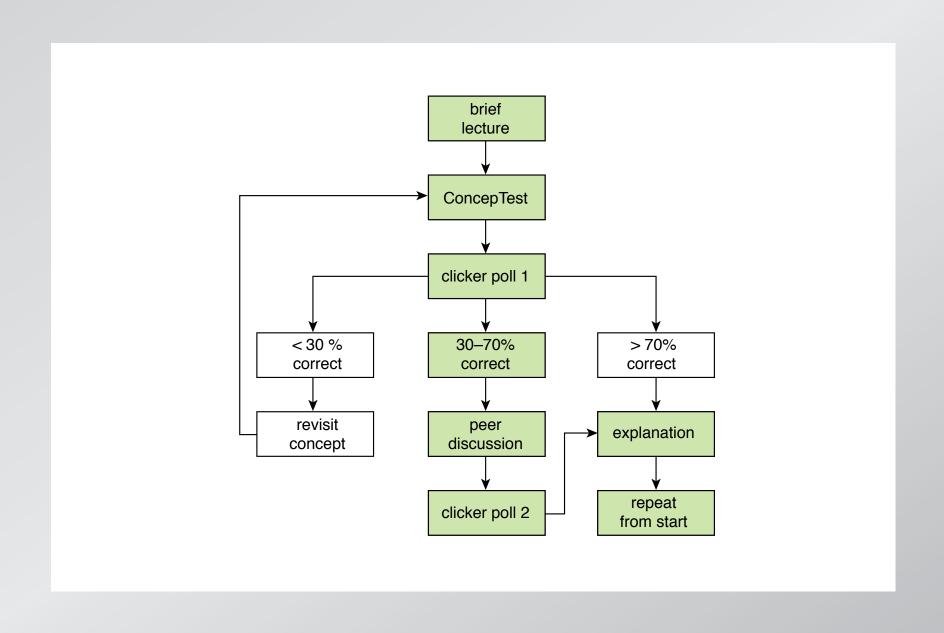


JiTT:

- prepares you for class
- prepares students for class
- helps you address student difficulties

Peer Instruction (PI)





PI:

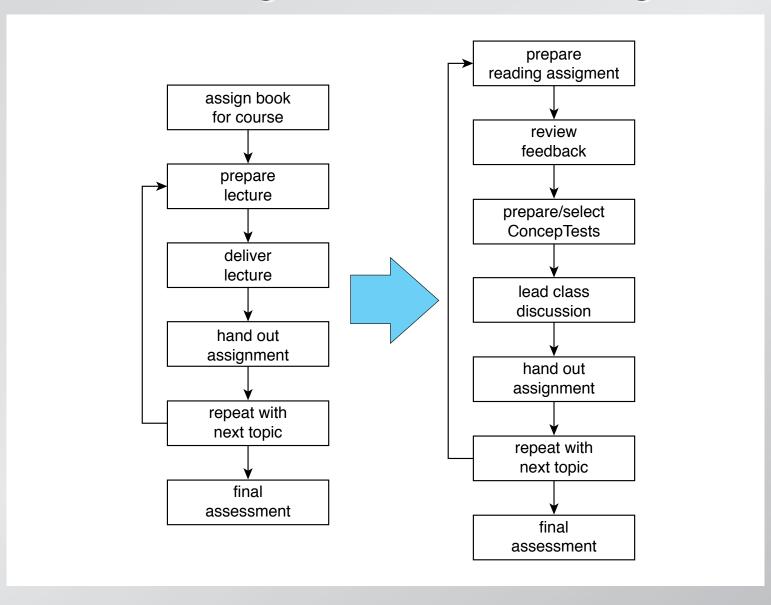
- helps students overcome difficulties
- encourages deep learning
- provides depth, not "coverage"
- helps you become aware of misconceptions

Benefits:

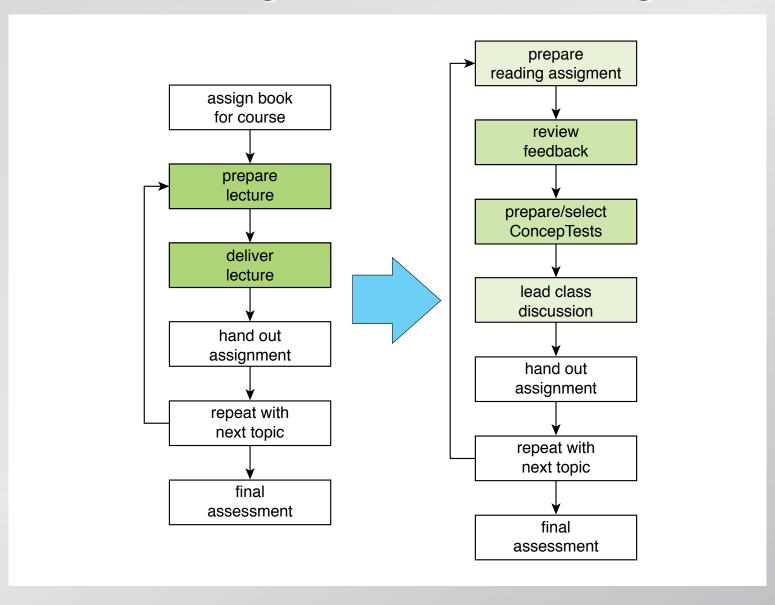
- helps develop conceptual models
- solidifies understanding
- provides feedback
- empowers students



transitioning: where does the effort go?



transitioning: where does the effort go?



"I very much like the approach, but think its hard to formulate ConcepTests"

What constitutes a good problem?

On a Saturday afternoon, you pull into a parking lot with unmetered spaces near a shopping area. You circle around, but there are no empty spots. You decide to wait at one end of the lot, where you can see (and command) about 20 spaces.

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How long do you have to wait before someone frees up a space?

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How long do you have to wait before someone frees up a space?

Requires:

Assumptions
Developing a model
Applying that model

On a Saturday afternoon, you pull into a parking lot with unmetered spaces near a shopping area. You circle around, but there are no empty spots. You decide to wait at one end of the lot, where you can see (and command) about 20 spaces. On average people shop for 2 hours.

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Requires:

Applying a (new) model

On a Saturday afternoon, you pull into a parking lot with unmetered spaces near a shopping area, where people are known to shop, on average, for 2 hours. You circle around, but there are no empty spots. You decide to wait at one end of the lot, where you can see (and command) about 20 spaces.

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$$t_{wait} = \frac{T_{shop}}{N_{spaces}}$$

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How long do you have to wait before someone frees up a space?

