Peer Instruction Workshop

0



Ready-to-Use Resources C.C. STRUCTION NOT AND NOT Nile : Egypt :: Ganges @Response(Na), Turning Technologies ResponseCard[®] RI 5/E A User's: Mai 310 21B AIA 6IF AID SIE Universidad del Norte Barranquilla, Colombia, 9 de junio, 2011 814 71G 017 ingtechnologies.com Ch

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unique ID on back of clicker





Peer Instruction...

- 1. Never heard of it.
- 2. Heard of it, but don't really know what it is.
- 3. Quite familiar with it.
- 4. I heard you speak about it so often, I could give your talk



Peer Instruction...

- 1. Never heard of it.
- 2. Don't use it in my classes, but I'm open to it.
- **3. Considering using it in my classes.**
- 4. I have used it it in my classes a few times only.
- 5. I use it regularly in my classes.

lectures focus on information transfer...

lectures focus on information transfer...

but education is much more!

1. information transfer

1. information transfer

2. assimilation of information

1. information transfer (easy)

2. assimilation of information (hard and left to student)

Solution: move information transfer out of classroom!

Includes Class-Tested, Ready-to-Use Resources

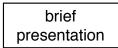
NSTRUCTION NOT AND NOT

EBIC MALUA

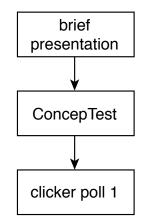
A User's Manual

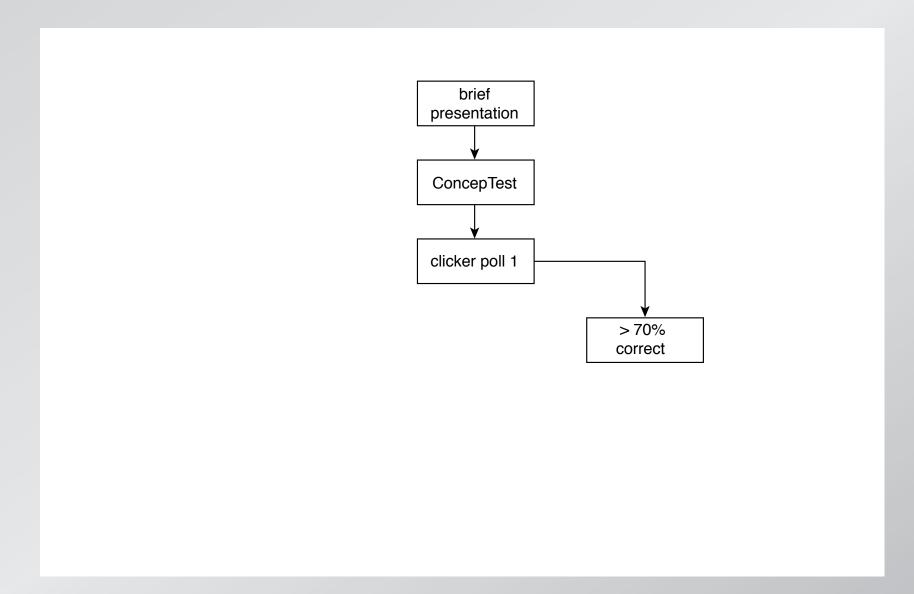
Main features:

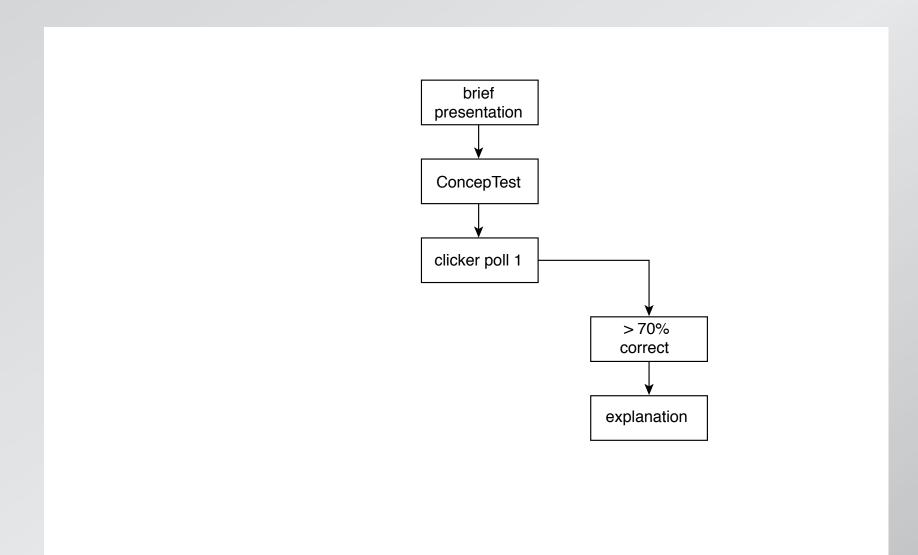
- pre-class reading
- in-class: depth, not 'coverage'
- ConcepTests

