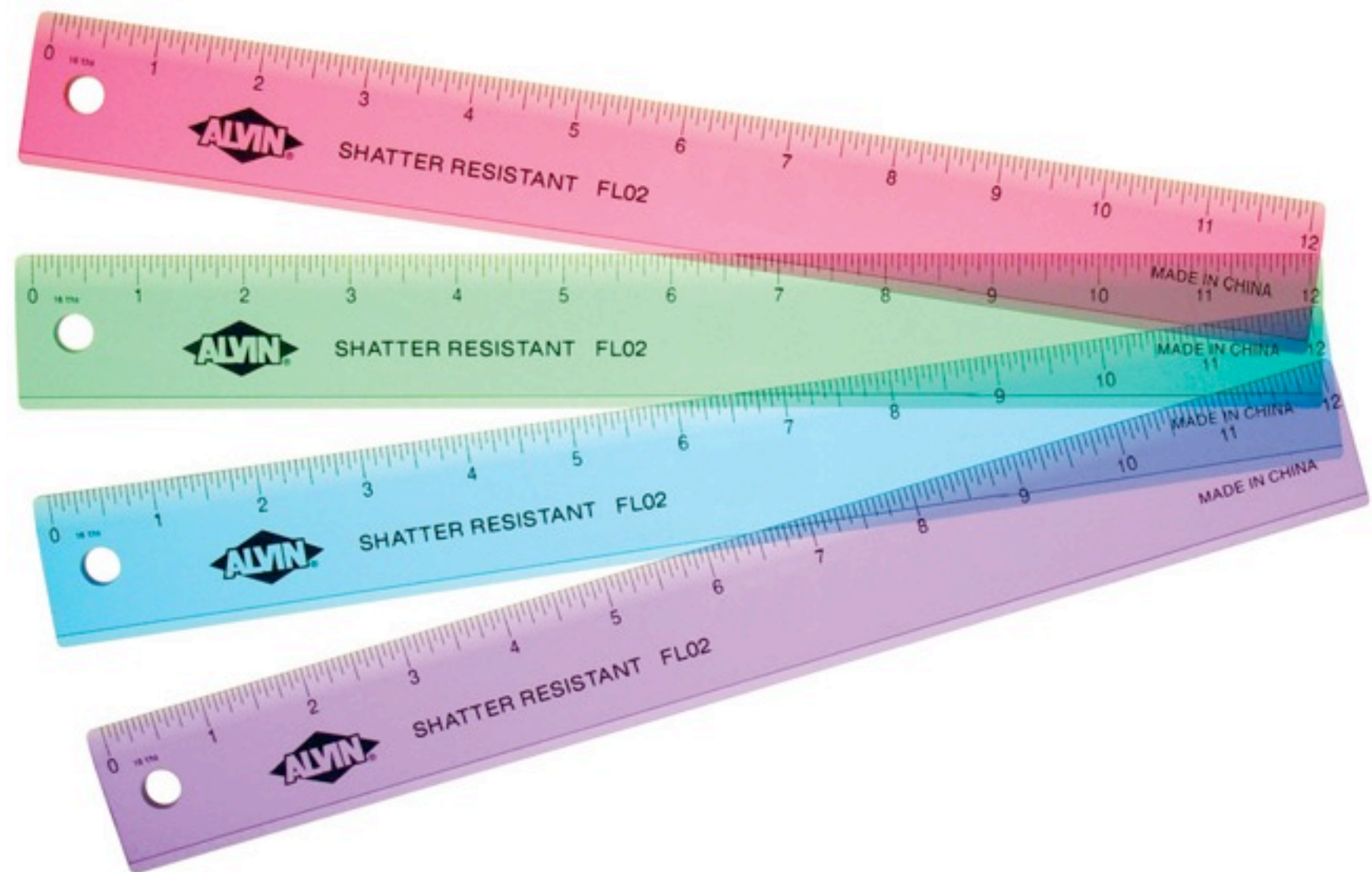
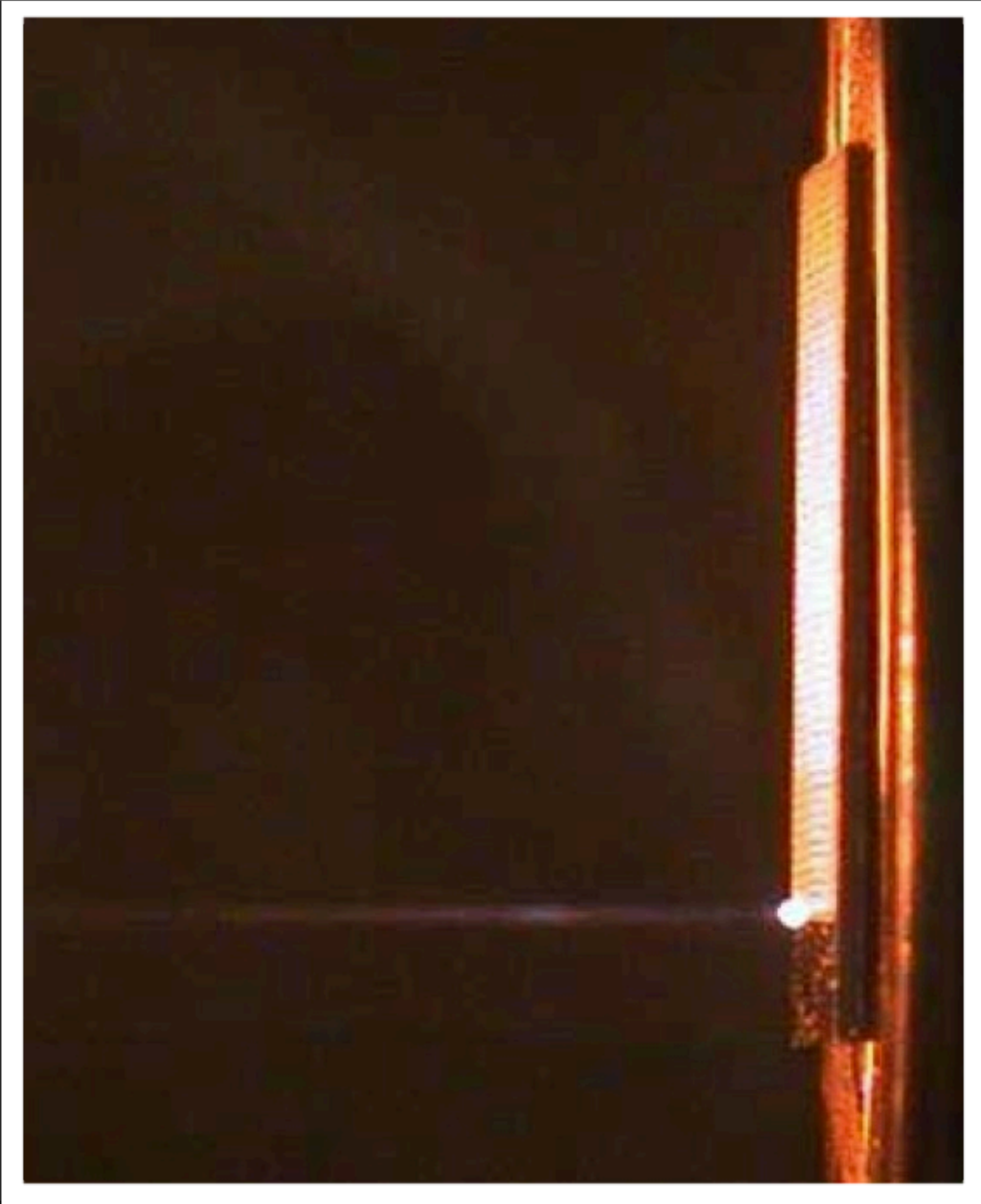
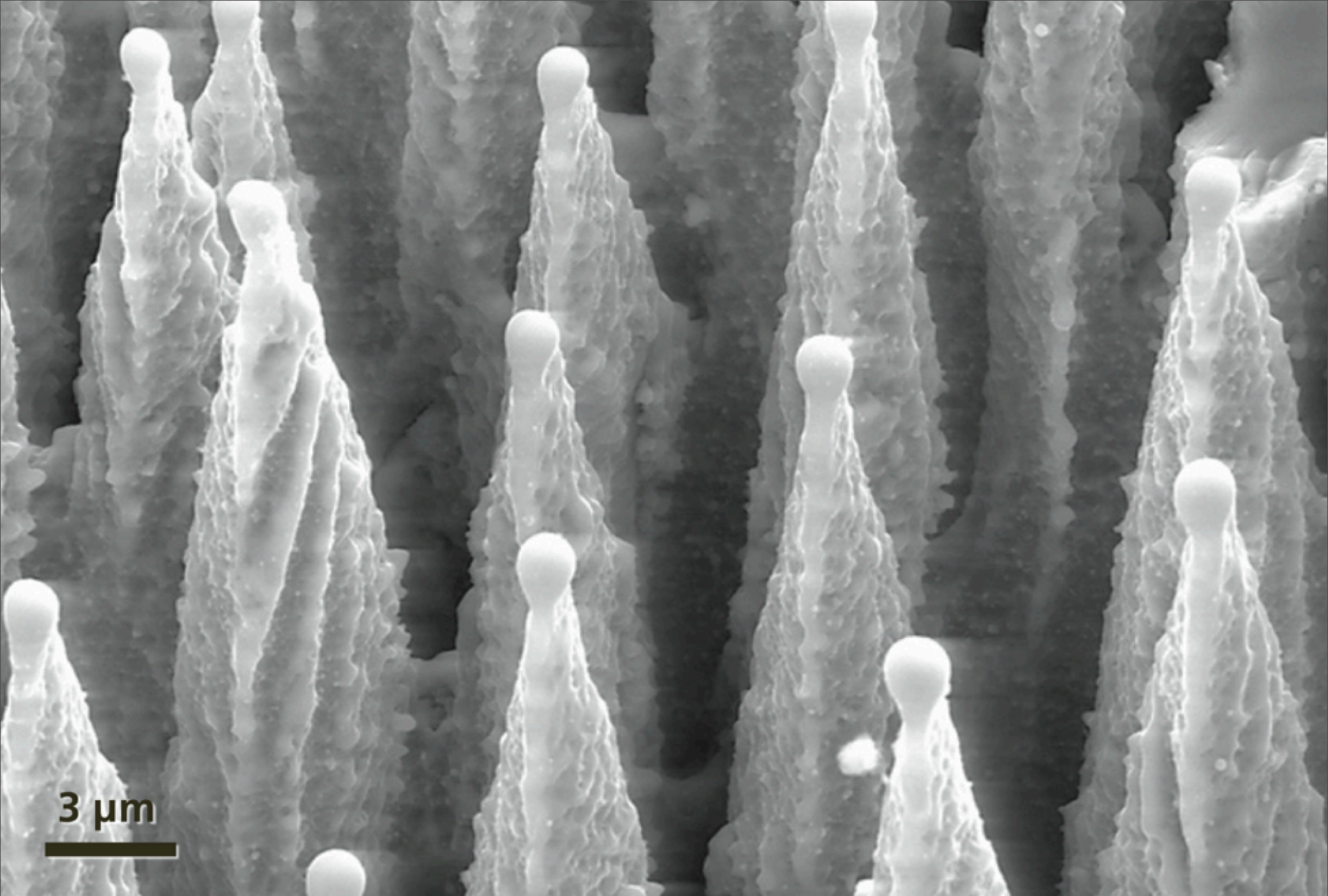