Requires:

Using a calculator

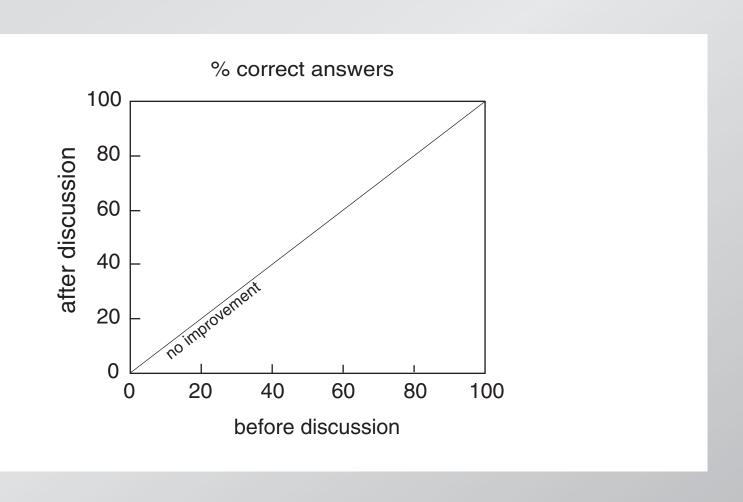
$$t_{wait} = \frac{T_{shop}}{N_{spaces}}$$

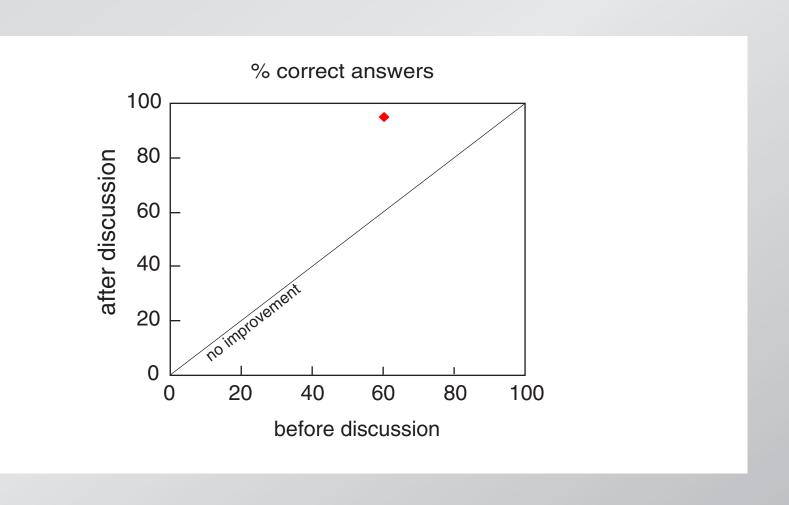
A good reading assignment question...

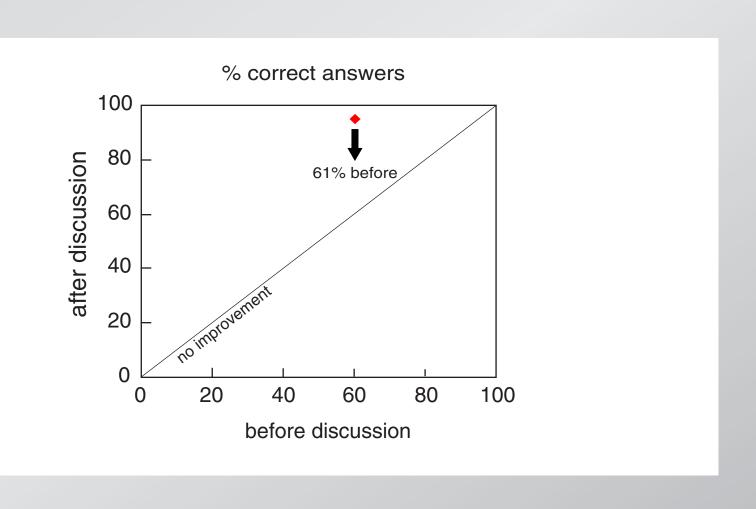
- relates to the reading assignment
- cannot be copied from the text
- tests a concept in the book in a new context
- lets students demonstrate familiarity with text

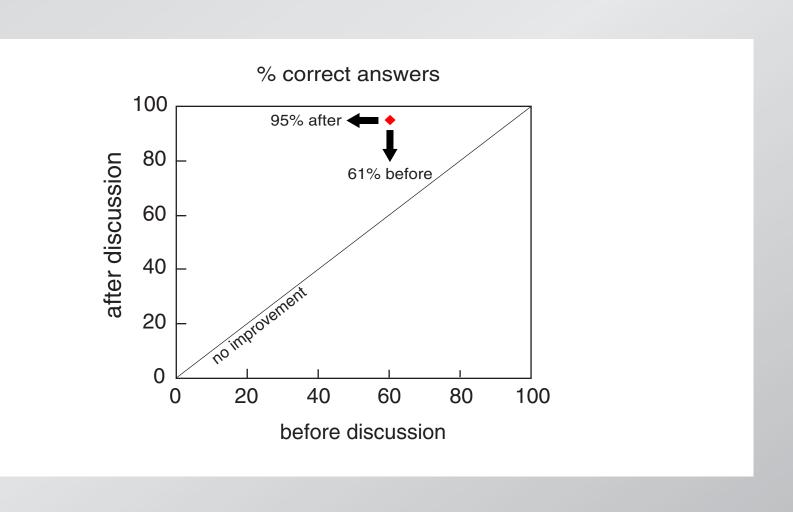
A good ConcepTest...

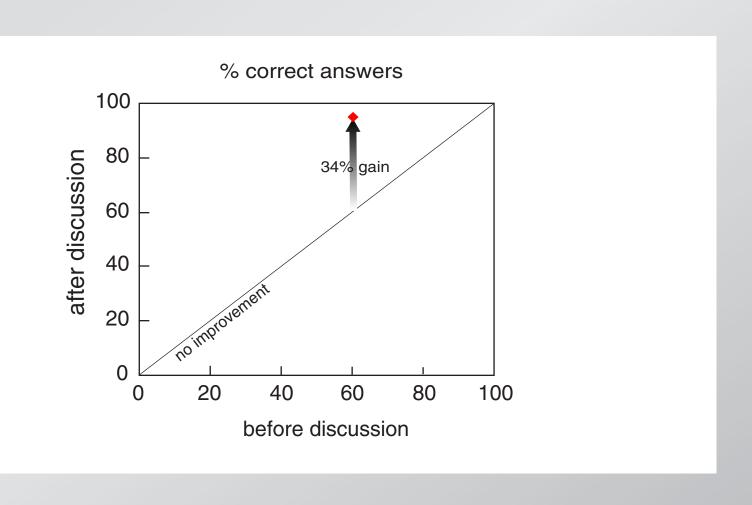
- helps develop mental models (or stimulate discussion)
- tests undestanding, not memorization
- is just challenging enough (30–70% rule)
- has appropriate distractors