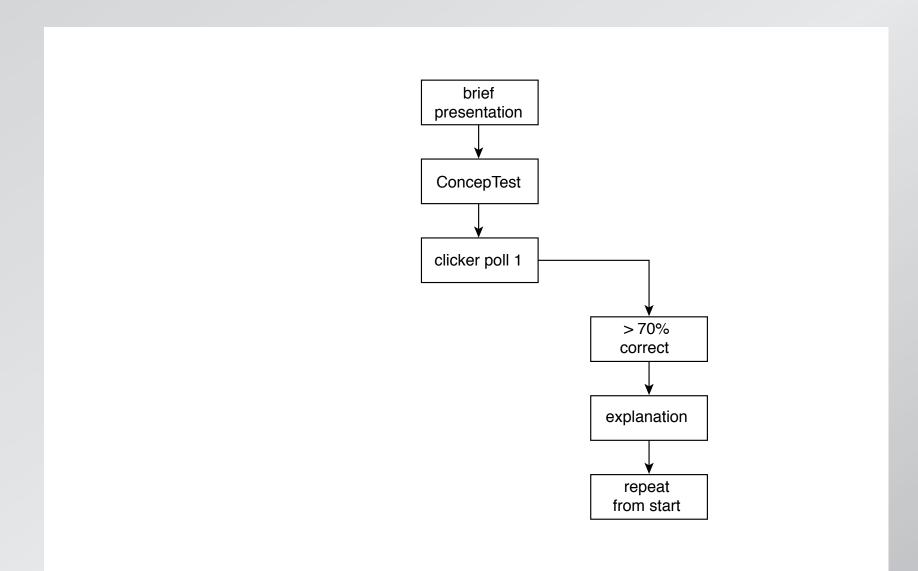


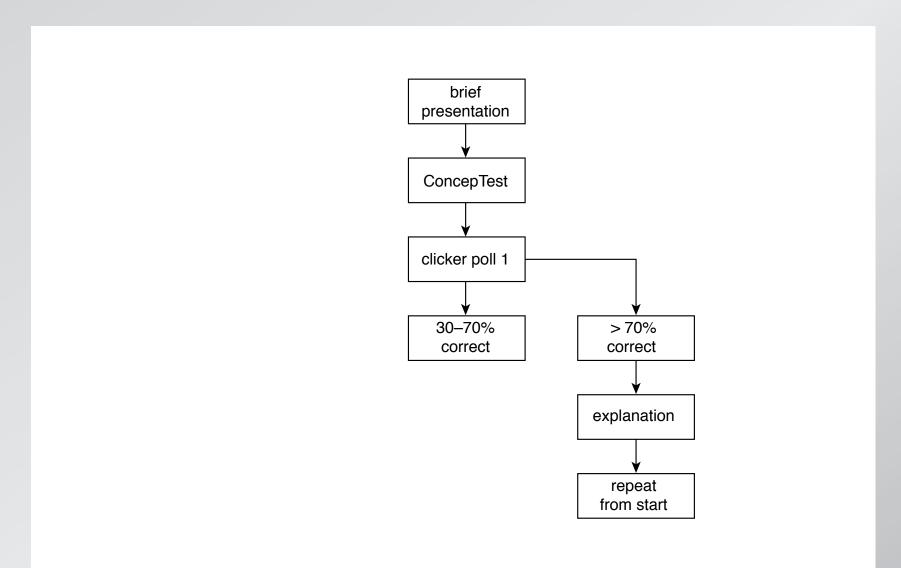


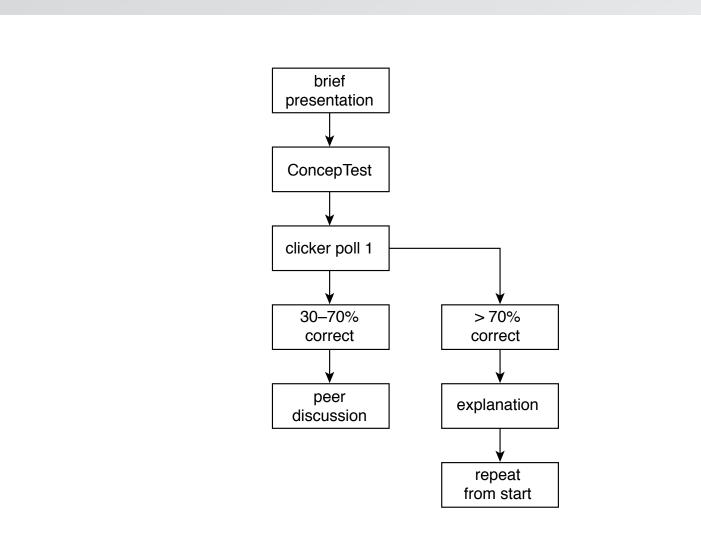


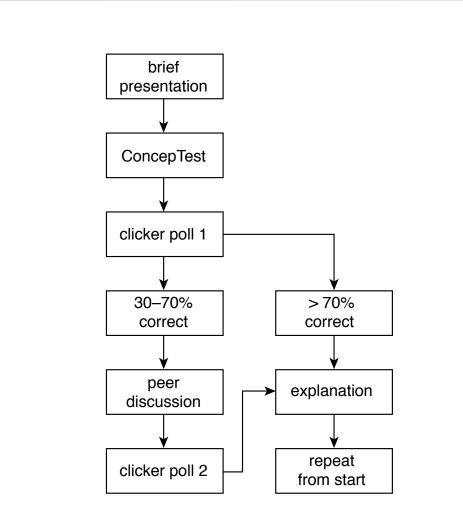


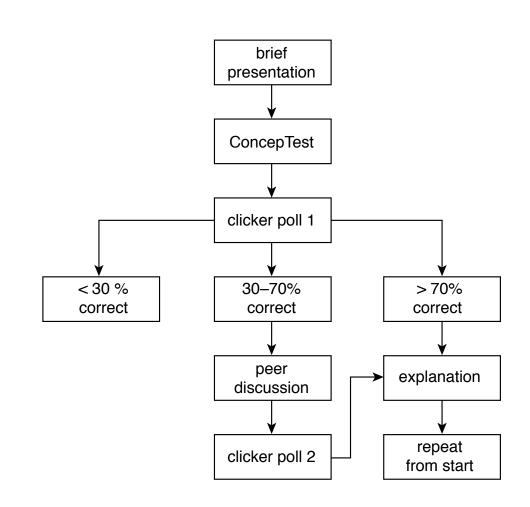


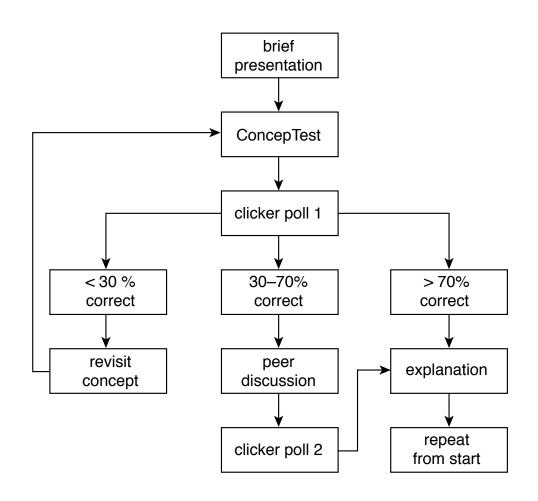




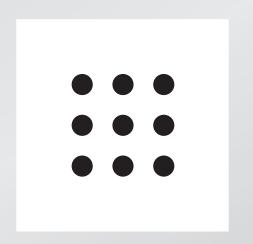




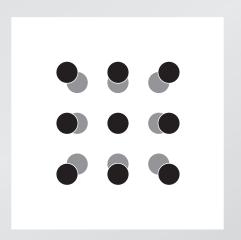




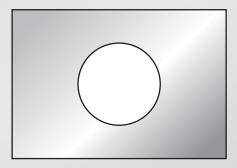
When metals heat up, they expand because all atoms get farther away from each other.



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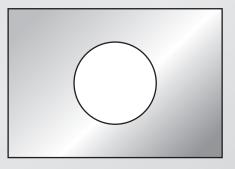
Consider a rectangular metal plate with a circular hole in it.



Consider a rectangular metal plate with a circular hole in it.

When the plate is uniformly heated, the diameter of the hole

- 1. increases.
- 2. stays the same.
- 3. decreases.





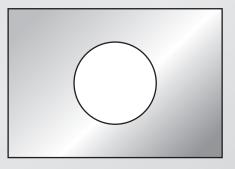


It's easy to fire up the audience!

Consider a rectangular metal plate with a circular hole in it.

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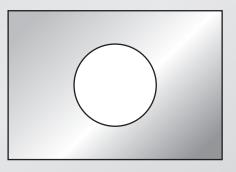
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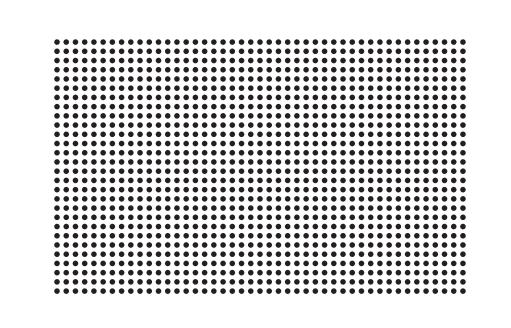
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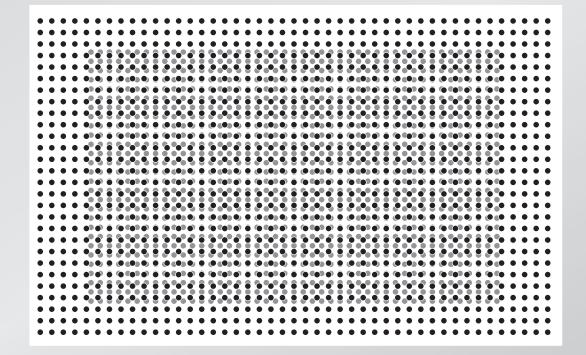
- 1. increases. 🖌
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- 3. decreases.



remember: all atoms must get farther away from each other!

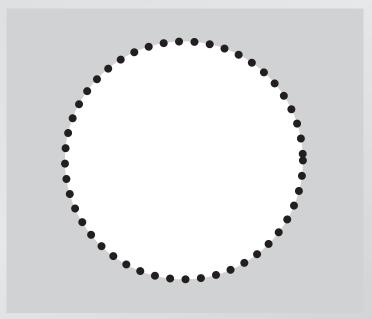


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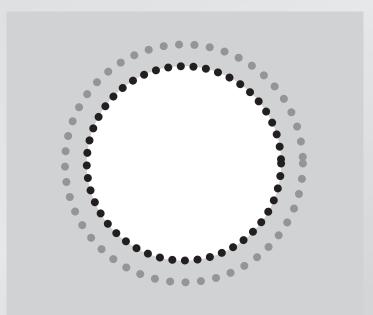


consider the atoms at the rim of the hole



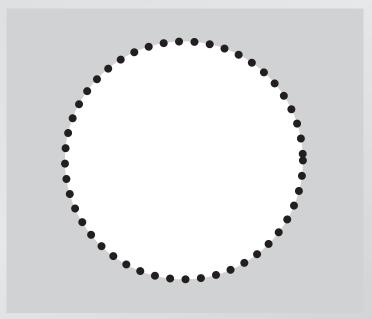


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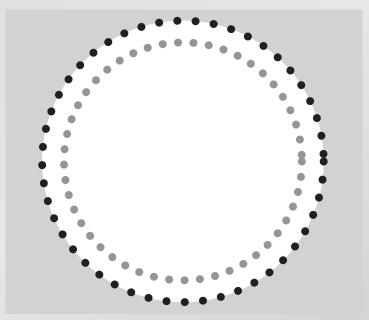


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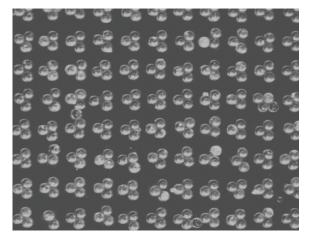


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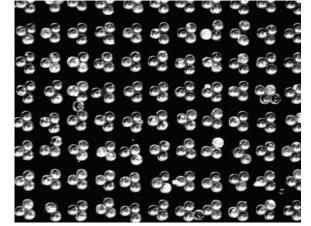


Let's try it!