Multidimensional Measurement in Education



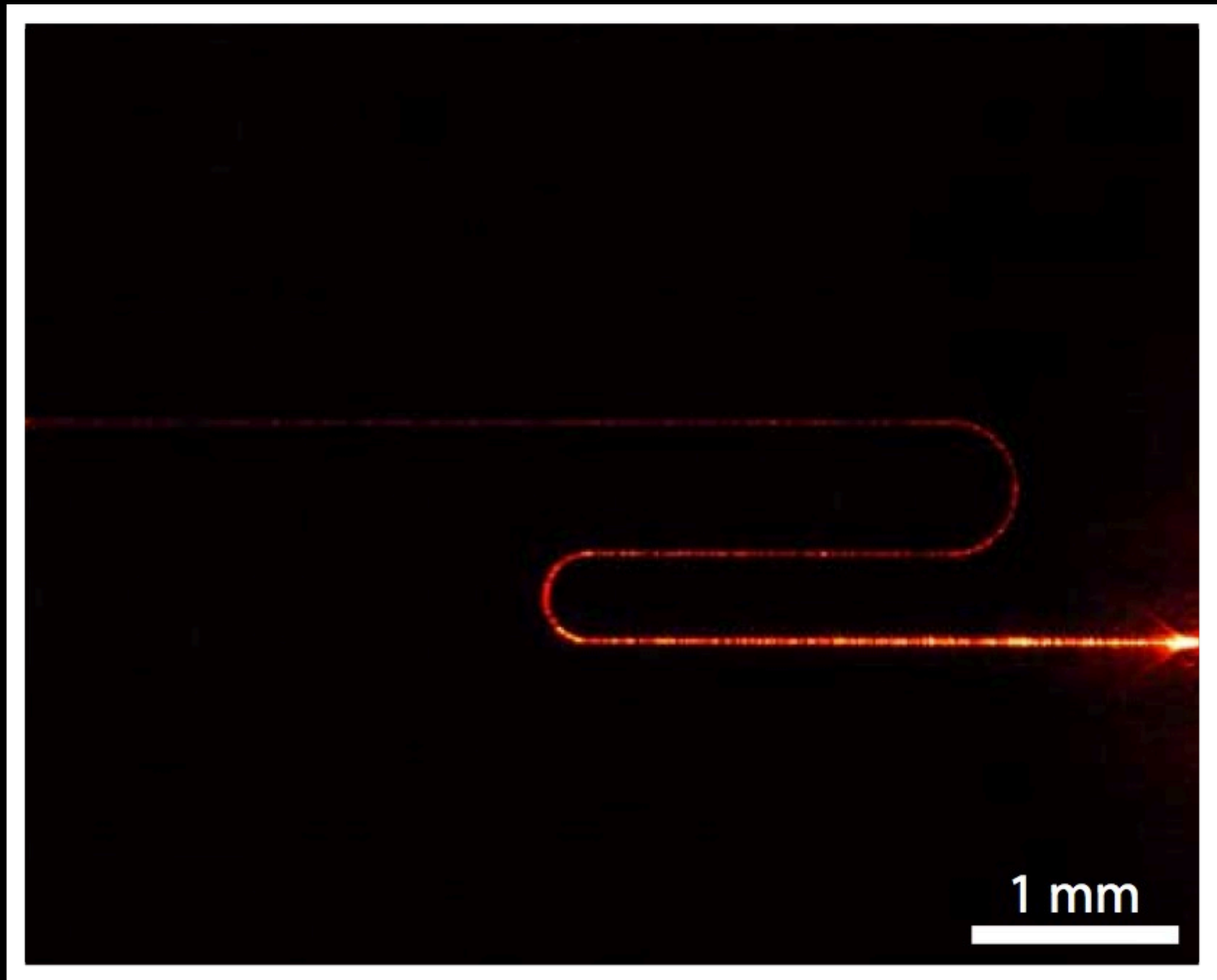


Ultrafast laser physics



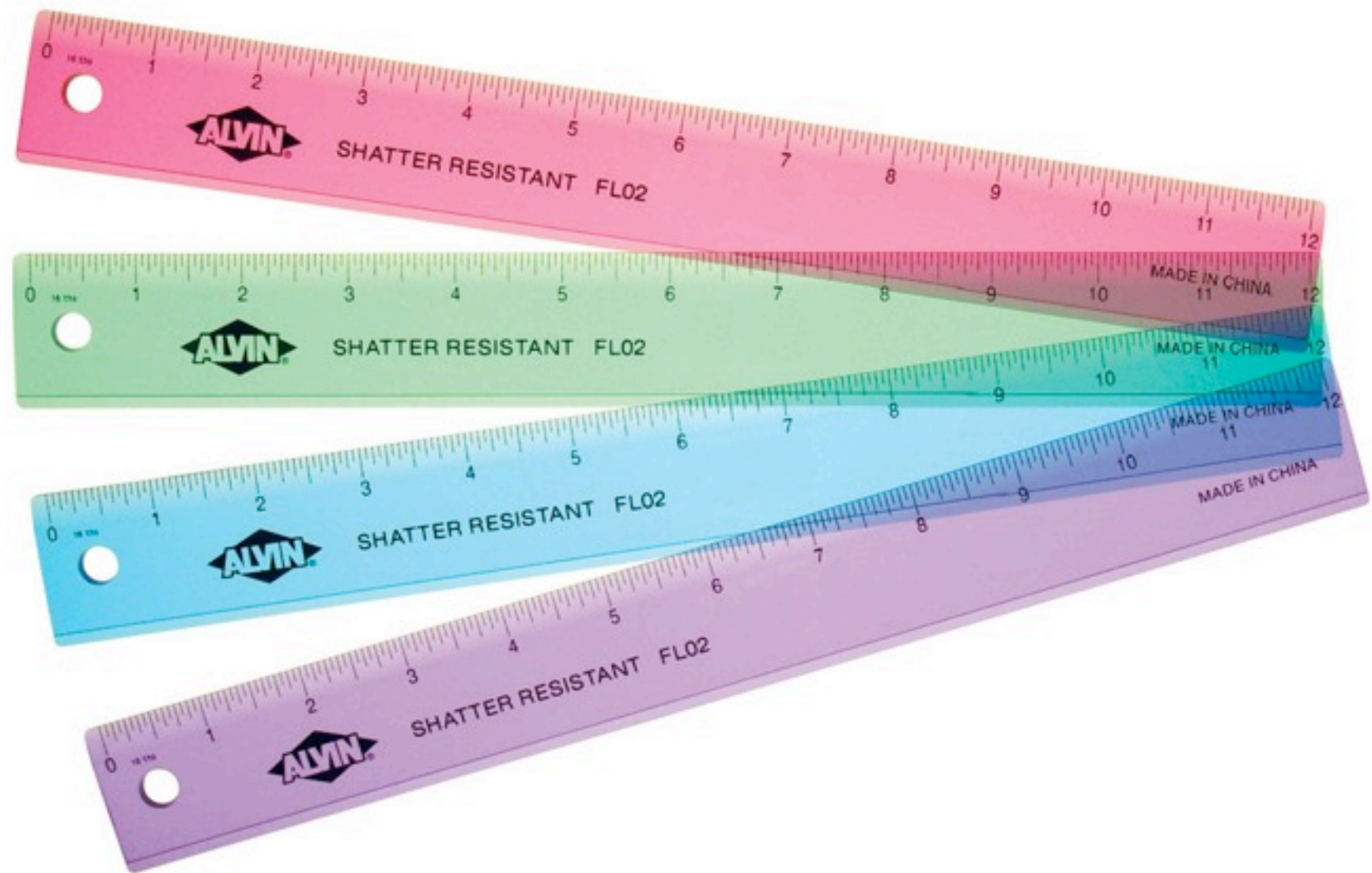
3 μm

Ultrafast laser physics



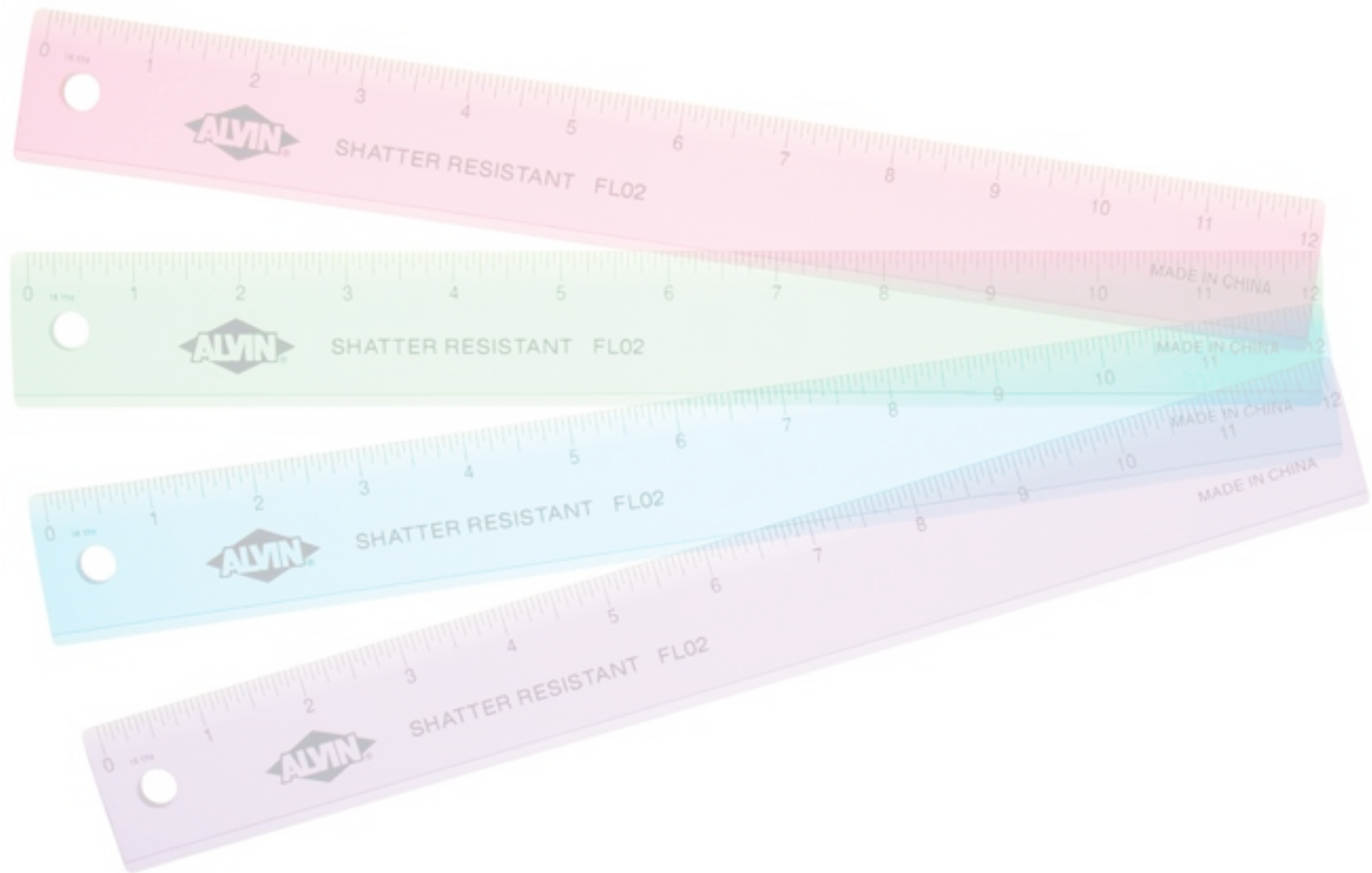
Ultrafast laser physics

Why multidimensional measurement?



Why multidimensional measurement?

Learning



Measurement