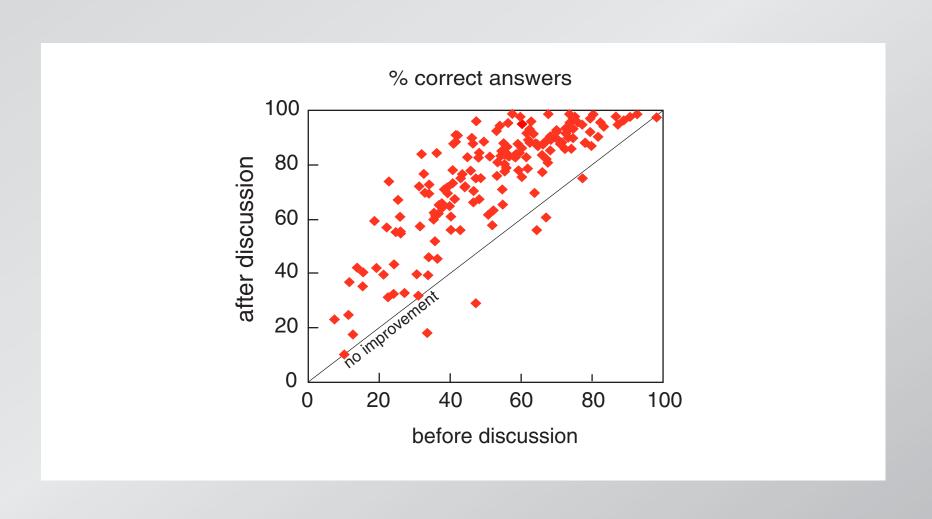


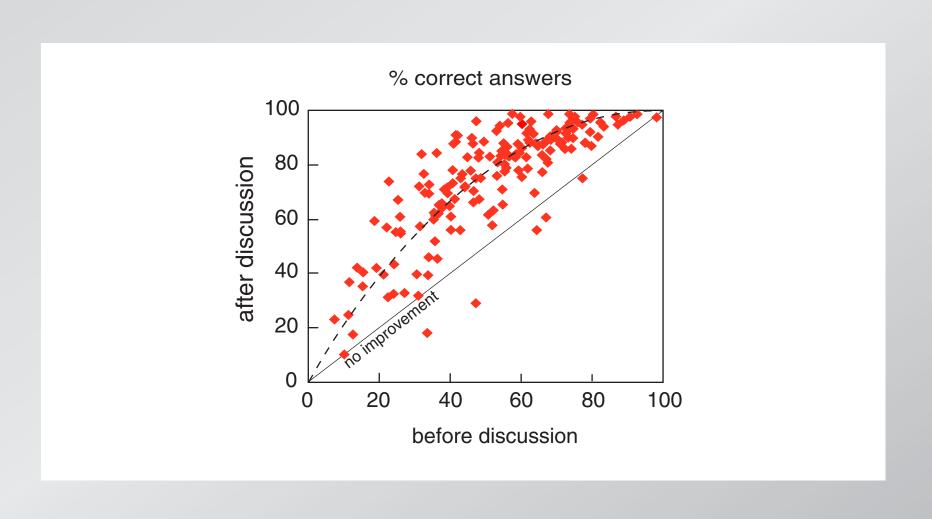


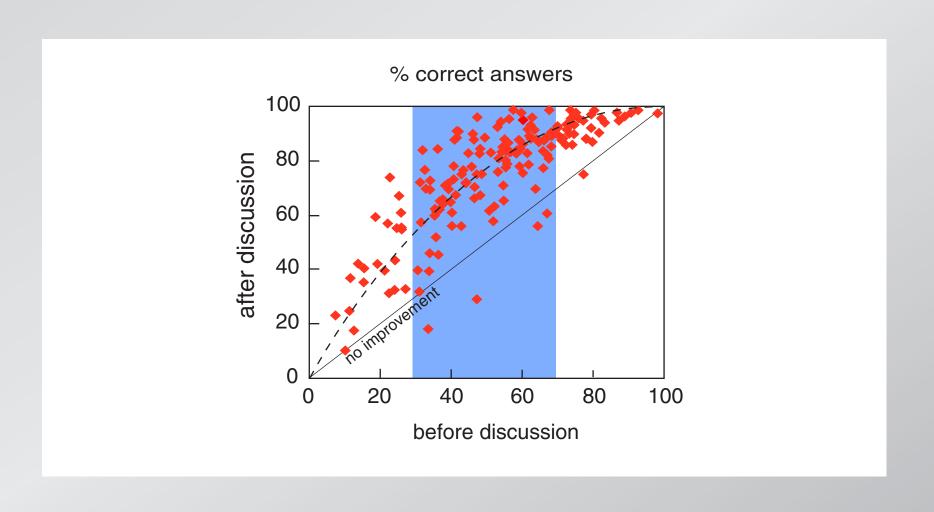


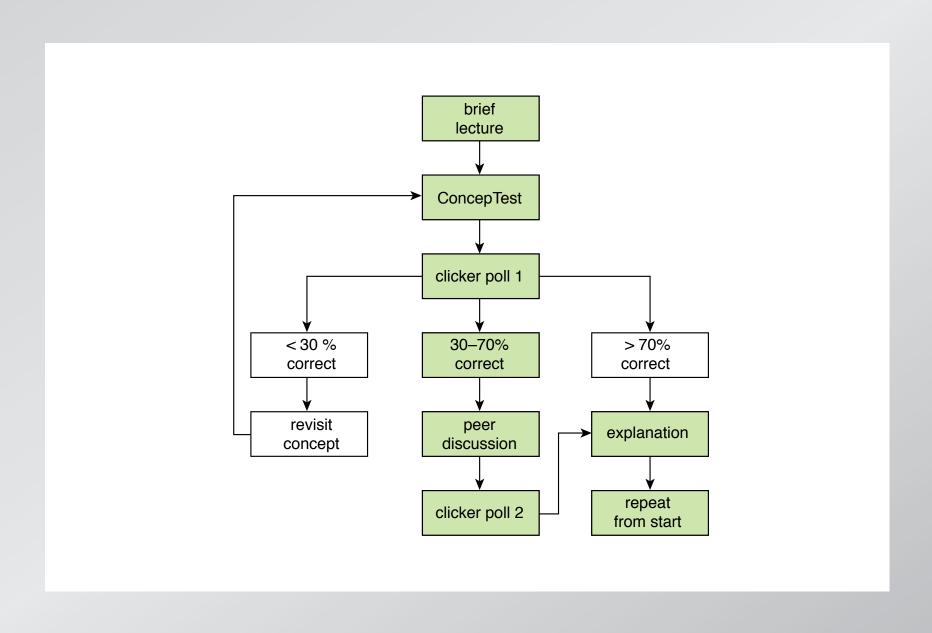












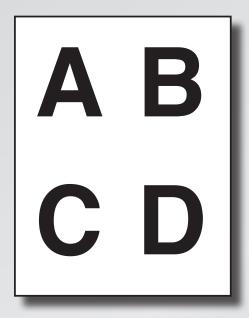
A good ConcepTest...