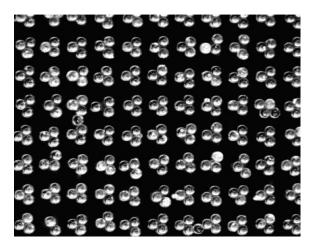
original



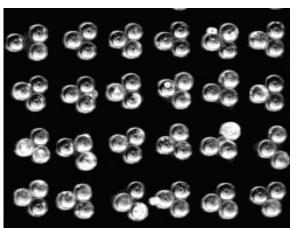
1. adjust contrast



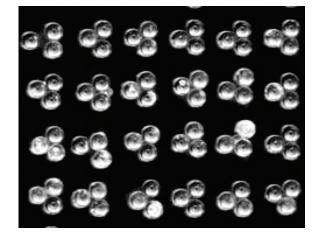
2. remove blemishes

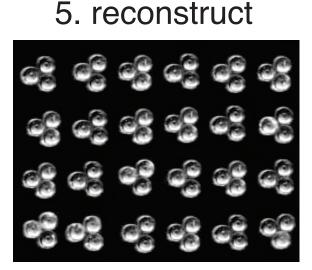


3. crop



4. remove outliers





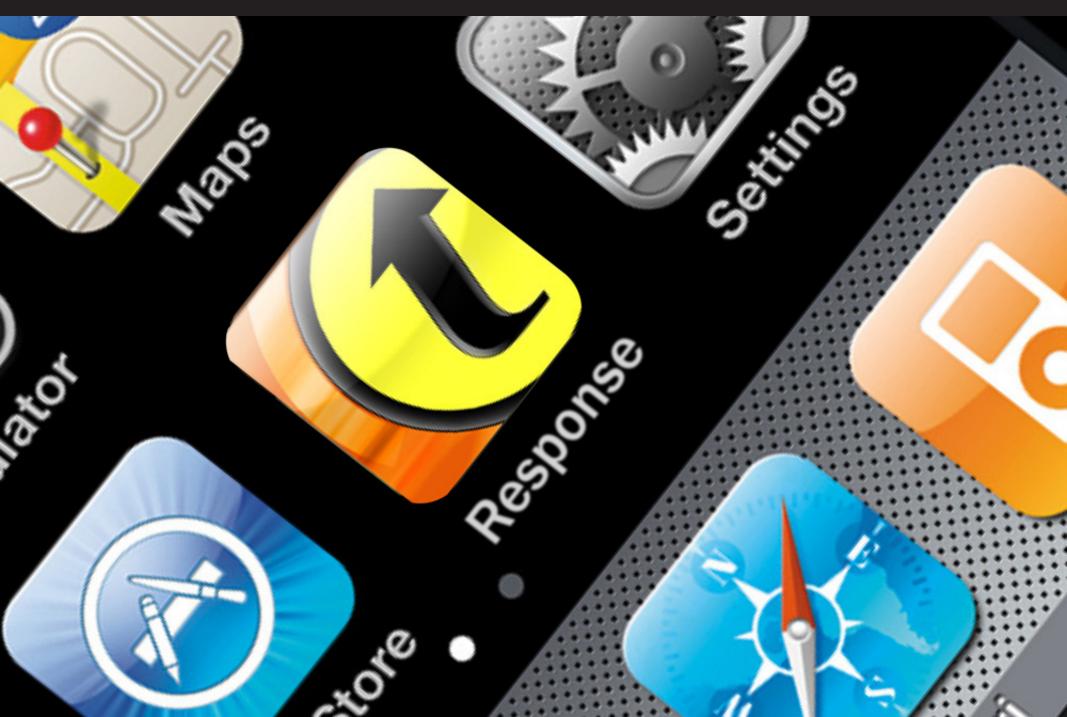


At which step were acceptable standards of ethics violated?

- **1. Optimize brightness/contrast**
- 2. Remove blemishes
- 3. Crop on optimal area
- 4. Remove outliers
- 5. Reconstruct image with parts copied from other locations

Outline

20



Outline

Setting the stage

• Making it happen

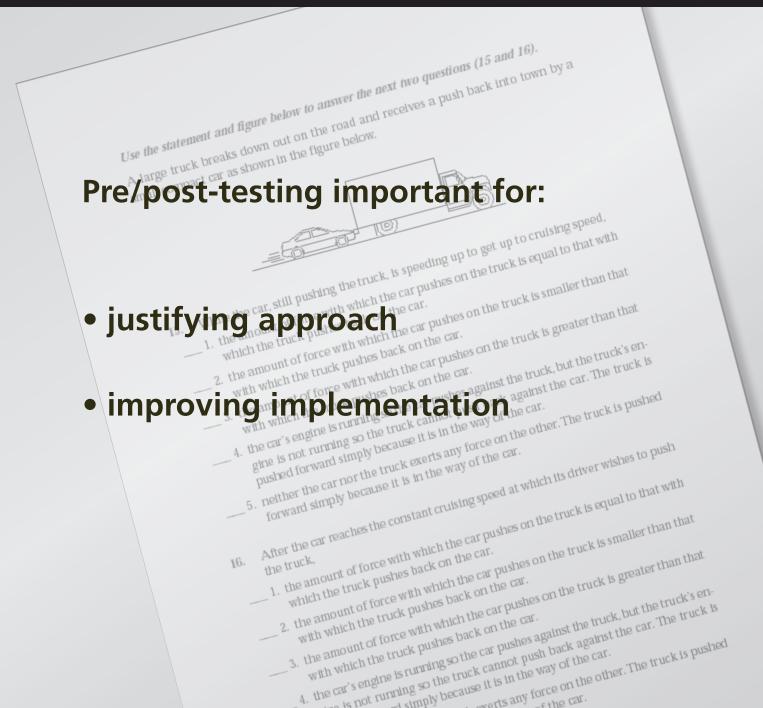
Overcoming barriers

To set stage for successful implementation, I need to...

To set stage for successful implementation, I need to...

(actions to take before course begins)

- convince yourself (and your colleagues)
- set learning goals
- select approaches
- identify resources





Evaluate assessment by comparing

student performance on various kinds of problems

What constitutes a good problem?

On a Saturday afternoon, you pull into a parking lot with unmetered spaces near a shopping area. You circle around, but there are no empty spots. You decide to wait at one end of the lot, where you can see (and command) about 20 spaces.

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Requires:

Assumptions Developing a model Applying that model

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Assuming people leave at regularly-spaced intervals, how long do you have to wait before someone frees up a space?

Requires:

Applying a (new) model

On a Saturday afternoon, you pull into a parking lot with unmetered spaces near a shopping area, where people are known to shop, on average, for 2 hours. You circle around, but there are no empty spots. You decide to wait at one end of the lot, where you can see (and command) about 20 spaces.

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$$t_{wait} = \frac{T_{shop}}{N_{spaces}}$$

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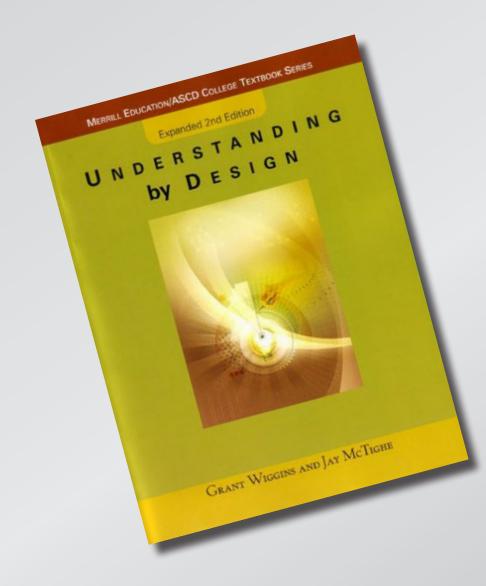
Requires:

Using a calculator

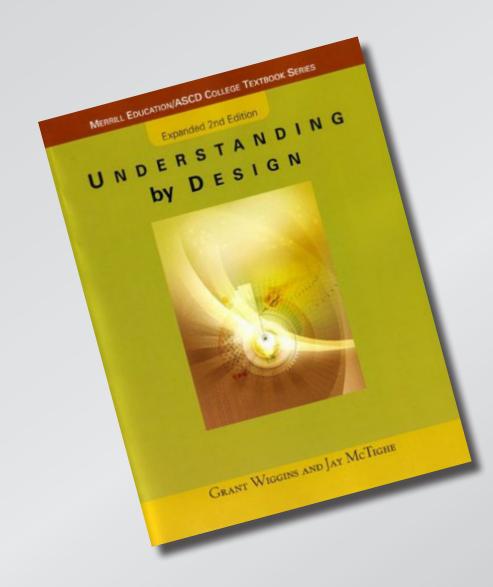
 $t_{wait} = \frac{T_{shop}}{N_{snar}}$

Need to test meaningful skills!