Why multidimensional measurement?

Learning : Standardized tests
Force concepts (FCI)

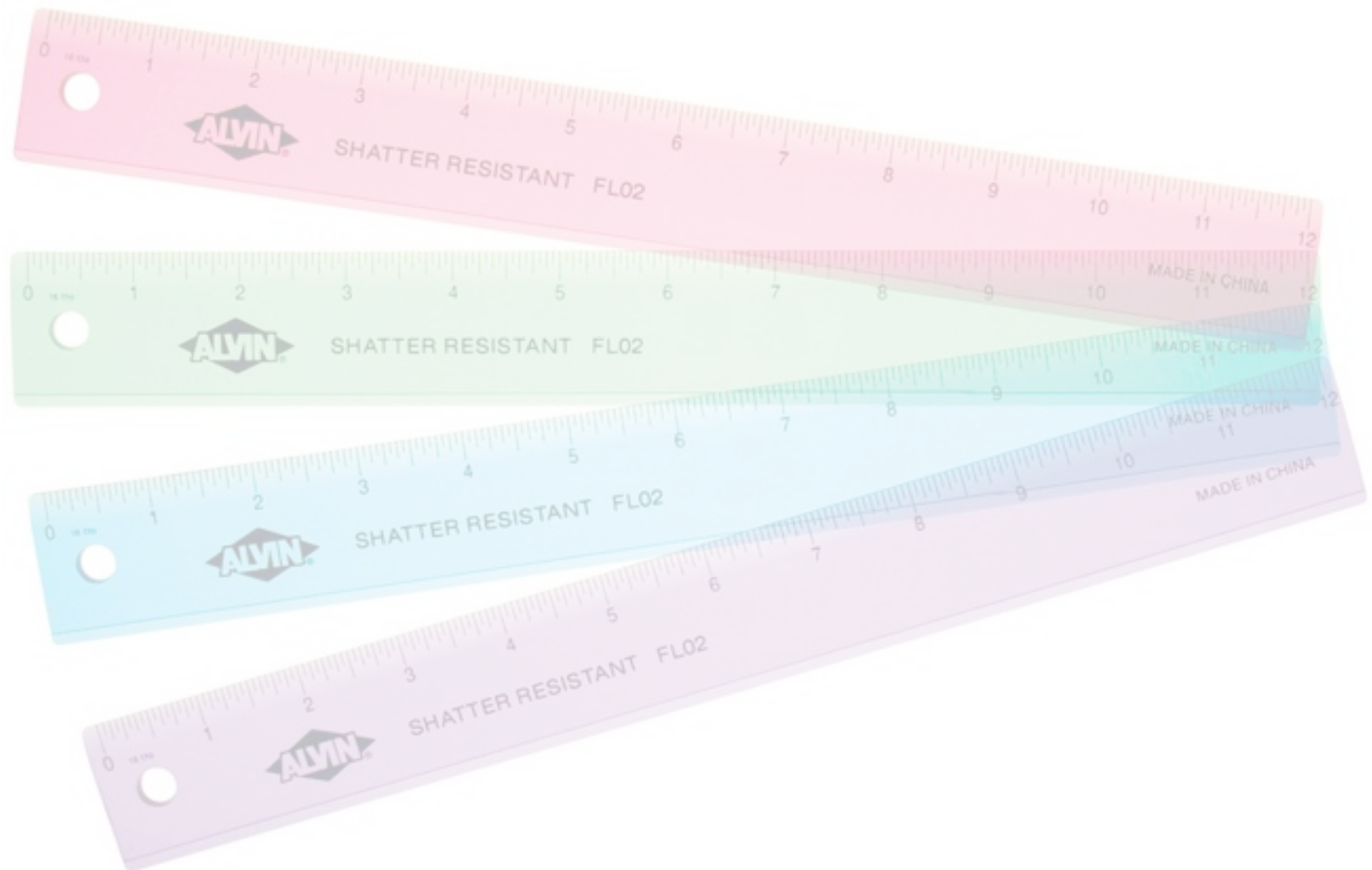
Mechanics concepts and
problem-solving (MBT)

Hestenes, et. al. 1992; Hestenes & Wells. 1992

Why multidimensional measurement?

Learning

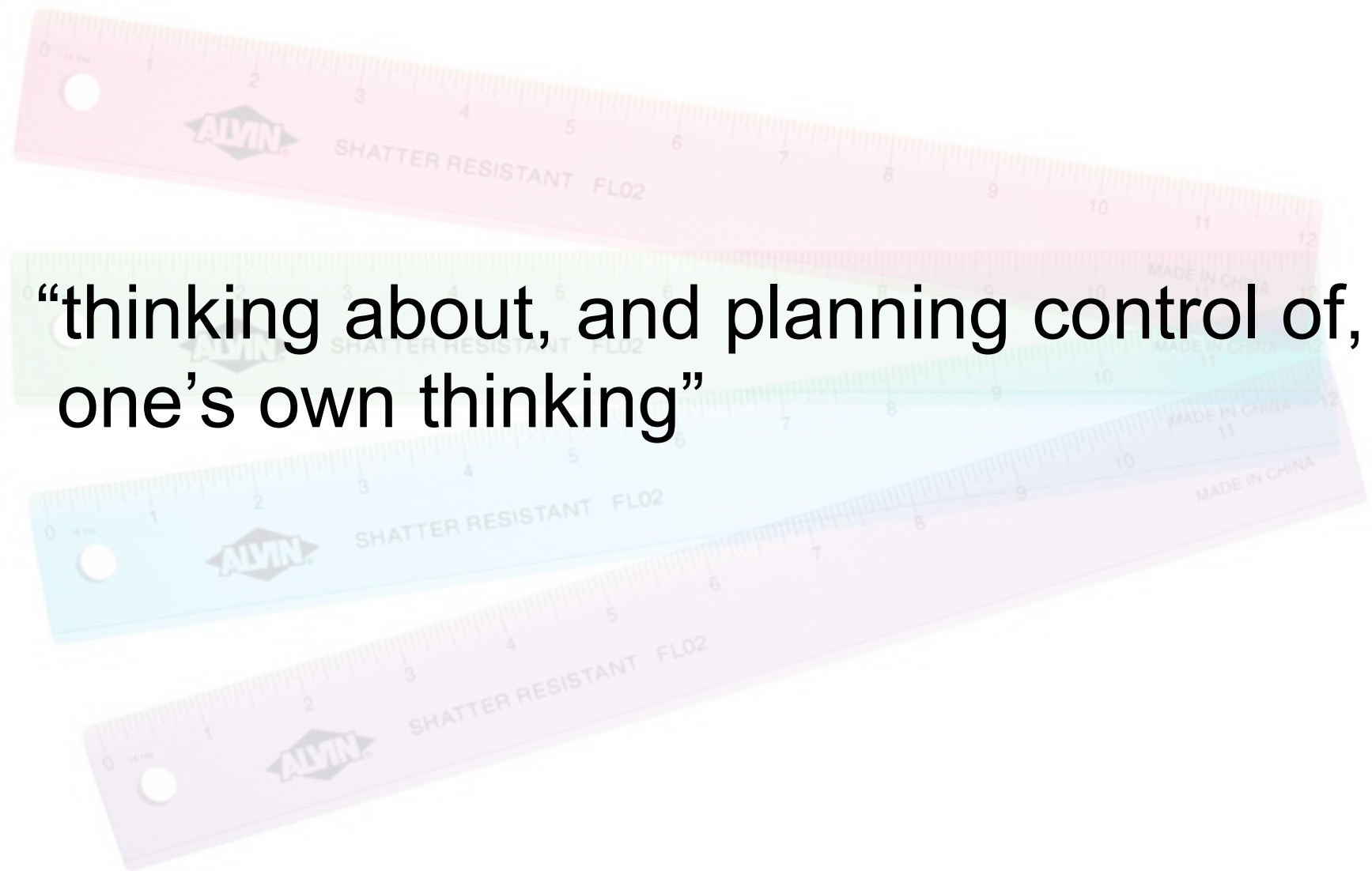
Metacognition



Measurement

Why multidimensional measurement?