- needs not be multiple choice
- needs not have a correct answer

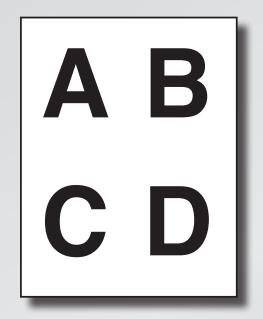


"Are clickers a required resource?"

Flashcards: simple and effective



Flashcards: simple and effective





Meltzer and Mannivanan, South Eastern Louisiana University

circumference of a circle:

 $2\pi R$

Imagine a rope that fits snugly along the equator.



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radius of circle with new circumference:

$$2\pi R = 2\pi R_{\rm E} + 1 \, \text{m}$$
, and so $R = R_{\rm E} + \frac{1 \, \text{m}}{2\pi}$.

You all got fired up!

You all got fired up!

(WITHOUT CLICKERS!)

It's not the technology, but the pedagogy!

"How do I respond when students ask me to lecture?"

Written on Wednesday Feb 16, two weeks into the course:

Subject: concerns

Professor Mazur,

Here are a few concerns. I speak for many of my classmates.

- 1) You are giving us WAY to much work. After spending multiple hours on the problem set, and not being able to figure out many of the questions, I now see that we have an additional 6 or 7 pages or homework in the workbook. I just spent 4 hours on the lab, and I am not confident on almost half of the questions. This is more work than I have had all semester in all of my other classes combined.
 - 2) If you are going to give us this much work, I would suggest re-structuring the lectures. I find the readings very difficult to understand. I am not a bad student (I got a solid A in physics 1a), but it is very difficult to internalize the readings. You should spend most of the lecture going over, point by point, the readings in their entirety. While the PRS clickers are fun, they do not help me understand the complex material.

I am extremely flustered by the incredibly large amount of work, and my inability to understand it, and I am strongly considering dropping the course.

Written on Monday May 23, just after the final exam:

Subject: Thanks!

Professor Mazur,

First of all I want to thank you for a great semester. You are an excellent professor, and it is clear that you truly care about each and

The exam went well today. I'm not sure to what extent you will curve the final grades (if at all), but it looks like I may be right around the cutoff point between an A and an A-. I studied as hard as I could and I'm keeping my fingers crossed about the A, but no matter what happens with my grade you should know that you are one of the best professors that I have ever had at Harvard.

Thanks again!

"Are there specific challenges when applying this technique to the social sciences?"

"[How to formulate questions in courses] where the learning trains analysis and argument, but where single solutions are rare?

Professors A and B teach the same mechanics class at the same college during different semesters. Professor A uses the traditional approach to teaching and lectures. Professor B uses Peer Instruction and students respond to the questions using clickers. Each class is evaluated using the traditional end-of-semester questionnaire and using the FCI to measure students' comprehension of mechanics. Both professors are middle-aged and male. The results are as follows.

A: student evaluation: 1.5/5.0; $\langle g \rangle = 0.42$

B: student evaluation: 3.7/5.0; $\langle g \rangle = 0.57$

Professor	Α	В
pedagogy	traditional	PI with clickers
student evaluation	1.5/5.0	3.7/5.0
FCI <g></g>	0.42	0.57

What might account for the large difference in evaluation?

- I. professor personality
- II. technology
- III. pedagogy
- 1. I only
- 2. II only
- 3. III only

- 4. II and III
- 5. I, II, and III
- 6. other combination



a couple of points worth noting:

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1. you got engaged

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"What about coverage?"

	"lectures"	PI	
coverage	complete	partial	

	"lectures"	PI
coverage	complete	partial
material learned	little	substantial

	"lectures"	PI
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what does coverage mean if little is retained?

Keep those good question coming!

Your next assignment:

work on ConcepTest development

http://bit.ly/KTHday2



Research Funding:

Pew Charitable Trust, Pearson/Prentice Hall, Davis Foundation, Engineering Information Foundation, Derek Bok Center for Teaching and Learning, National Science Foundation

for more information and a copy of this presentation:

http://mazur.harvard.edu

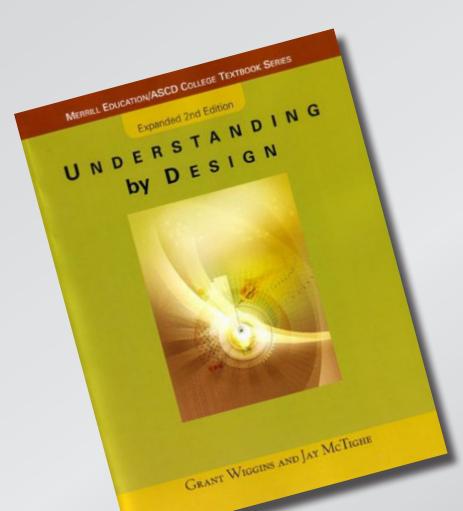




Looking ahead

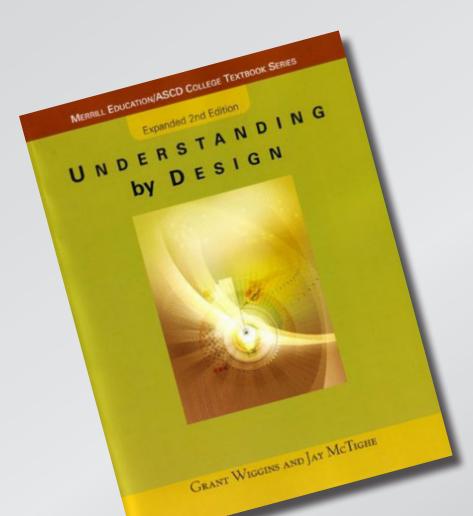
Improve course by setting learning goals

Setting learning goals



Grant Wiggins and Jay McTighe, *Understanding by Design* (Prentice Hall, 2001)