Setting learning goals



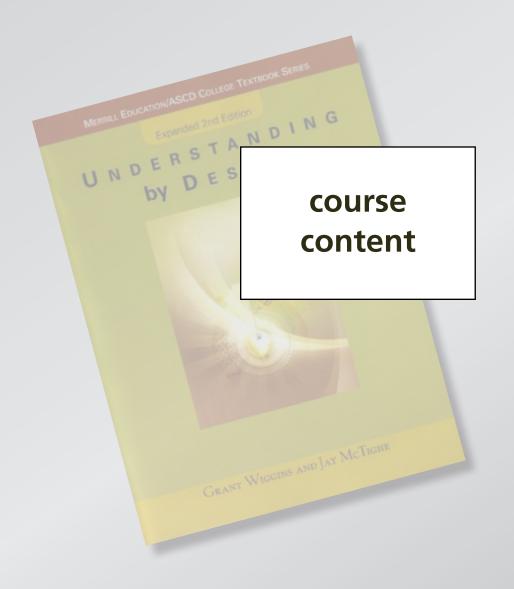
Setting learning goals



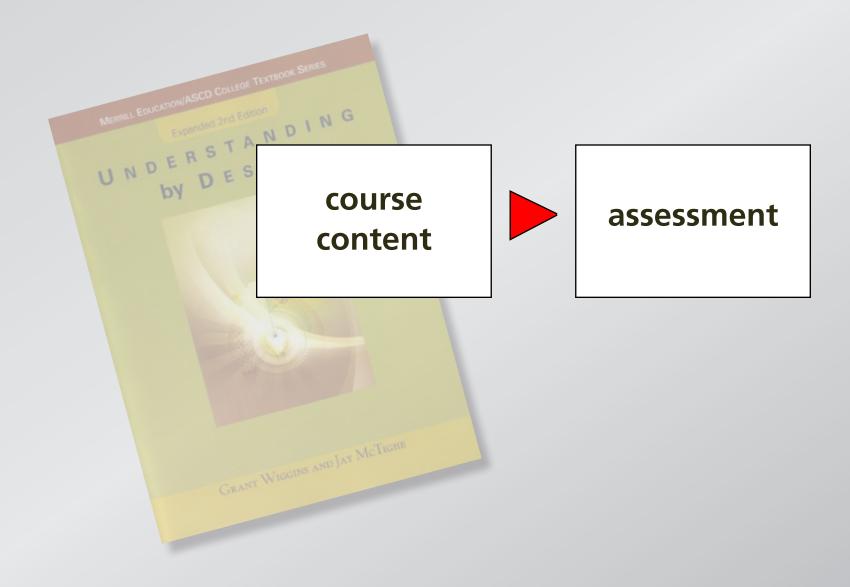
approach, not content

focus on understanding

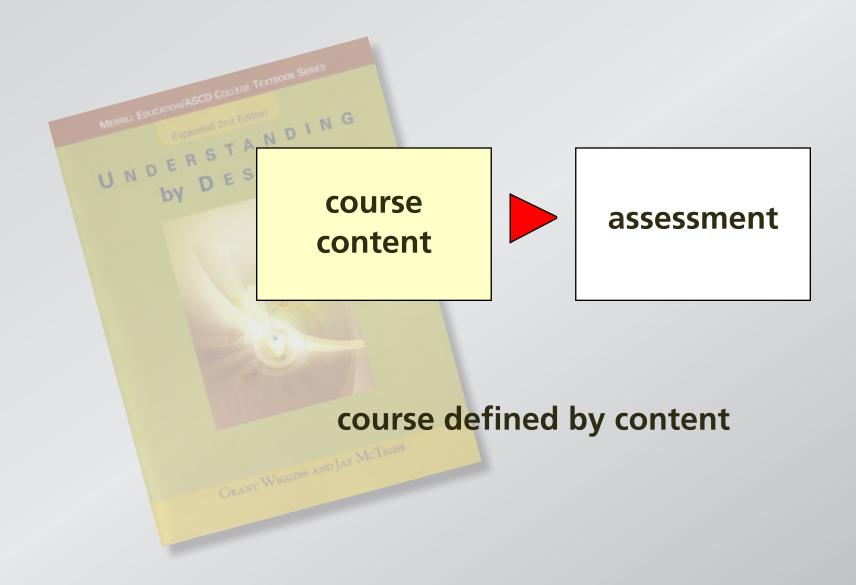
Traditional approach to course planning

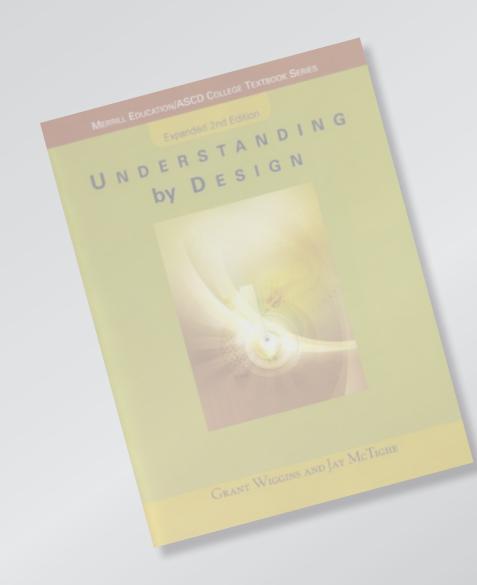


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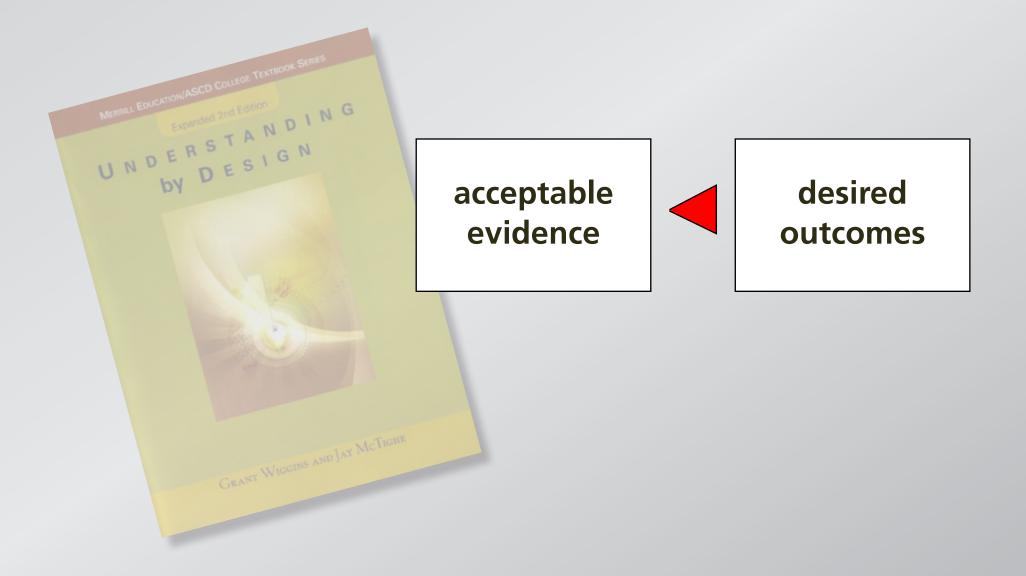