Metacognition : “thinking about, and planning control of, one’s own thinking”



Girash, J. (2013)

Why multidimensional measurement?

Metacognition :

“thinking about, and planning control of,
one’s own thinking”

related to transfer of learning
to new situations

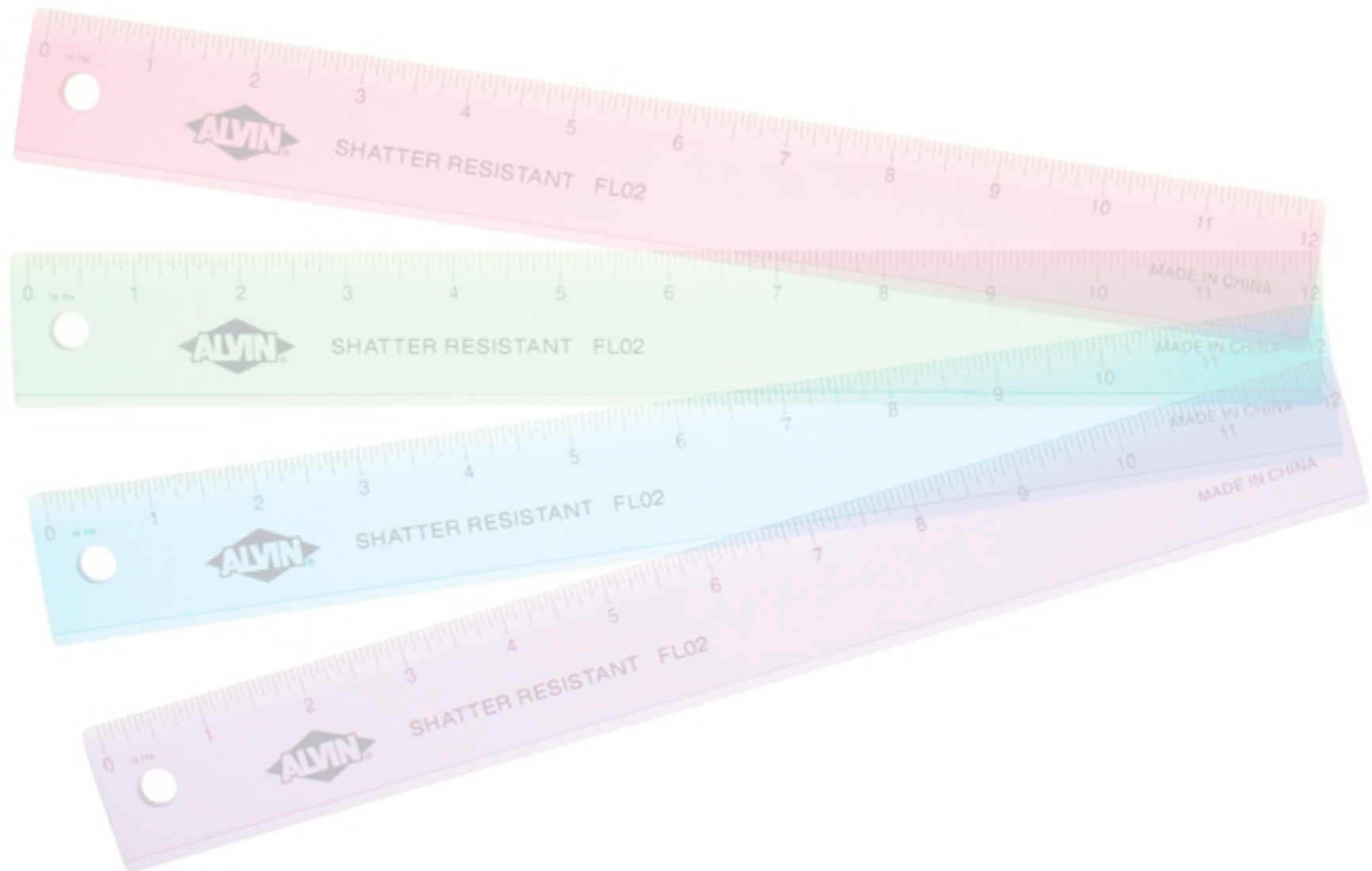
The National Research Council (2000)

Why multidimensional measurement?

Learning

Metacognition

Interaction

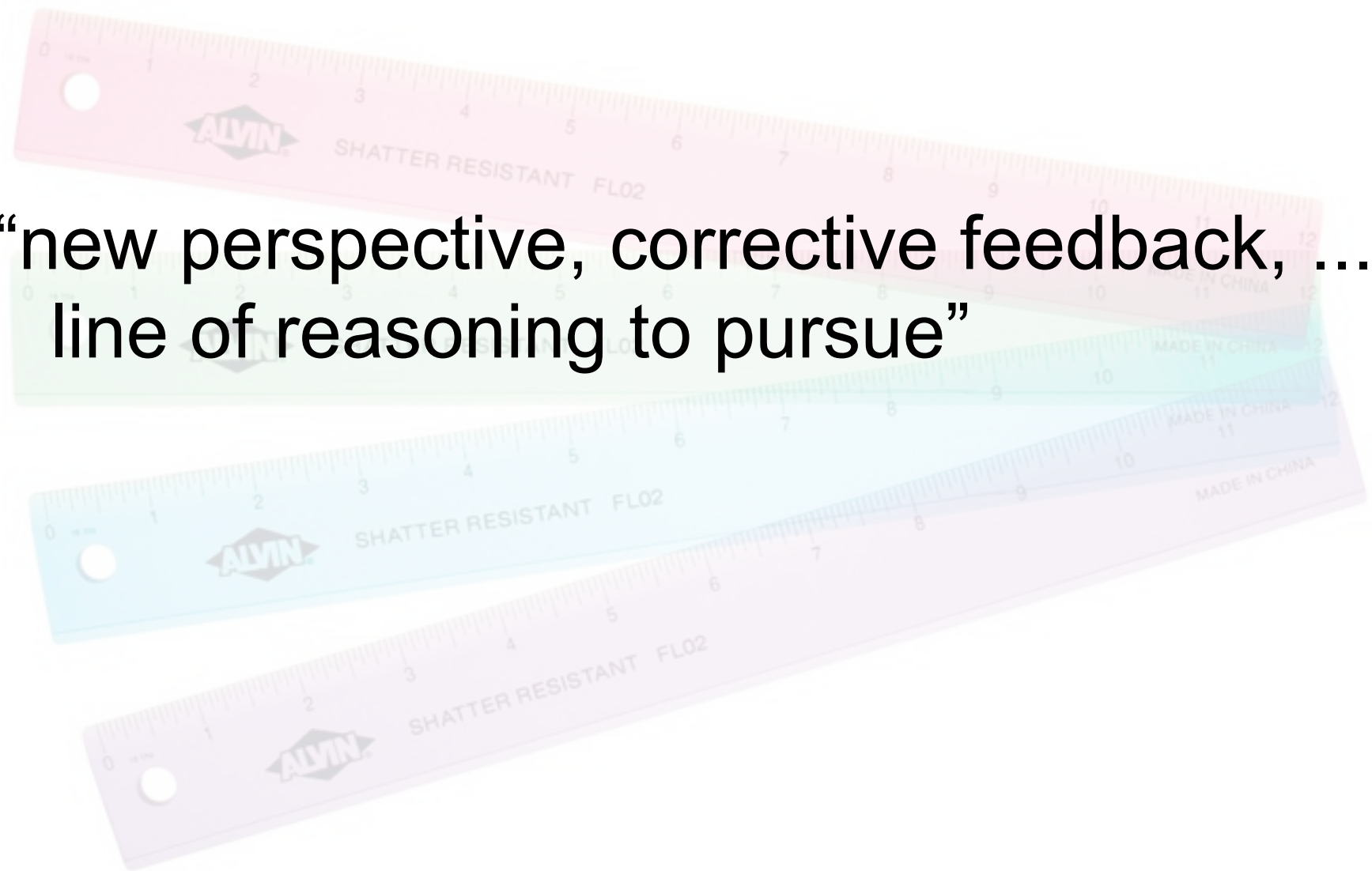


Measurement

Why multidimensional measurement?

“new perspective, corrective feedback, ...
line of reasoning to pursue”

Interaction :



Chi (2009)

Why multidimensional measurement?