Setting learning goals



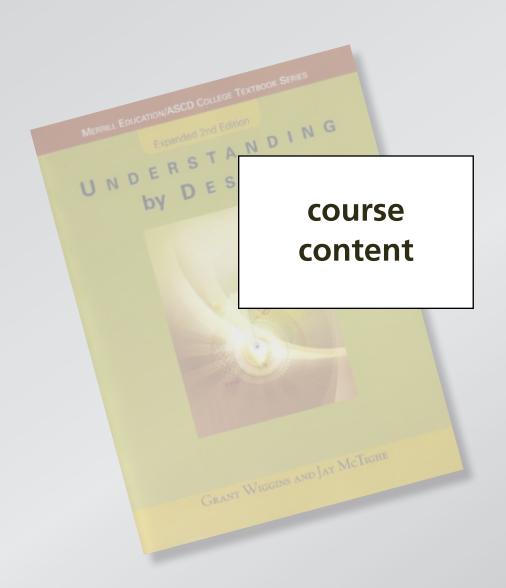
approach, not content

focus on understanding

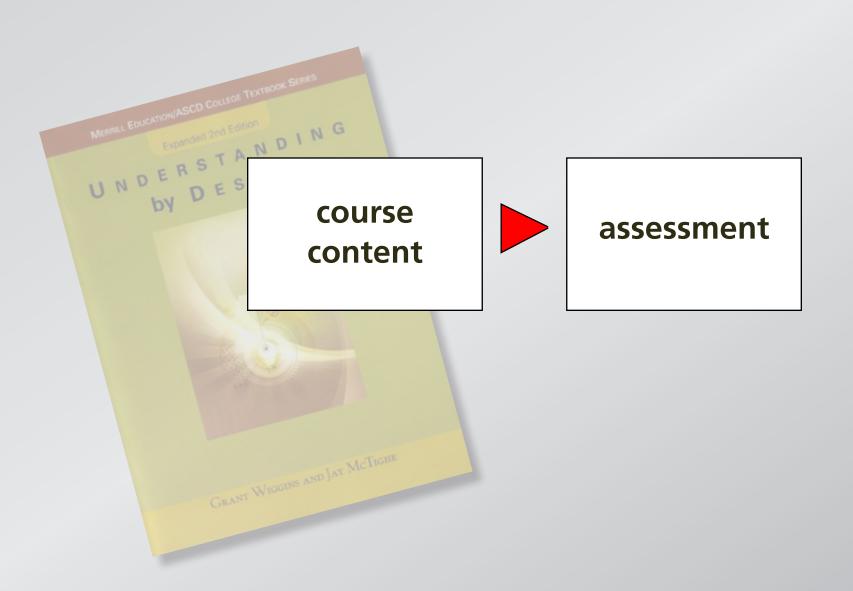
backward design

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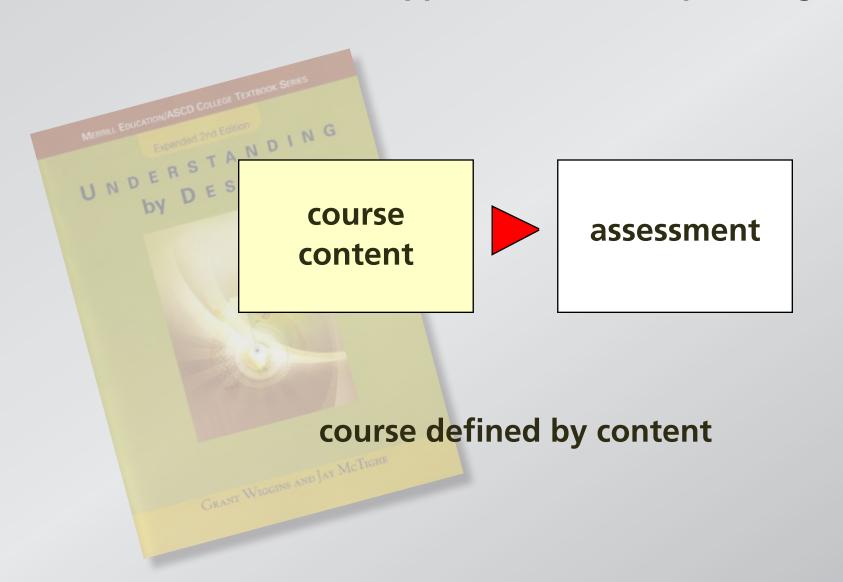
Traditional approach to course planning