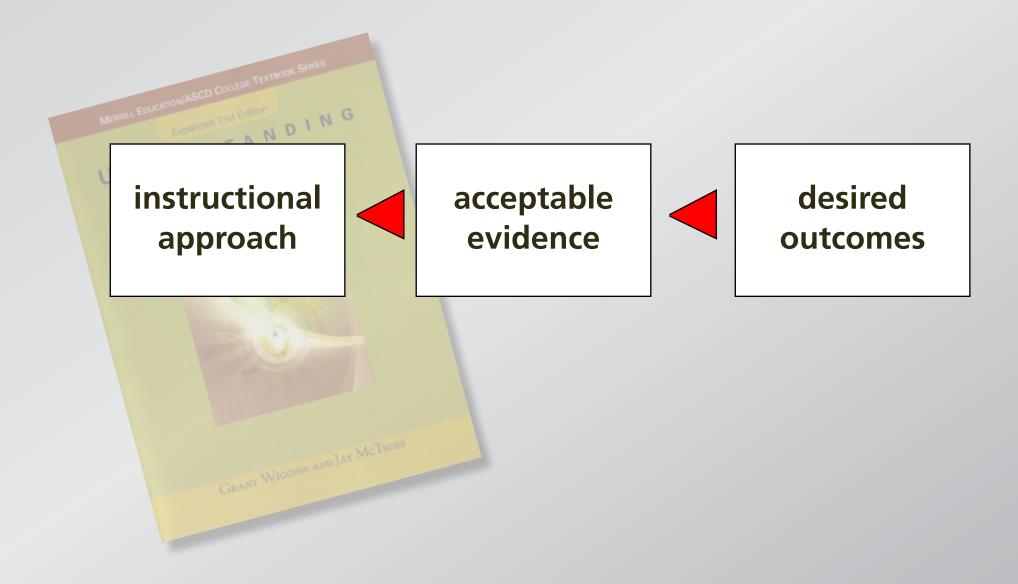
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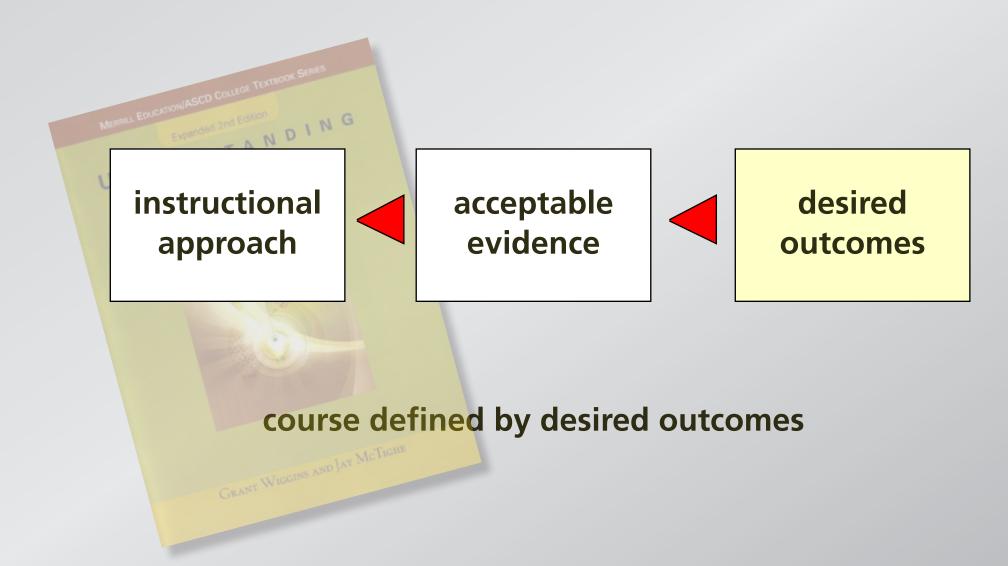








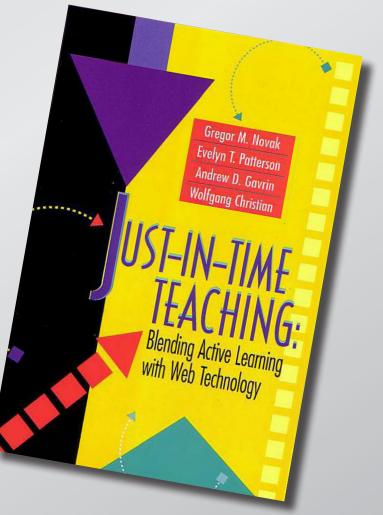




How to move information transfer out of classroom?

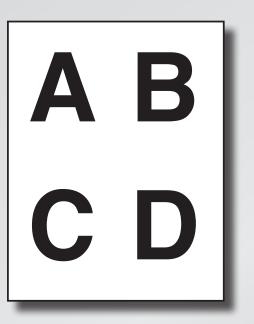
Approaches for reading:

- Reading quizzes
- Summaries
- Just-in-time-Teaching (www.jitt.org)

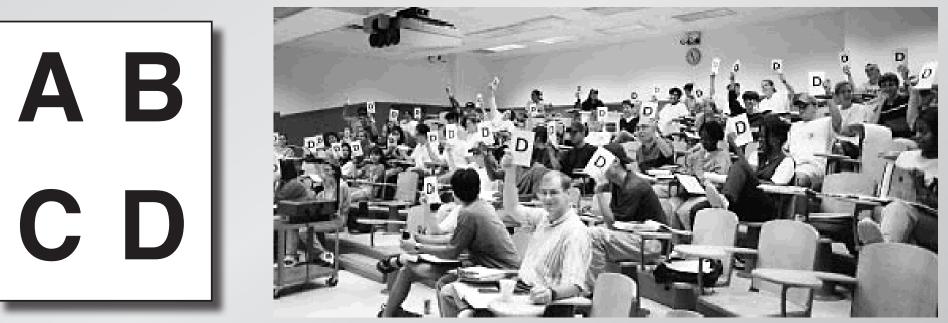


Are clickers a required resource?

Flashcards: simple and effective



Flashcards: simple and effective



Meltzer and Mannivanan, South Eastern Louisiana University

Imagine a rope that fits snugly along the equator.



Let's try it!

Imagine a rope that fits snugly along the equator.

Suppose the rope is cut and 1 m of rope is inserted between the cut ends. If the rope were to maintain a circular shape, how far off the surface of the Earth would it float?

- 1. the width of a few atoms
- 2. the width of a few hairs
- 3. the height of a curb
- 4. exactly 1 m
- 5. more than 1 m





You all got fired up!

You all got fired up!

(WITHOUT CLICKERS!)

It's not the technology, but the pedagogy!

It's not the technology, but the pedagogy!

(but clickers do offer advantages)

Let's try it!

Imagine a rope that fits snugly along the equator.

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the width of a few atoms
the width of a few hairs
the height of a curb
exactly 1 m

5. more than 1 m



circumference at equator:

 $2\pi R_{\rm E}$

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 $2\pi R_{\rm E}$

new circumference:

 $2\pi R_{\rm E} + 1 \,{\rm m}$

circumference at equator:

 $2\pi R_{\rm E}$

new circumference:

 $2\pi R_{\rm E} + 1 \,{\rm m}$

radius of circle with new circumference:

 $2\pi R = 2\pi R_{\rm E} + 1 \,{\rm m}, \text{ and so } R = R_{\rm E} + \frac{1 \,{\rm m}}{2\pi}.$

Outline

Setting the stage

• Making it happen

Overcoming barriers

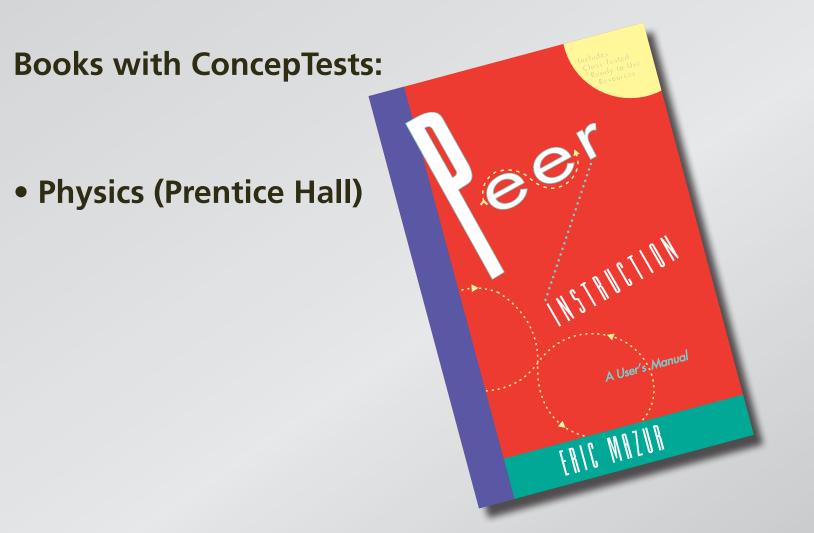
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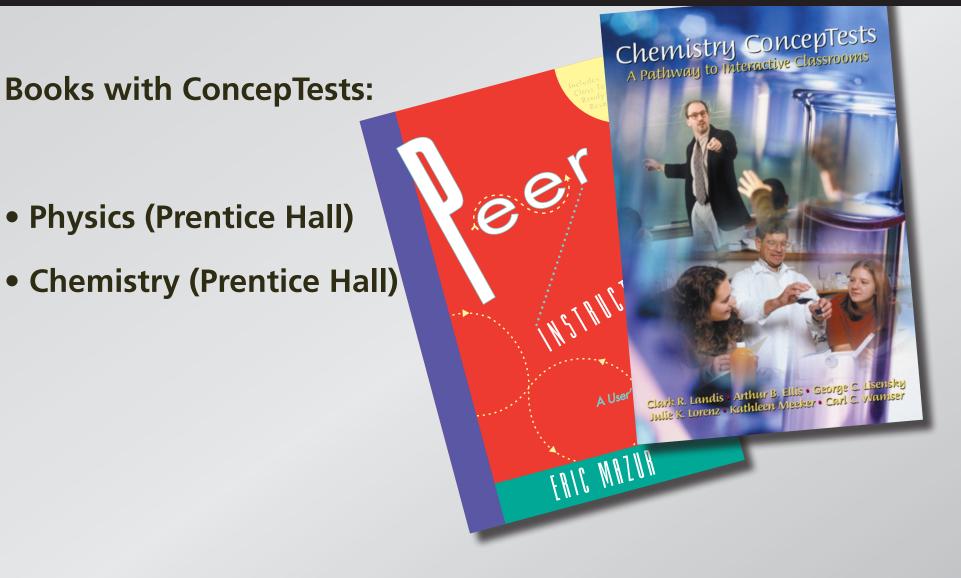
To make it happen, I need to...