“new perspective, corrective feedback, ...
line of reasoning to pursue”

Interaction : “create a shared mental model...
neither...may have been able to come
up with the shared understanding on
her own”

Chi (2009)

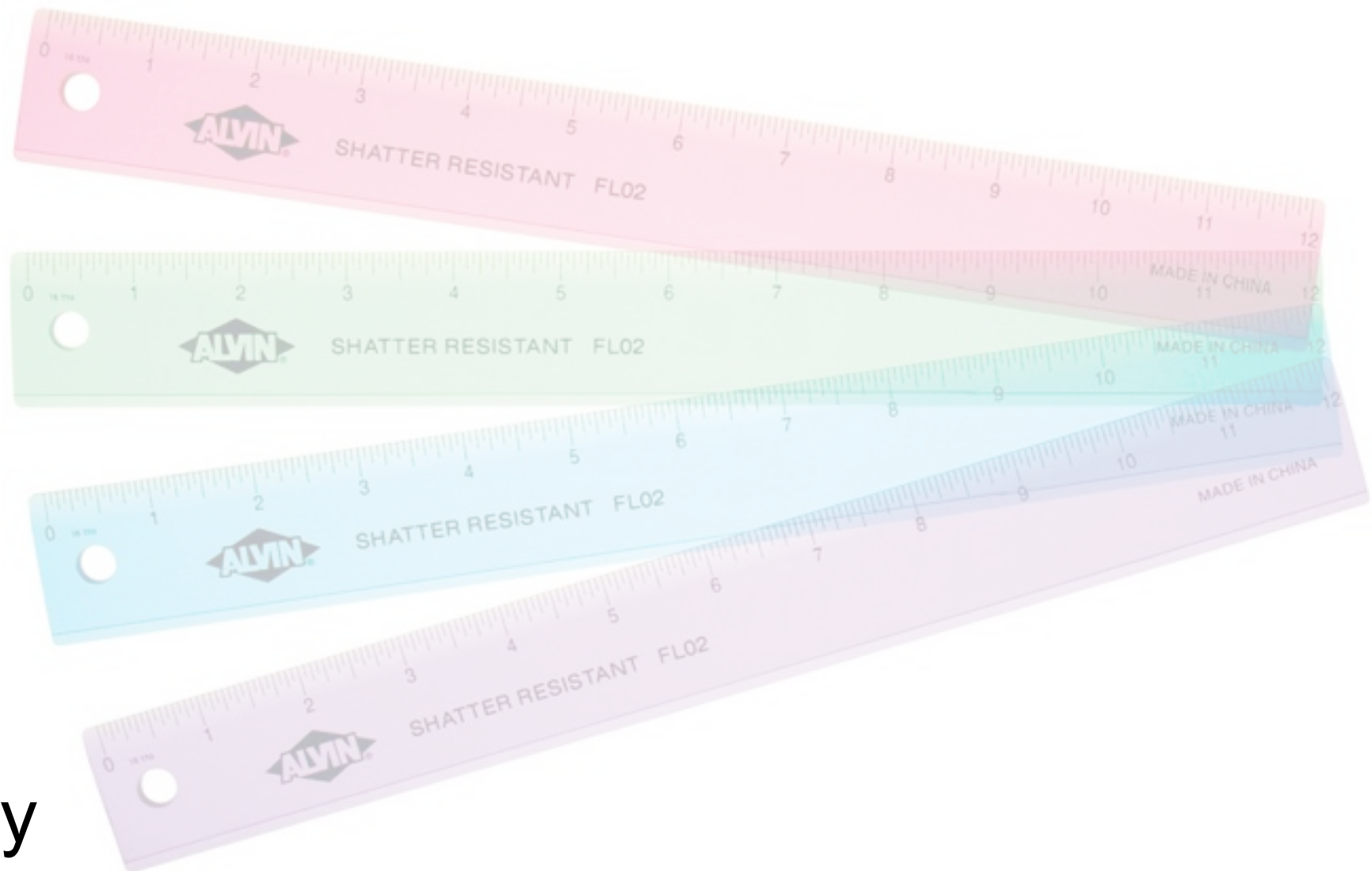
Why multidimensional measurement?

Learning

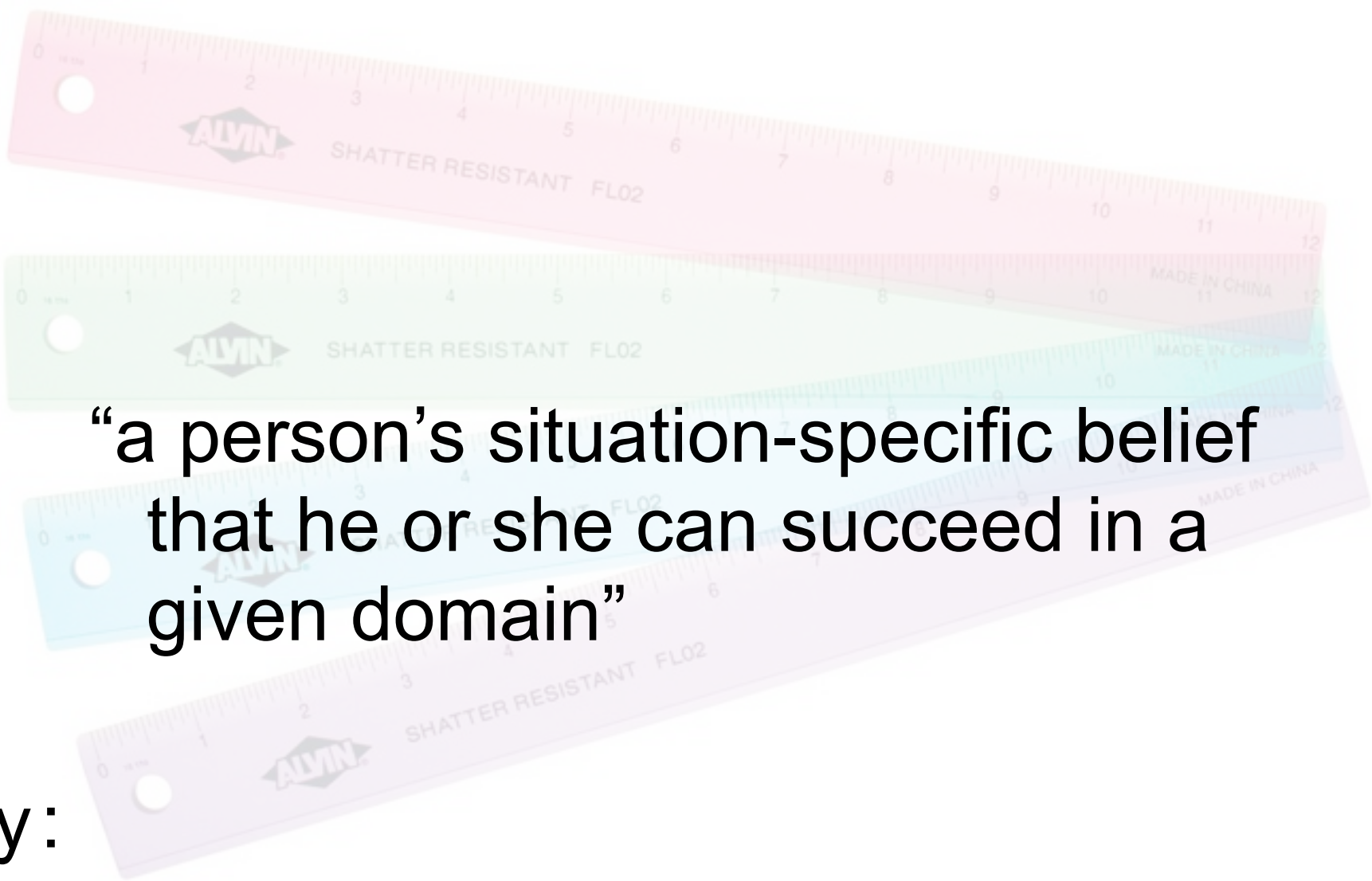
Metacognition

Interaction

Self-efficacy



Why multidimensional measurement?

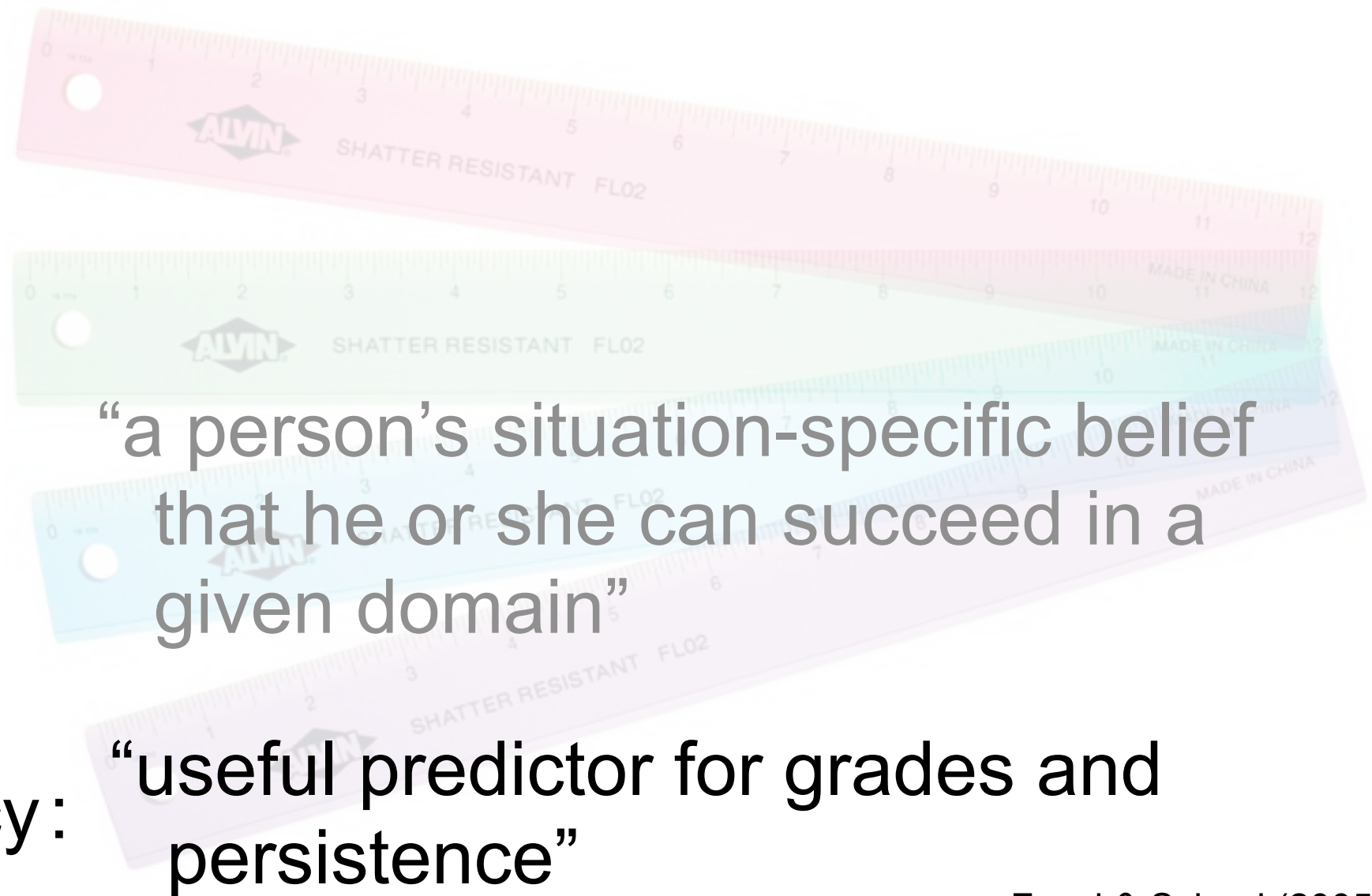


“a person’s situation-specific belief
that he or she can succeed in a
given domain”

Self-efficacy:

Fencl & Scheel (2005)

Why multidimensional measurement?