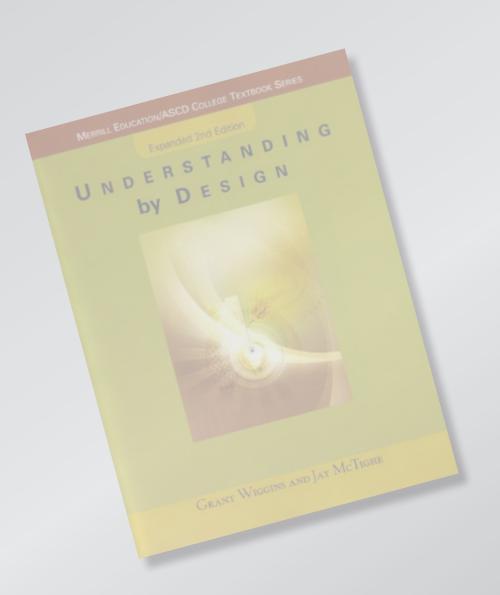
Traditional approach to course planning



Traditional approach to course planning

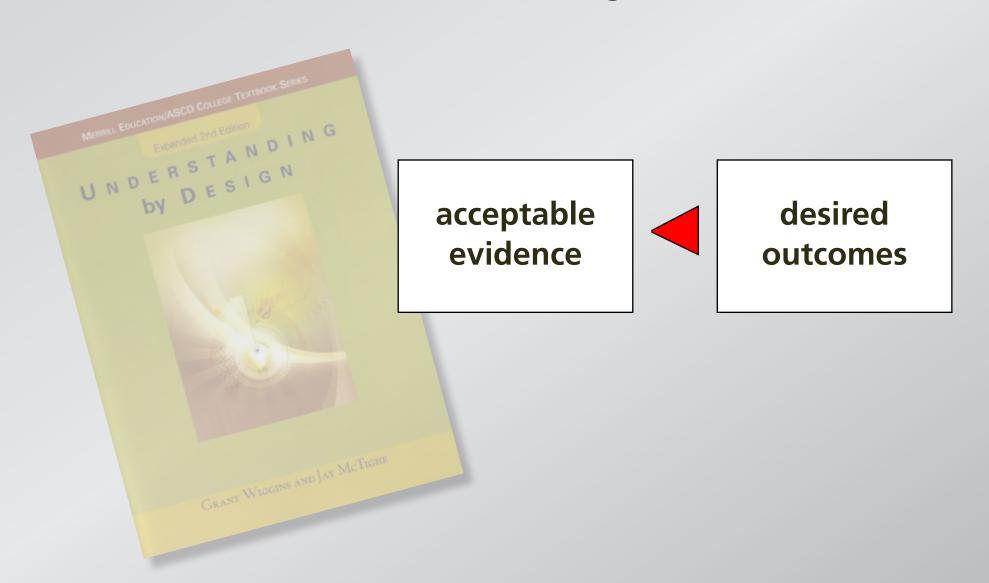


Backward design

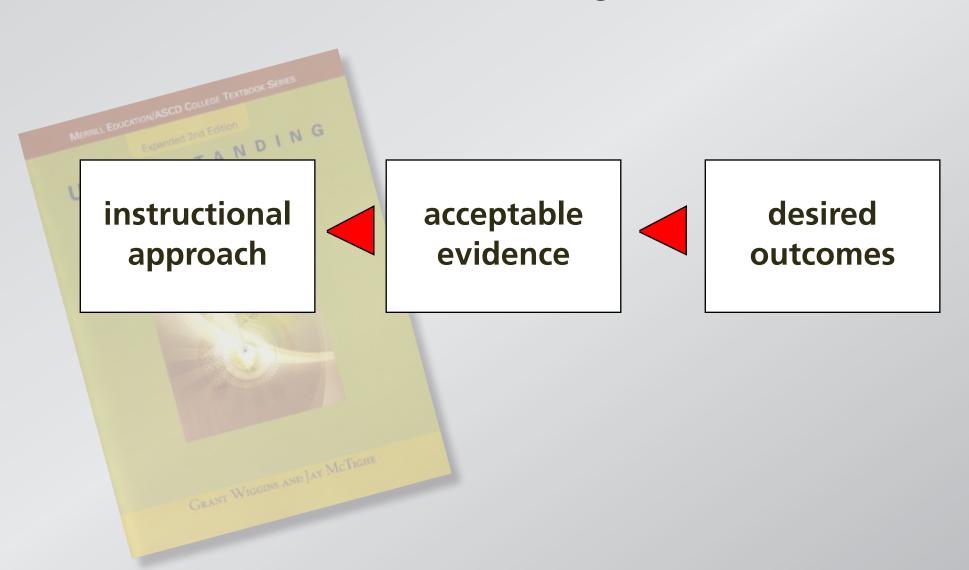


desired outcomes

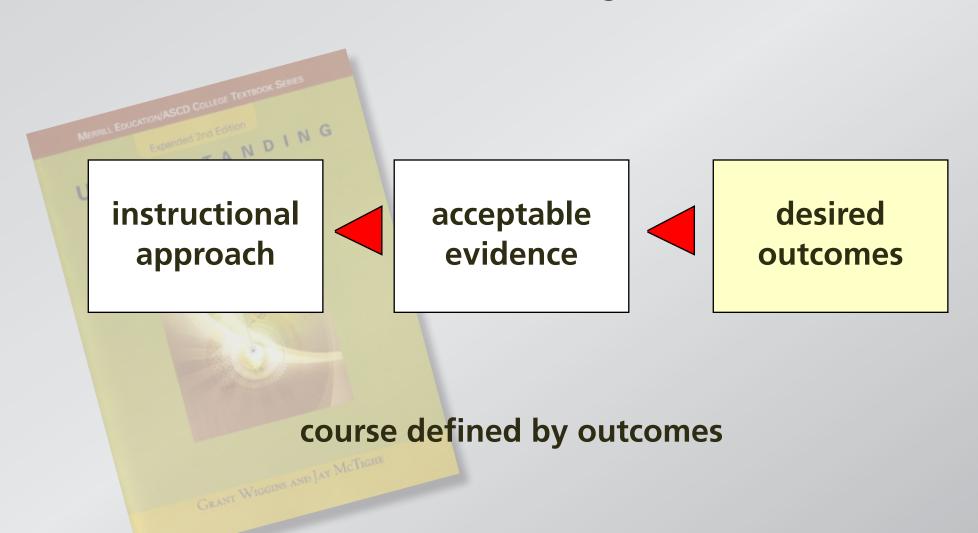
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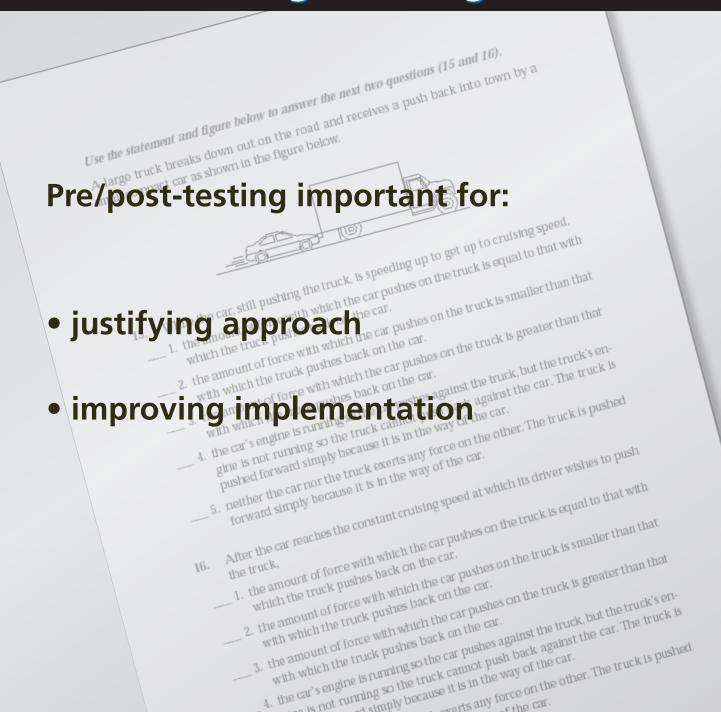
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To set stage for successful implementation, I need to...

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(actions to take before course begins)

- convince yourself (and your colleagues)
- set learning goals
- select approaches
- identify resources



Evaluate assessment by comparing student performance on various kinds of problems

How to move information transfer out of classroom?

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(WITHOUT CLICKERS!)

It's not the technology, but the pedagogy!

(but clickers do offer advantages)

To make it happen, I need to...

To make it happen, I need to...