(actions to take during course)

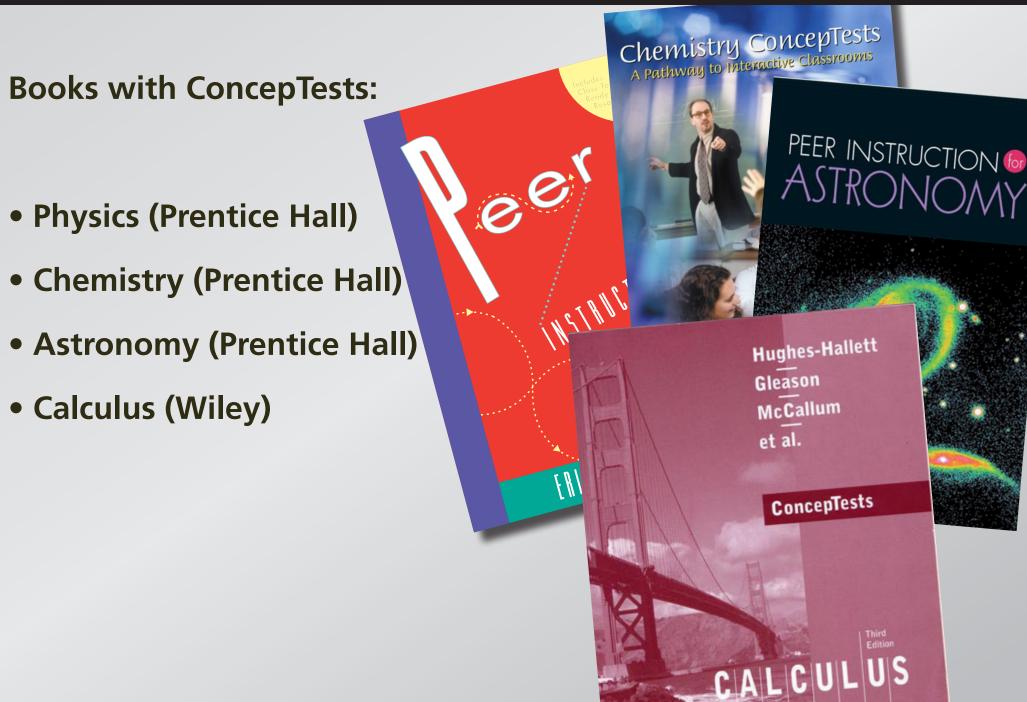
find or develop good questions

know how to manage time



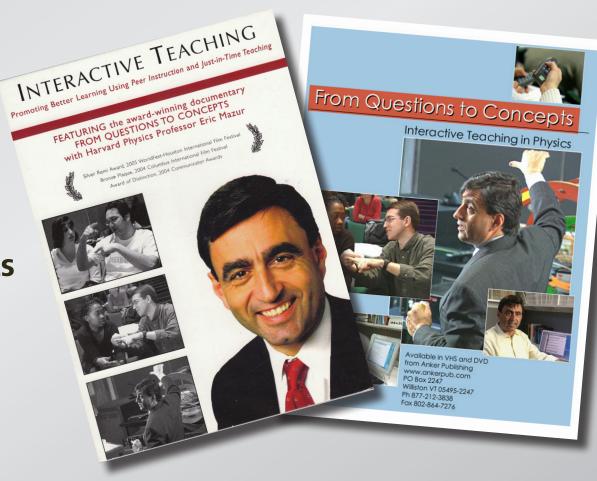






Videos:

- Interactive Teaching DVD
- From questions to concepts



Types of questions

- survey
- discussion
- model testing
- select from list

Which of the following airlines tries to save fuel by suggesting that its passengers use the bathroom before boarding?

- **1. Delta Airlines**
- 2. Lufthansa
- 3. All Nippon Airways
- 4. British Midland Airways
- 5. Air France
- 6. JAL
- 7. Aboriginal Air Services
- 8. Aeroflot
- 9. Are you kidding me? None of the above.

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hole in plate/circumference model