“a person’s situation-specific belief that he or she can succeed in a given domain”

Self-efficacy: “useful predictor for grades and persistence”

Fencl & Scheel (2005)

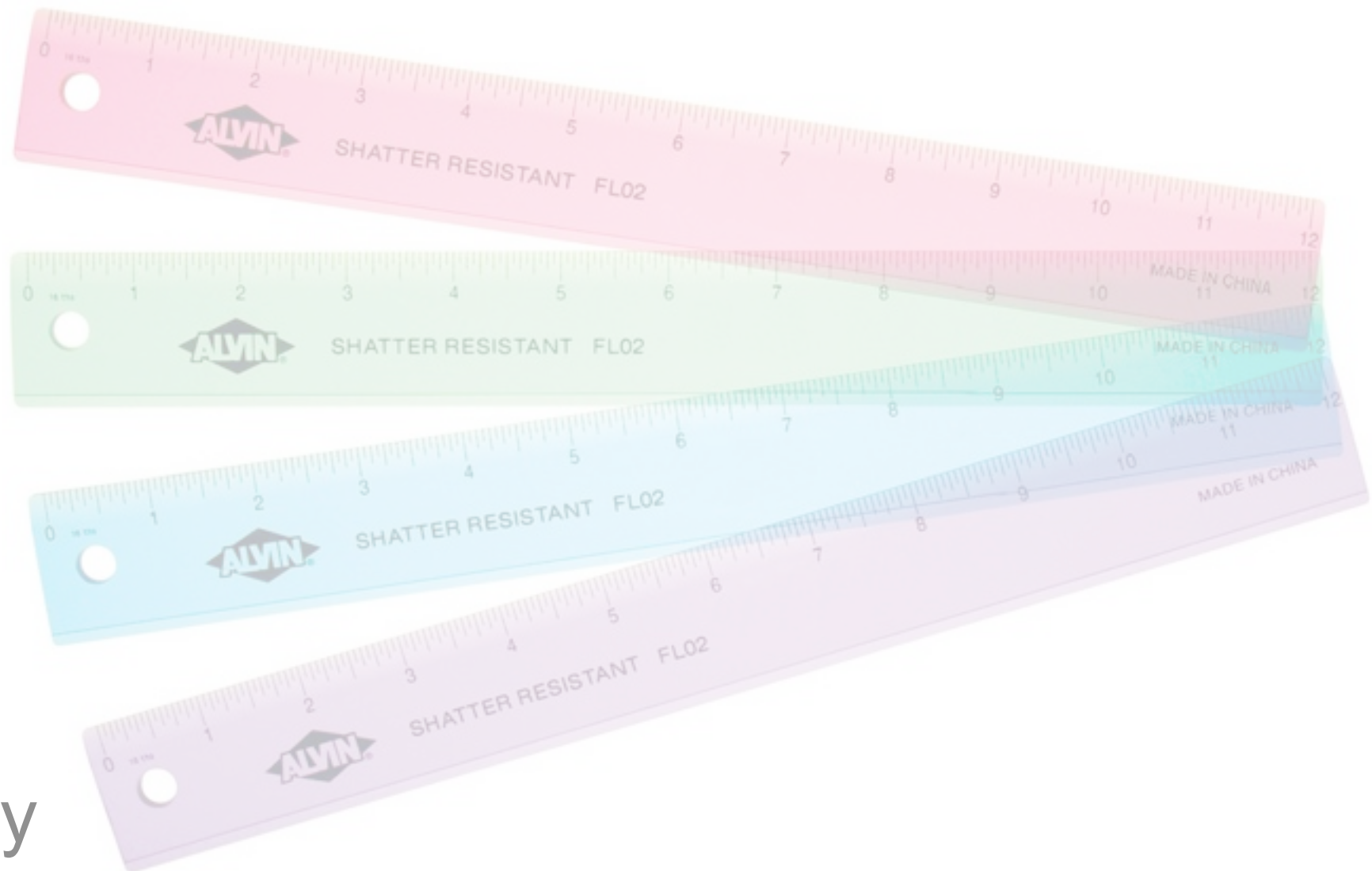
Why multidimensional measurement?

Learning

Metacognition

Interaction

Self-efficacy



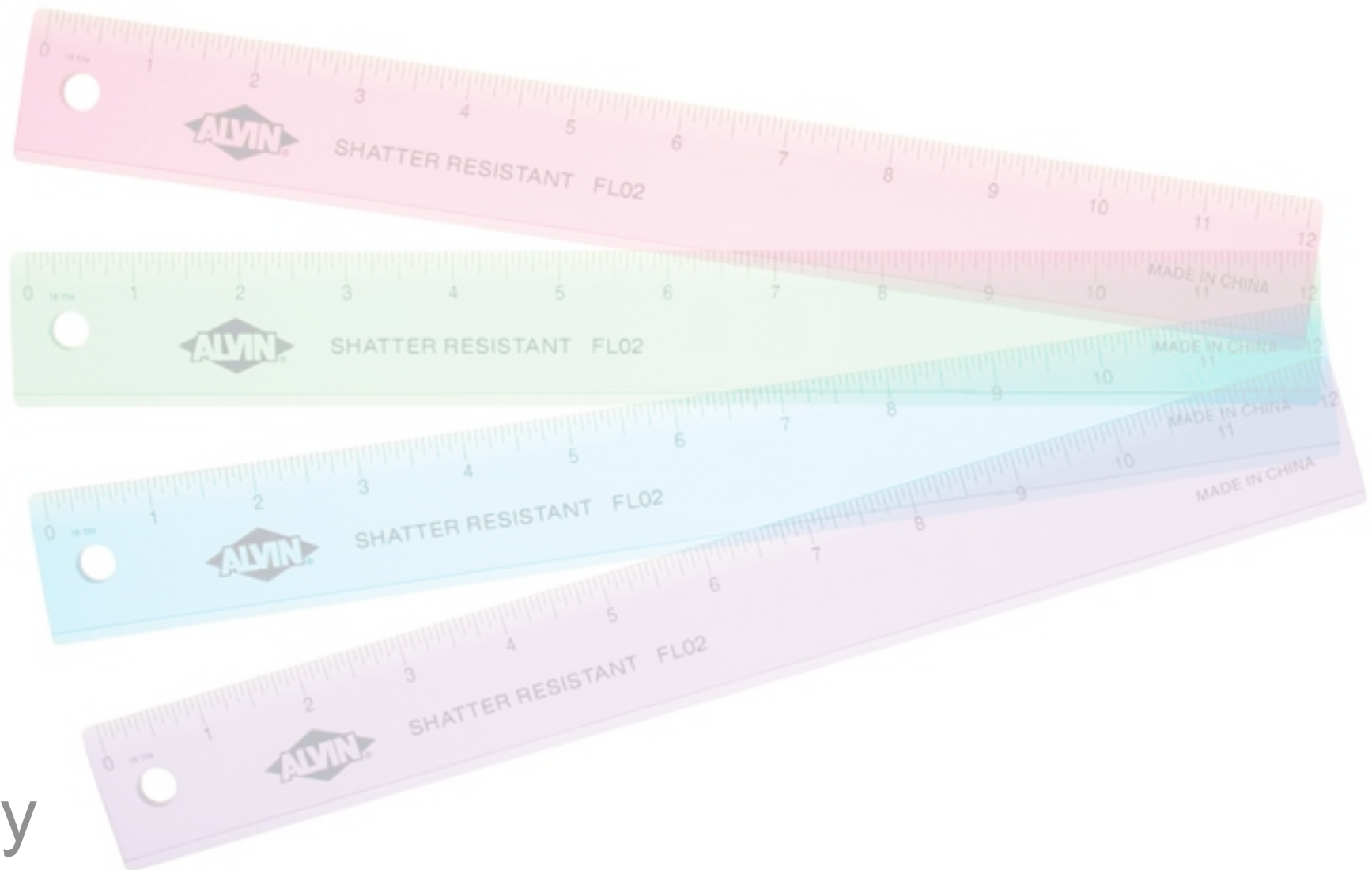
Why multidimensional measurement?