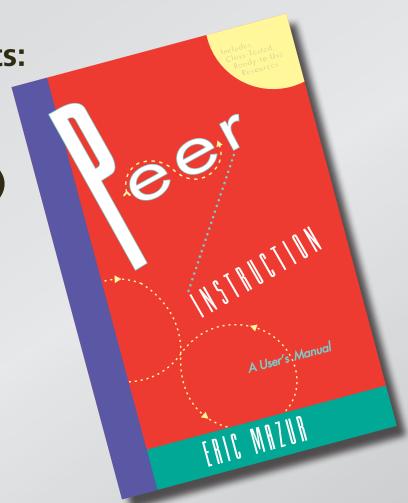
(actions to take during course)

find or develop good questions

know how to manage time

Books with ConcepTests:

Physics (Prentice Hall)



Books with ConcepTests:

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Chemistry (Prentice Hall)

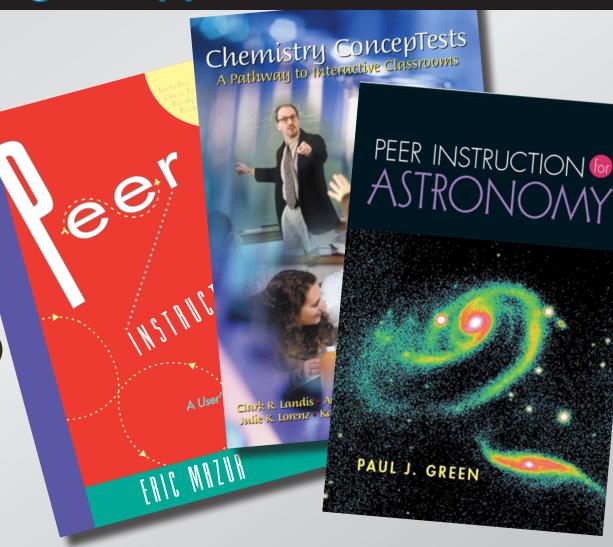


Books with ConcepTests:

Physics (Prentice Hall)

Chemistry (Prentice Hall)

Astronomy (Prentice Hall)



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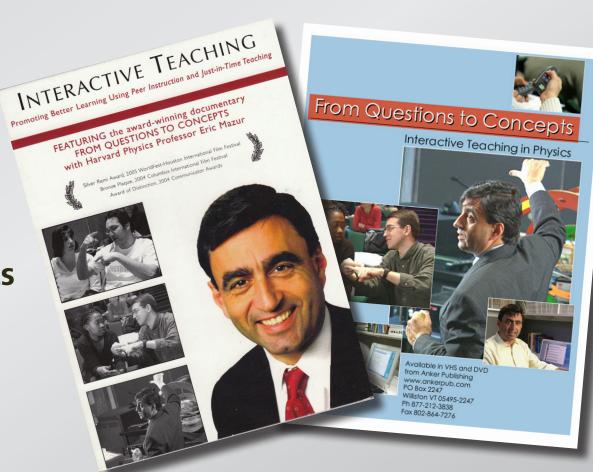
Astronomy (Prentice Hall)

Calculus (Wiley)



Videos:

- Interactive Teaching DVD
- From questions to concepts



Types of questions

- survey
- discussion
- model testing
- select from list

It's not the technology, but the pedagogy!

(but clickers do offer advantages)

Which of the following airlines tries to save fuel by suggesting that its passengers use the bathroom before boarding?

- 1. Delta Airlines
- 2. Lufthansa
- 3. All Nippon Airways
- 4. British Midland Airways
- 5. Air France
- 6. JAL
- 7. Aboriginal Air Services
- 8. Aeroflot
- 9. Are you kidding me? None of the above.

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hole in plate/circumference model

Professor A/B discussion

airline fact

hole in plate/circumference model

Professor A/B discussion

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fact-recall not engaging

Good conceptual questions (ConcepTests):

- are based on common student difficulties
- focus on single concept
- require more than "plug and chug"
- are clear and concise
- are of manageable difficulty

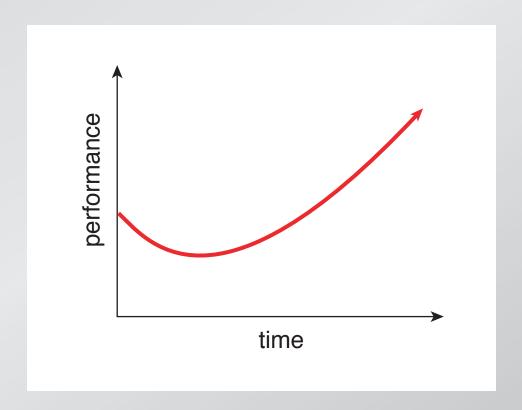
must adjust level to audience

Outline Setting the stage Making it happen Overcoming barriers

What are some potential barriers?

- skepticism
- growing pains
- negative feedback
- limited circle of influence

After changing, things might get worse before they get better!



Better understanding leads to more — not fewer — questions!

(must recognize confusion as step towards understanding)

Things to do:

- take/analyze data
- motivate students
- be prepared for initial adjustments

Why is change so hard?

	"lectures"	PI	considered
coverage	complete	partial	requirement
preclass reading	none	cover everything	
confusion	little none	substantial	
evaluations	known	unknown	