Professor A/B

discussion

airline

fact

hole in plate/circumference model

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airline

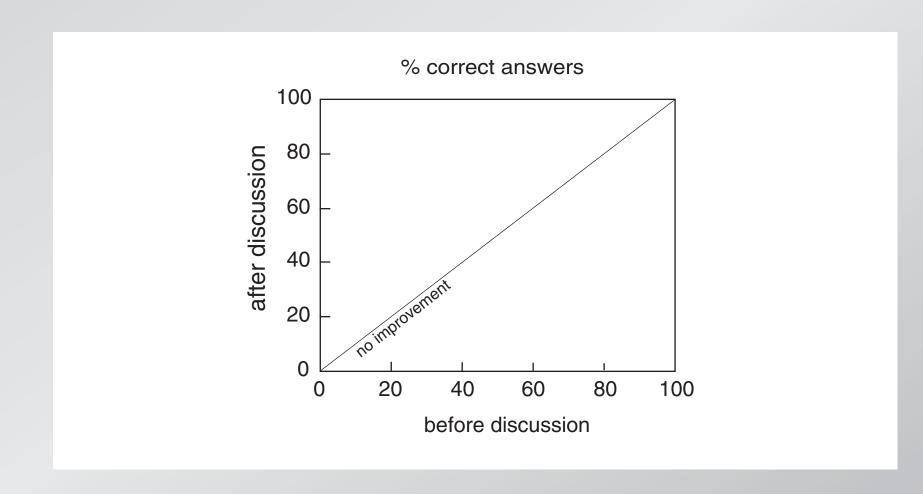
fact

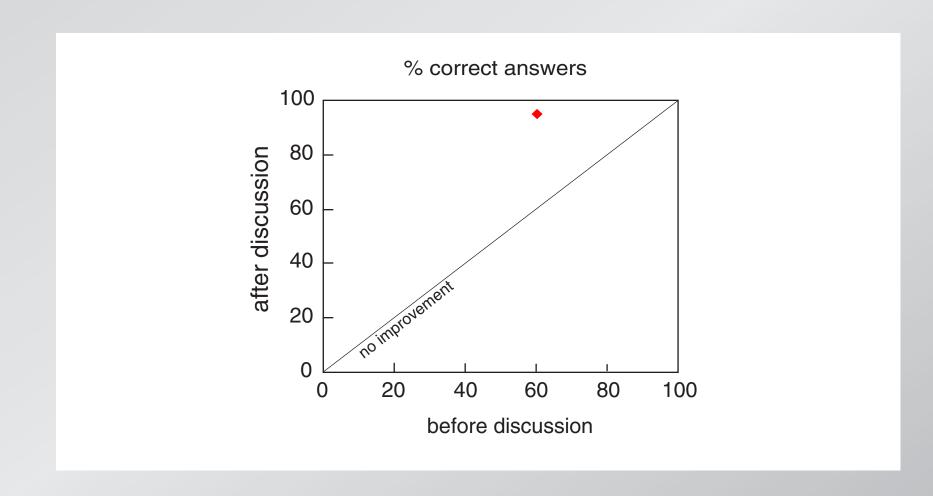
fact-recall not engaging

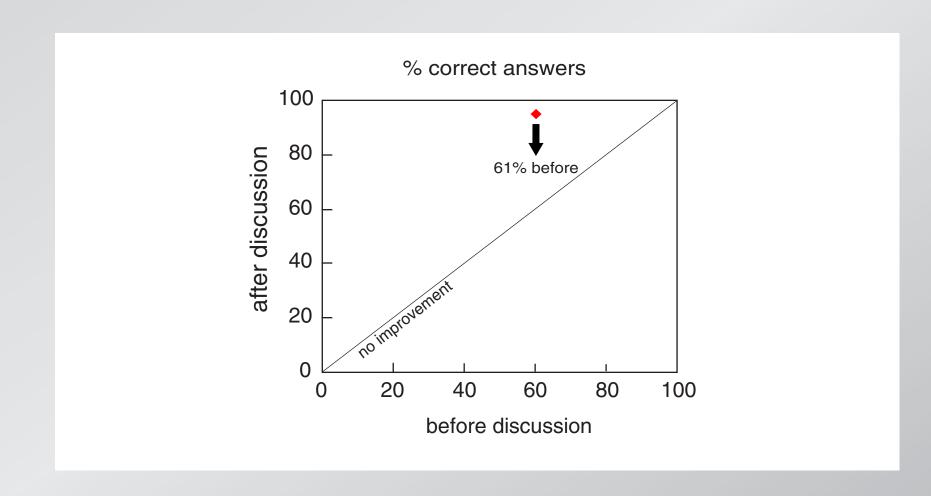
Good conceptual questions (ConcepTests):

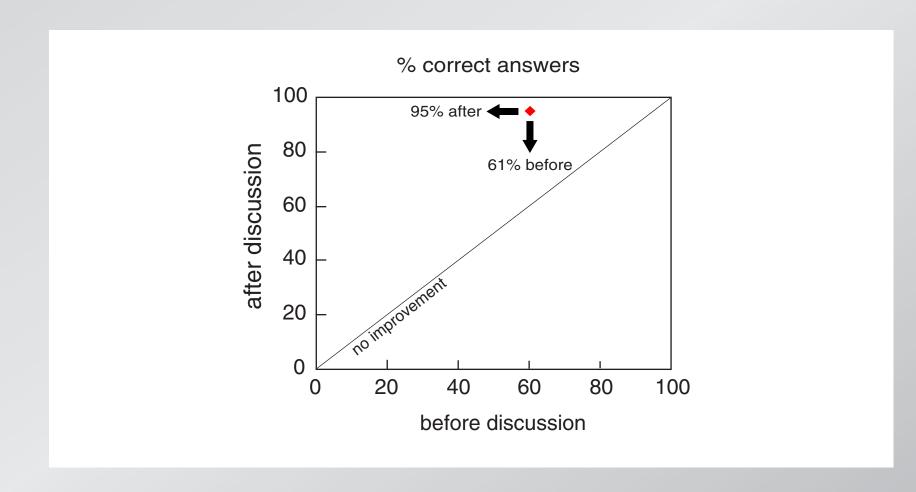
- are based on common student difficulties
- focus on single concept
- require more than "plug and chug"
- are clear and concise
- are of manageable difficulty