Learning

Metacognition

Interaction

Self-efficacy



Lens to Learning

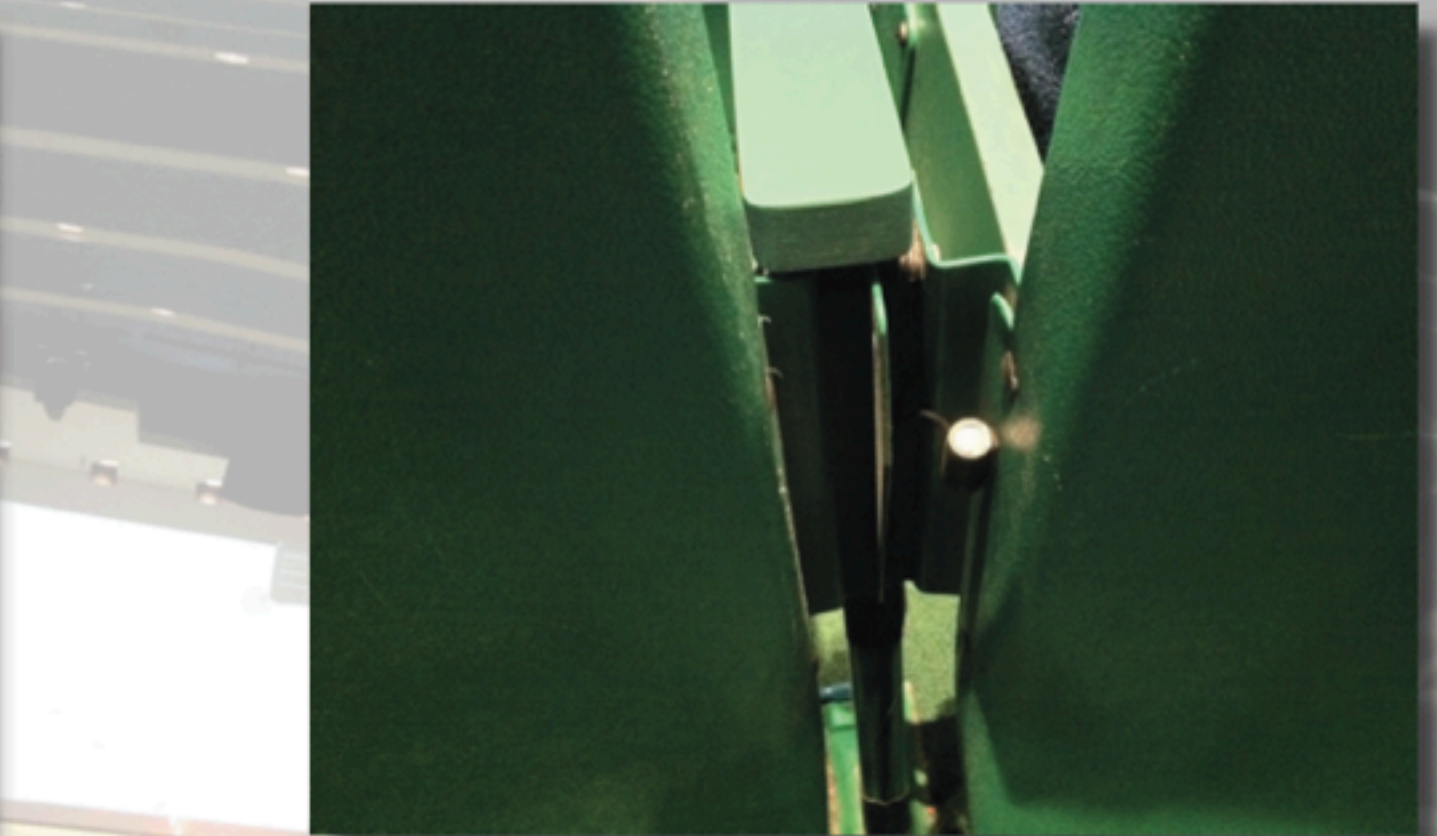
Measurement

Interaction



Measurement

Interaction



Measurement

Interaction



Measurement

Interaction



Measurement

Interaction

Group exams



Measurement

Interaction



Check self/peer assessment

Title *

Participation *

- ☐ Never
- ☐ Rarely
- ☐ Sometimes
- ☐ About half the time
- ☐ Most of the time
- ☐ All the time

Participates fully in team activities.

Well-prepared *

- ☐ Never
- ☐ Rarely
- ☐ Sometimes
- ☐ About half the time
- ☐ Most of the time



Peer Assessment

[Home](#) > [Check self/peer assessment](#)

Check self/peer assessment

| Team | Teammate Name | Relative Contribution |
|---------|---------------|--------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| Team 19 | Jerry | * <input type="radio"/> -3 <input type="radio"/> -2 <input type="radio"/> -1 <input type="radio"/> 0 <input checked="" type="radio"/> 1 <input type="radio"/> 2 <input type="radio"/> 3 |
| Team 19 | Duo | * <input type="radio"/> -3 <input type="radio"/> -2 <input type="radio"/> -1 <input type="radio"/> 0 <input type="radio"/> 1 <input checked="" type="radio"/> 2 <input type="radio"/> 3 |
| Team 19 | Helan | * <input type="radio"/> -3 <input type="radio"/> -2 <input type="radio"/> -1 <input checked="" type="radio"/> 0 <input type="radio"/> 1 <input type="radio"/> 2 <input type="radio"/> 3 |



Positive dynamics:

Partnering. Good at motivating others to work long hours to meet a deadline

Empathizing. Takes time to understand perspectives when differences of opinion occur

Negative dynamics:

Dismissive. Makes an effort to listen to team members' ideas (not dismissive)

Altringer, B. (2013)

The NB system: text annotation

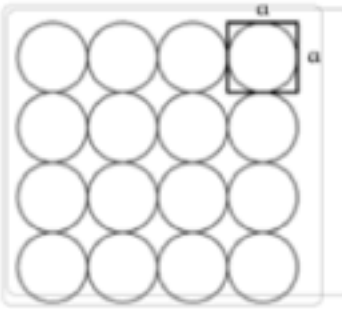
Home r20-dimensions-light-bending.pdf r19-dimensions-hydrogen.pdf 2 global comments New ...

94

5.3.2 Atomic sizes and substance densities

Hydrogen has a diameter of 1\AA . A useful consequence is the rule of thumb is that a typical interatomic spacing is 3\AA . This approximation gives a reasonable approximation for the densities of substances, as this section explains.

Let A be the atomic mass of the atom; it is (roughly) the number of protons and neutrons in the nucleus. Although A is called a mass, it is dimensionless. Each atom occupies a cube of side length $a \sim 3\text{\AA}$, and has mass $A m_{\text{proton}}$. The density of the substance is

$$\rho = \frac{\text{mass}}{\text{volume}} \sim \frac{A m_{\text{proton}}}{(3\text{\AA})^3}.$$


You do not need to remember or look up m_{proton} if you multiply this fraction by unity in the form of N_A/N_A , where N_A is Avogadro's number:

$$\rho \sim \frac{A m_{\text{proton}} N_A}{(3\text{\AA})^3 \times N_A}.$$

The numerator is $A \text{ g}$, because that is how N_A is defined. The denominator is

$$3 \cdot 10^{-23} \text{ cm}^3 \times 6 \cdot 10^{23} = 18.$$

So instead of remembering m_{proton} , you need to remember N_A . However, N_A is more familiar than m_{proton} because N_A arises in chemistry and physics. Using N_A also emphasizes the connection between microscopic and macroscopic values. Carrying out the calculations:

$$\rho \sim \frac{A}{18} \text{ g cm}^{-3}.$$

Zyto et. al.

The NB system: text annotation

Home
r20-dimensions-light-bending.pdf
r19-dimensions-hydrogen.pdf
2 global comments
New ...

94

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$$\rho \sim \frac{A}{18} \text{ g cm}^{-3}.$$

Polls
Spotlight
Notes

6 note(s) 2 ?

Anonymous 18 Mar, 02:06PM options...

Class

How did we get 3 from 1?

Anonymous 18 Mar, 06:36PM options...

Class

The 1 was the diameter of hydrogen. The 3 is typical interatomic spacing.

Anonymous 18 Mar, 07:28PM options...

Class

The 3 and 1 represent 2 different values- 3A is the typical space between all atoms while 1A is the diameter for Hydrogen specifically.

Anonymous 18 Mar, 07:42PM options...

Class 2 ?

So hydrogen atoms are spaced by only 1A when most other atoms are typically spaced between 3A? Am I following you correctly?

Anonymous 18 Mar, 09:15PM options...

Class

Why doesn't this vary significantly based on the size of the atom? Or is three just the overall average.

Instructor 19 Mar, 09:13AM options...

Class

As you say, it is based on the size of the atom. Hydrogen is at the very small end, and uranium is at the large end. 3 Angstroms is a good average size to use for the common atoms in ordinary substances. As a *very* rough approximation, think of the diameter as 1 Angstrom per shell. (The number of shells is the row number in the periodic table.)

1 note(s)

Too many 's' in this sentence.

1 note(s)

This paragraph's wording is a little bit confusing.

1 note(s)

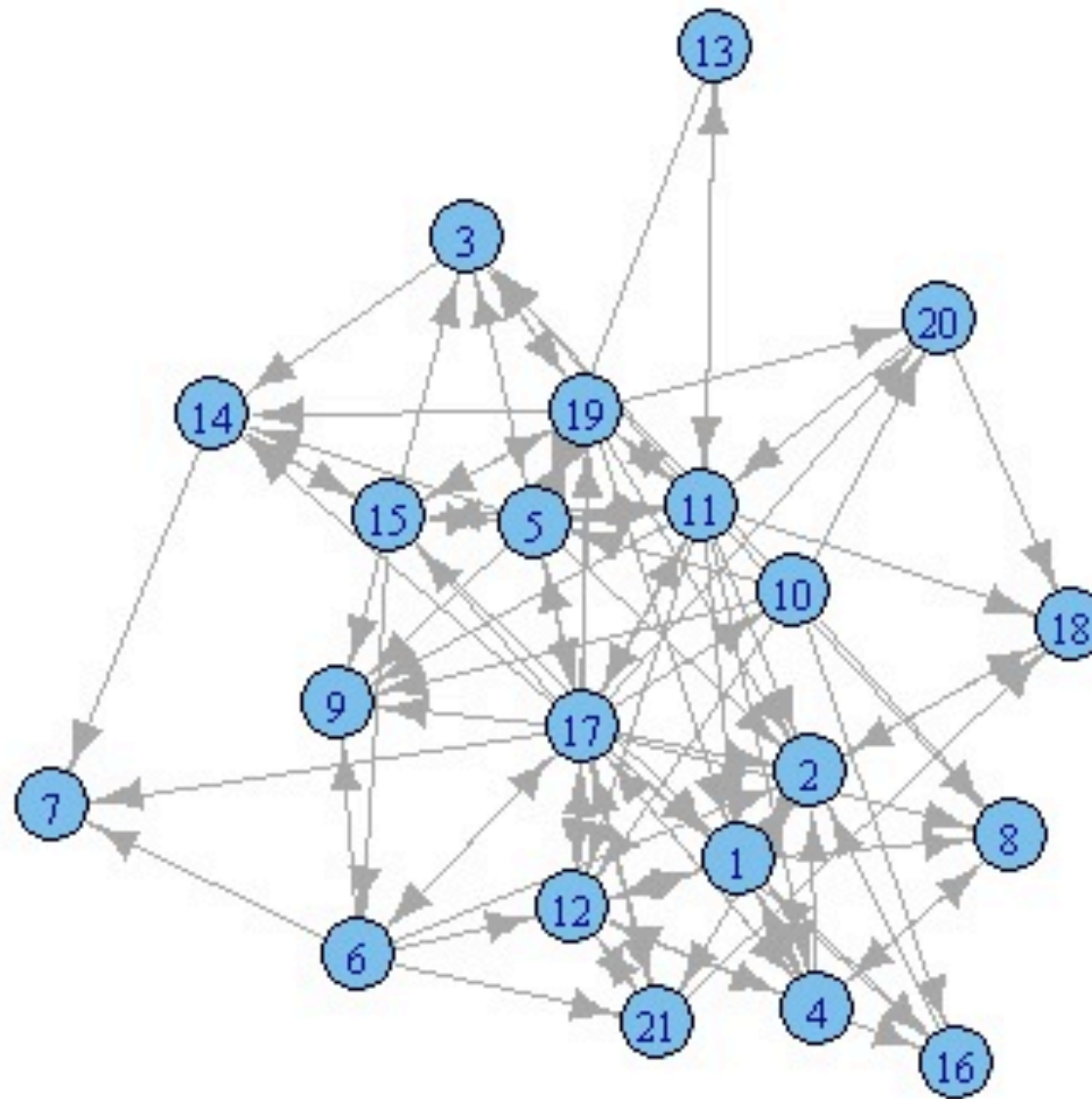
I feel like I have seen this drawing before...where it asks you what is the area of wholes in betwe