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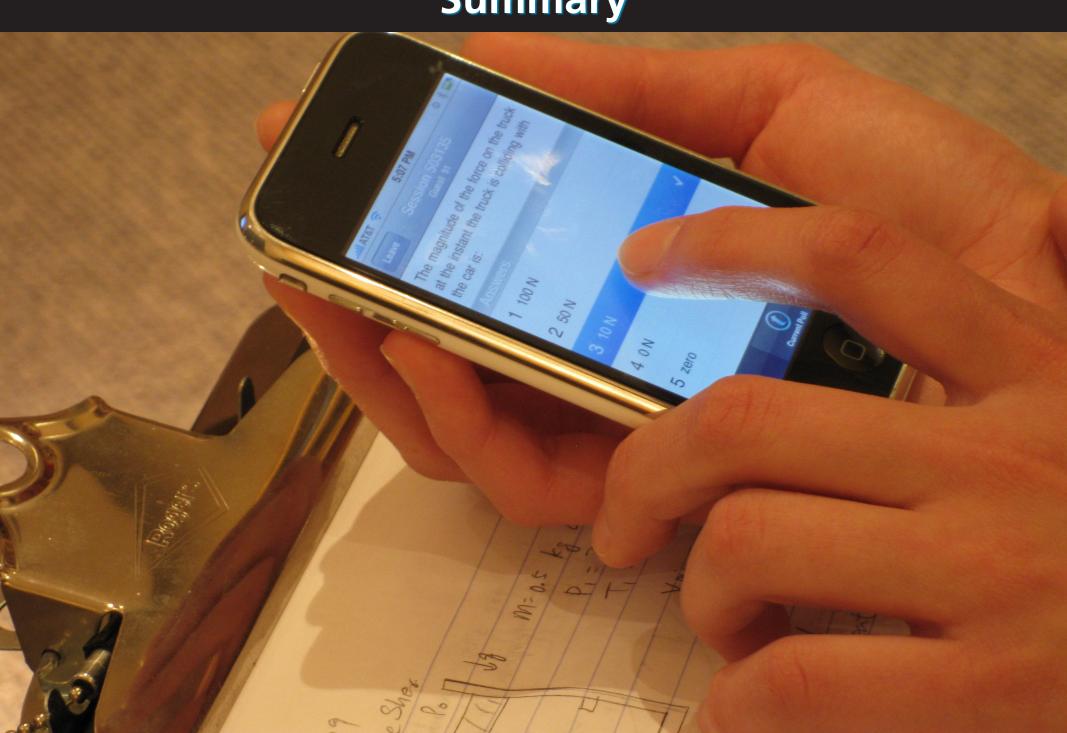
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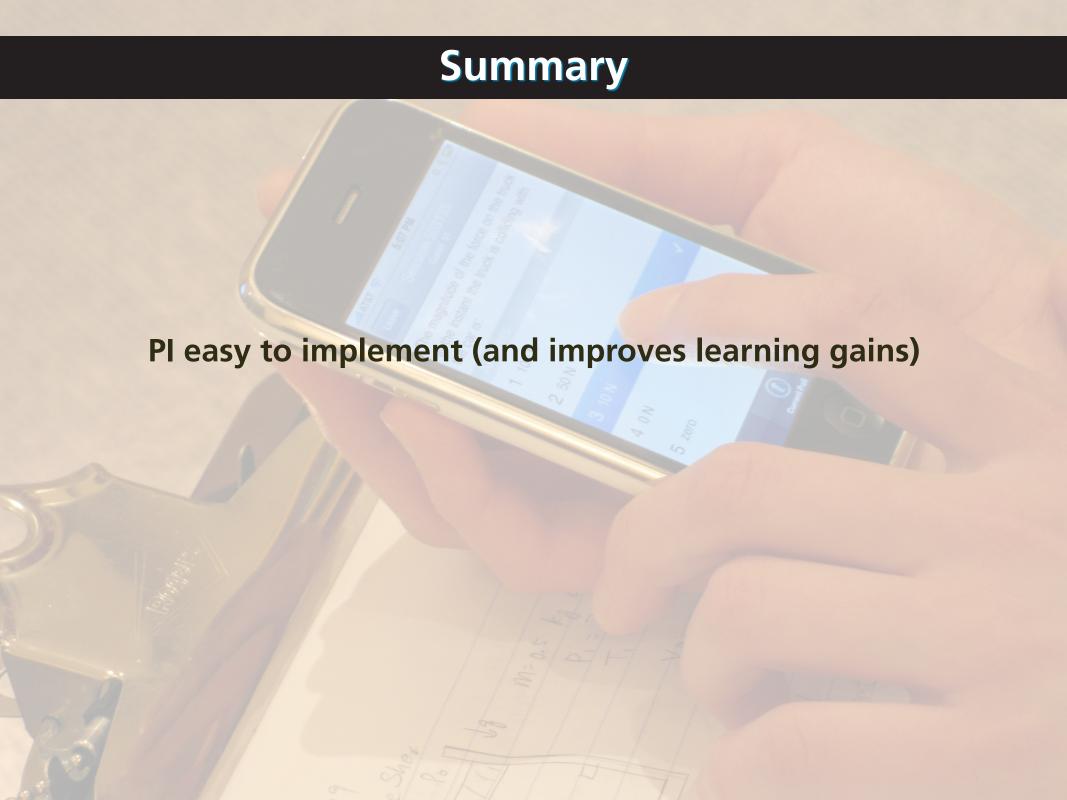
But PI leads to better learning and retention

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(neither of which is traditionally measured)

Summary





Conclusion

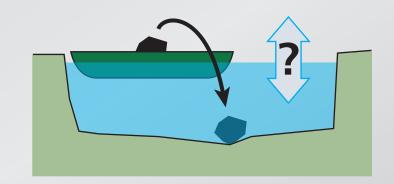
PI easy to implement (and improves learning gains)

technology facilitates active engagement (but not required)

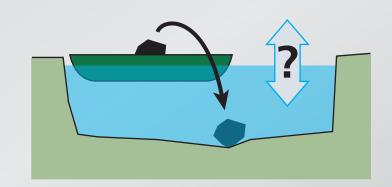
Conclusion

not just a polling tool, but an engagement tool!

A boat carrying a large boulder is floating on a small pond. The boulder is thrown overboard and sinks to the bottom of the pond.



A boat carrying a large boulder is floating on a small pond. The boulder is thrown overboard and sinks to the bottom of the pond.



After the boulder sinks to the bottom of the pond, the level of the water in the pond is

- 1. higher than
- 2. the same as
- 3. lower than

it was when the boulder was in the boat.

We all make mistakes!

When we hold a page of printed text in front of a mirror, the text on the image in the mirror runs from right to left:

The New York Times

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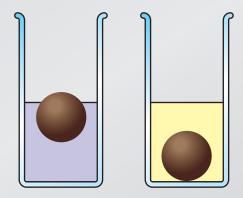
The New York Times

Why is it that right and left are interchanged and not top and bottom? Because:

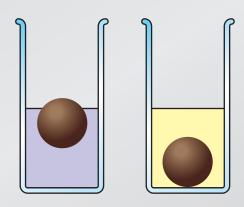
- 1. the mirror is oriented vertically.
- 2. we have two eyes in the horizontal plane.
- 3. the Earth's gravitation is directed downward.
- 4. a habit we have when looking at images in a mirror.
- 5. It only appears to run from left to right.

It's "simple" only if you know the answer

Consider an object that floats in water, but sinks in oil. When the object floats in water, most of it is submerged.

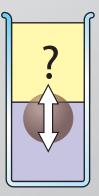


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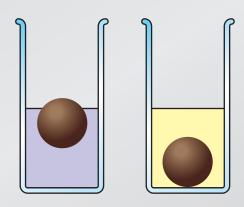


If we slowly pour the oil on top of the water so it completely covers the object, the object

- 1. moves up.
- 2. stays in the same place.
- 3. moves down.

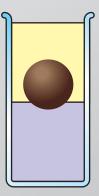


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It's easy to make simple demonstrations fascinating!

The specific heat at constant volume for a monatomic crystal approaches zero at low temperature even though the specific heat for a monatomic gas remains $\frac{3}{2}k$ per atom. Why is this so?

The specific heat at constant volume for a monatomic crystal approaches zero at low temperature even though the specific heat for a monatomic gas remains $\frac{3}{2}k$ per atom. Why is this so?

- 1. Potential energy doesn't play a role for the monatomic gas, but it does for the crystal.
- 2. The particles are indistinguishable in the gas, but not in the crystal.
- 3. The energy difference between allowed states for the crystal is much larger than it is for the atoms.

Research Funding:

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