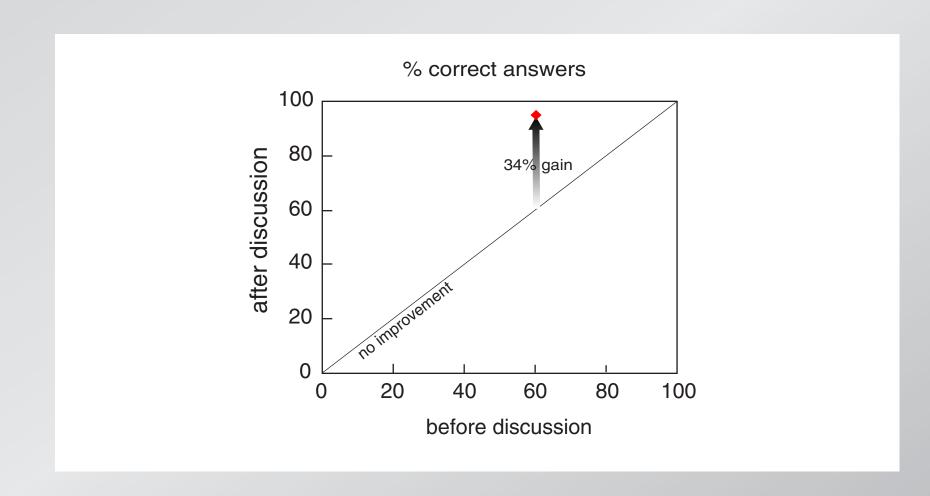
must adjust level to audience

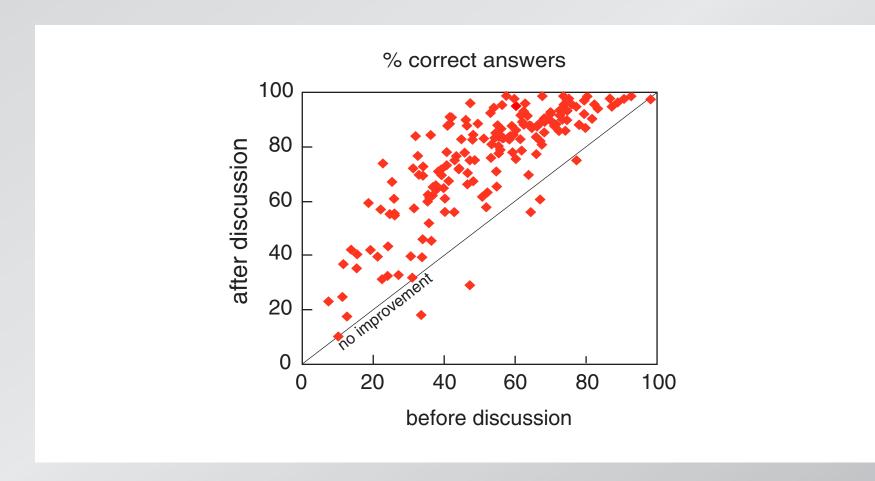


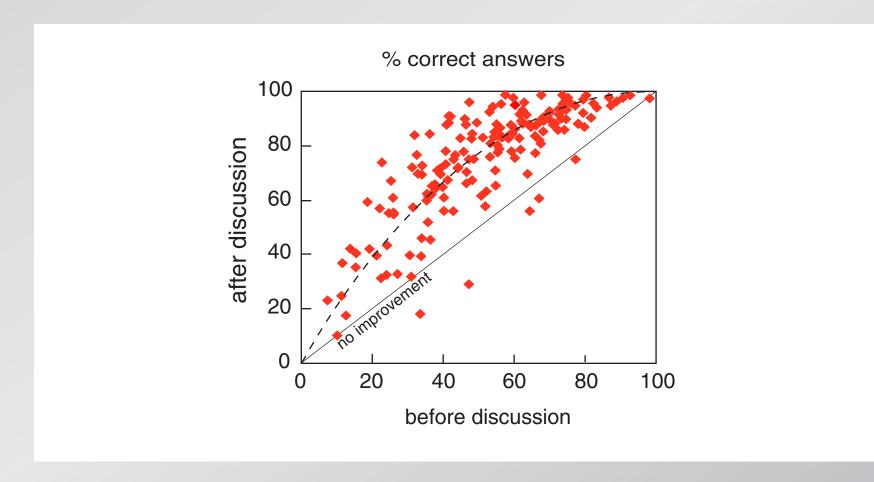


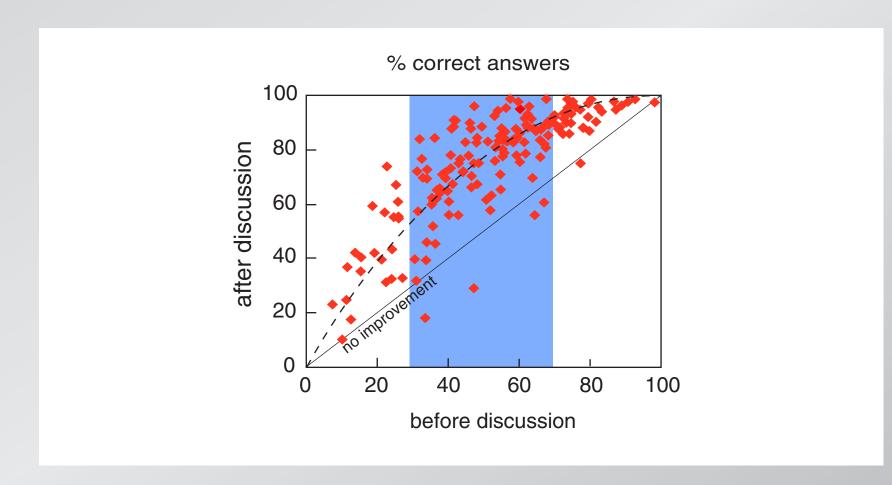




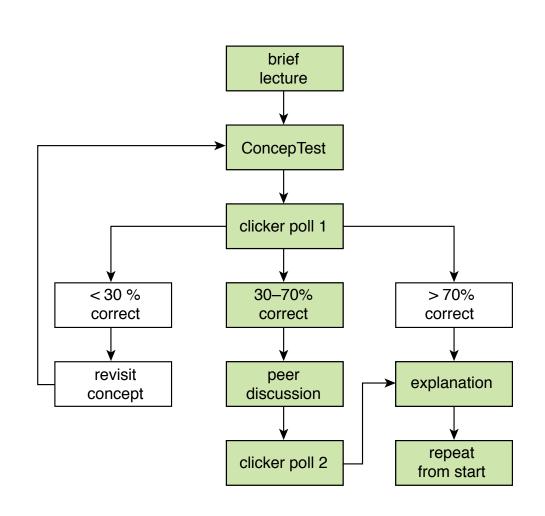








Peer Instruction: a primer



Outline

Setting the stage

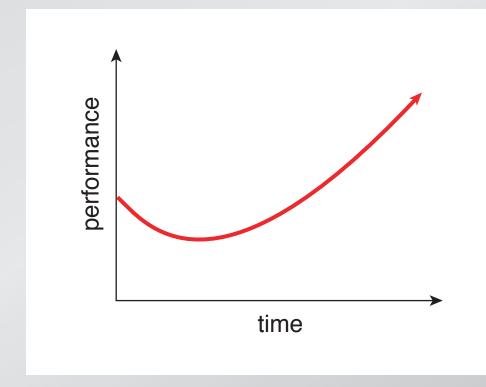
• Making it happen

Overcoming barriers

What are some potential barriers?

- skepticism
- growing pains
- negative feedback
- limited circle of influence

After changing, things might get *worse* before they get better!



Better understanding leads to *more* — not fewer — questions!

(must recognize confusion as step towards understanding)

Things to do:

- take/analyze data
- motivate students
- be prepared for initial adjustments

Why is change so hard?

	"lectures"	PI
coverage	complete	partial
preclass reading	none	cover everything
confusion	little none	substantial
evaluations	known	unknown

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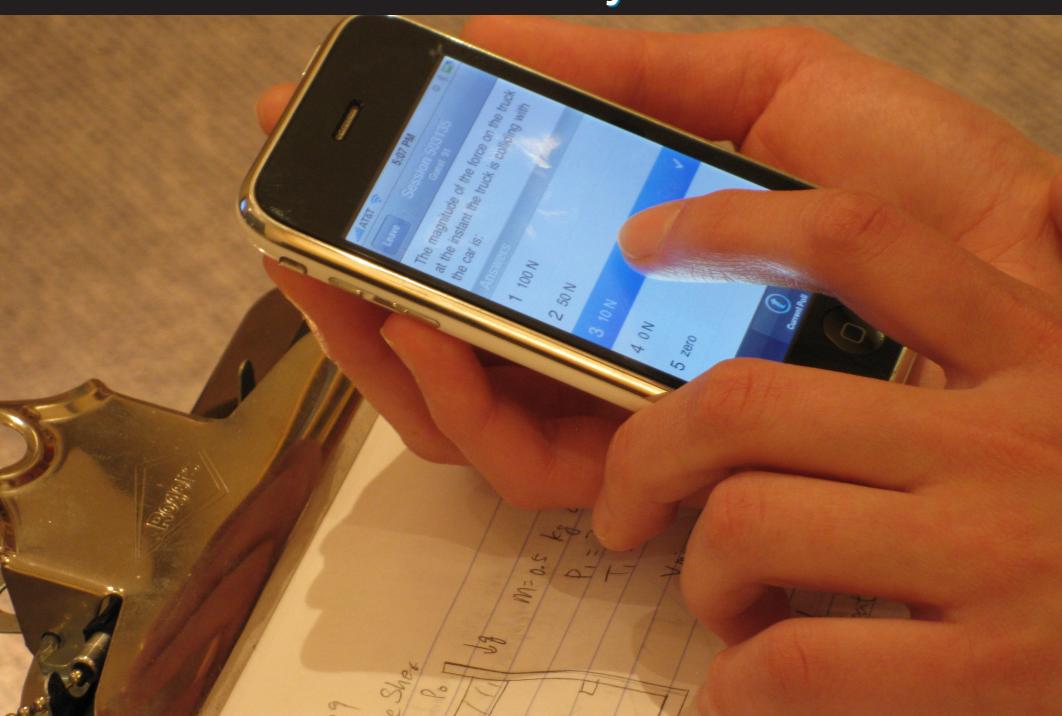
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But PI leads to better learning and retention

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(neither of which is traditionally measured)





Summary

Pl easy to implement (and improves learning gains)

Conclusion

Pl easy to implement (and improves learning gains)

technology facilitates active engagement (but not required)

Conclusion

not just a polling tool, but an engagement tool!

Research Funding:

Pew Charitable Trust, Pearson/Prentice Hall, Davis Foundation, Engineering Information Foundation, Derek Bok Center for Teaching and Learning, National Science Foundation

for a copy of this presentation:

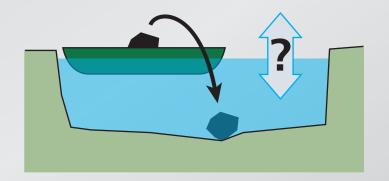
http://mazur-www.harvard.edu

response cards:

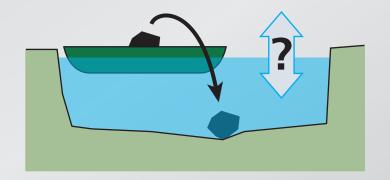
www.turningtechnologies.com



A boat carrying a large boulder is floating on a small pond. The boulder is thrown overboard and sinks to the bottom of the pond.



A boat carrying a large boulder is floating on a small pond. The boulder is thrown overboard and sinks to the bottom of the pond.



After the boulder sinks to the bottom of the pond, the level of the water in the pond is

- 1. higher than
- 2. the same as
- 3. lower than

it was when the boulder was in the boat.



We all make mistakes!

When we hold a page of printed text in front of a mirror, the text on the image in the mirror runs from right to left:

The New York Times

When we hold a page of printed text in front of a mirror, the text on the image in the mirror runs from right to left:

The New York Times

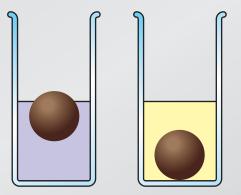
Why is it that right and left are interchanged and not top and bottom? Because:

- **1**. the mirror is oriented vertically.
- 2. we have two eyes in the horizontal plane.
- 3. the Earth's gravitation is directed downward.
- 4. a habit we have when looking at images in a mirror.
- 5. It only appears to run from left to right.

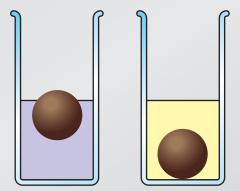


It's "simple" only if you know the answer

Consider an object that floats in water, but sinks in oil. When the object floats in water, most of it is submerged.

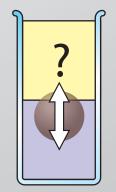


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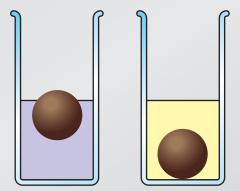


If we slowly pour the oil on top of the water so it completely covers the object, the object

- 1. moves up.
- 2. stays in the same place.
- 3. moves down.

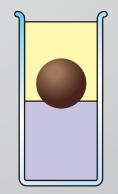


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It's easy to make simple demonstrations fascinating!

The specific heat at constant volume for a monatomic crystal approaches zero at low temperature even though the specific heat for a monatomic gas remains $\frac{3}{2}k$ per atom. Why is this so?

The specific heat at constant volume for a monatomic crystal approaches zero at low temperature even though the specific heat for a monatomic gas remains $\frac{3}{2}k$ per atom. Why is this so?

- 1. Potential energy doesn't play a role for the monatomic gas, but it does for the crystal.
- 2. The particles are indistinguishable in the gas, but not in the crystal.
- 3. The energy difference between allowed states for the crystal is much larger than it is for the atoms.