2 note(s) 1 ✓

If it's not a mass, don't call it a mass. Just introduce it as the

Zyto et. al.

The NB system: text annotation



The NB system: text annotation

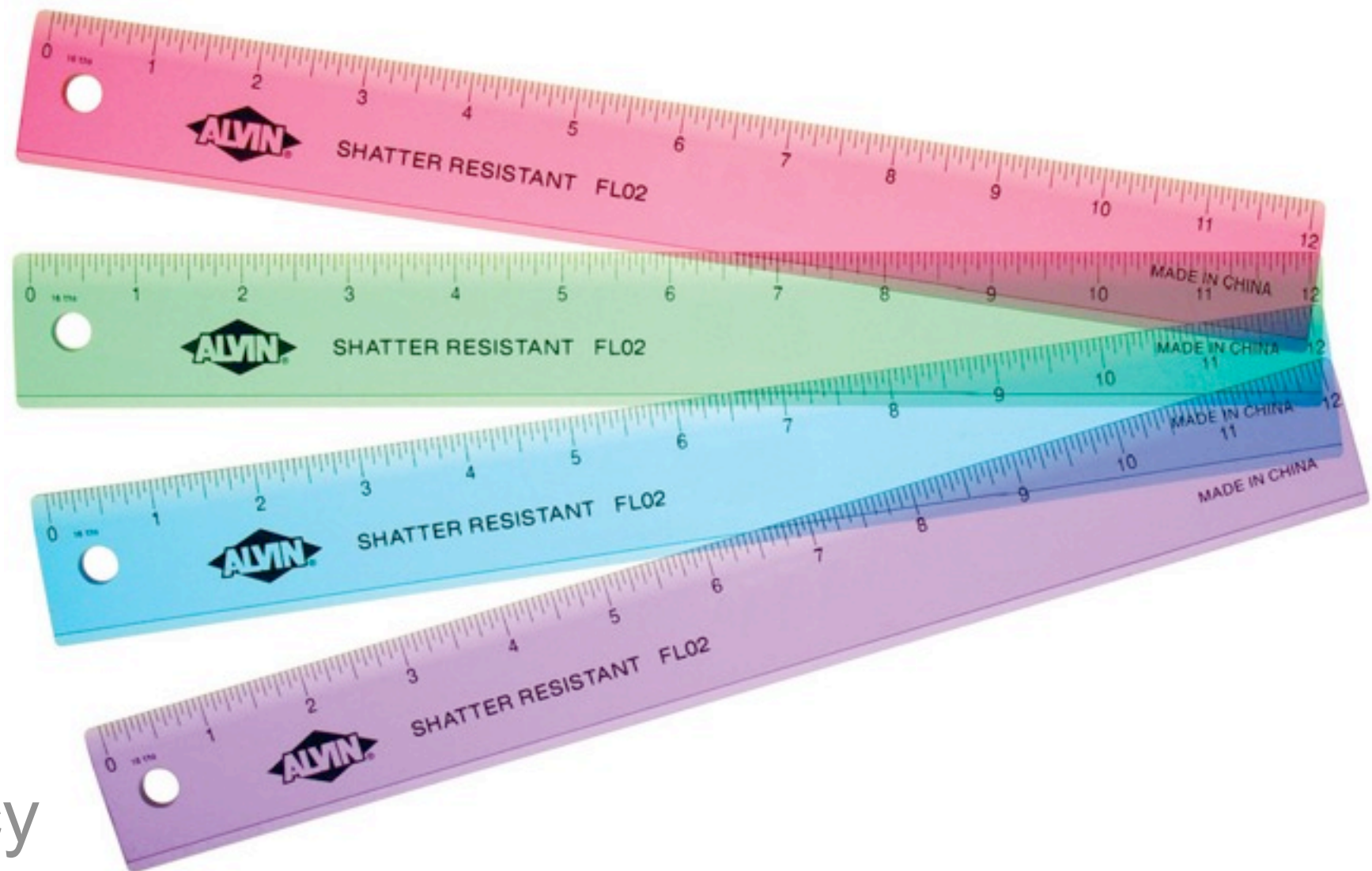
| Score | Description & Criteria |
|-------|-----------------------------------------------------------------------------------------|
| 2 | Demonstrates thorough and thoughtful reading AND insightful interpretation of the text. |
| 1 | Demonstrates reading, but no (or only superficial) interpretation of the text. |
| 0 | Does not demonstrate any thoughtful reading of the text. |

Learning

Metacognition

Interaction

Self-efficacy



Measurement

Interaction

Learning

Learning

Interaction

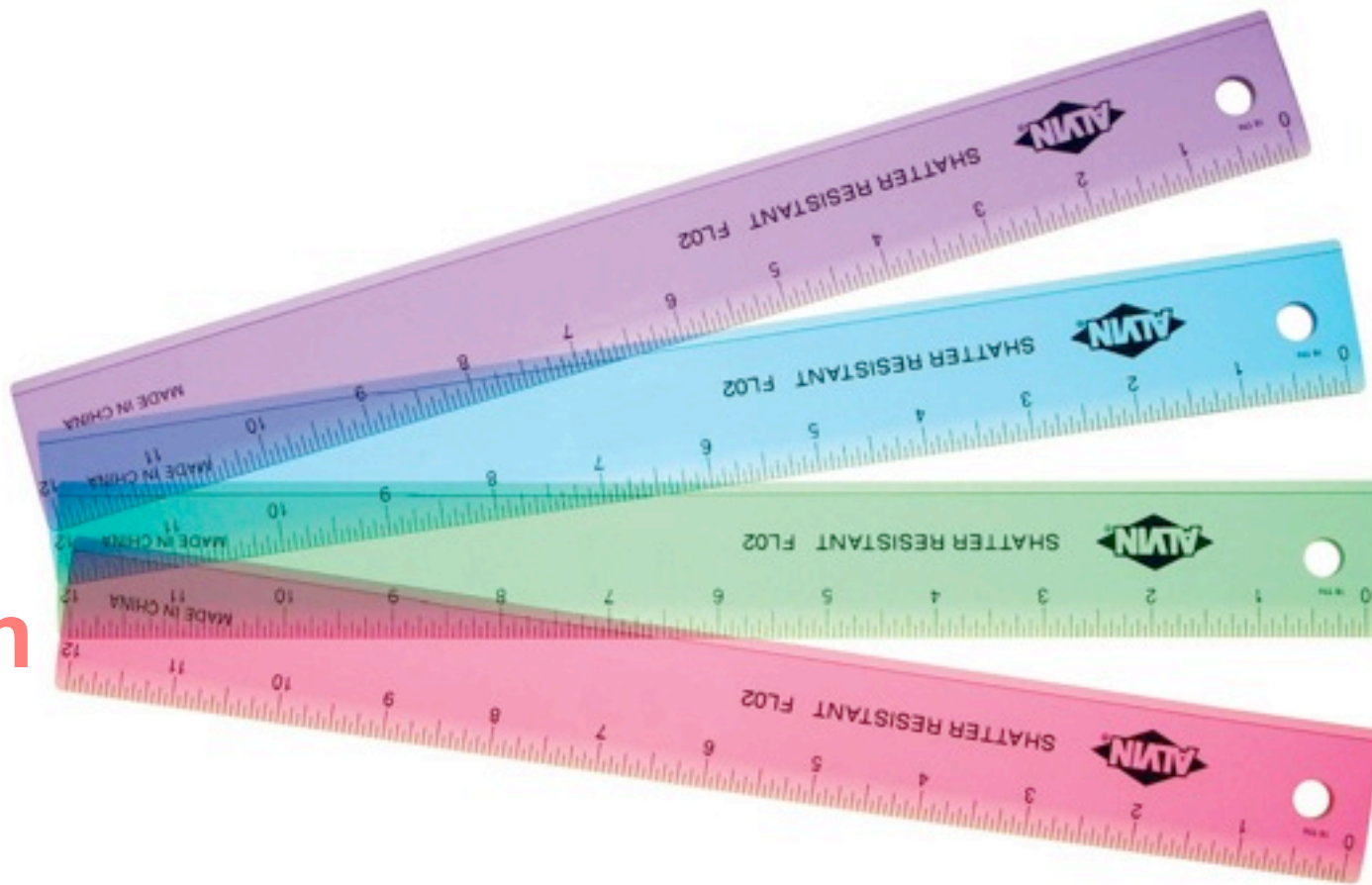
NB annotations

Lens to Learning
videos

Group exams

Social networks

Peer assessments



Measurement

Interaction

Learning

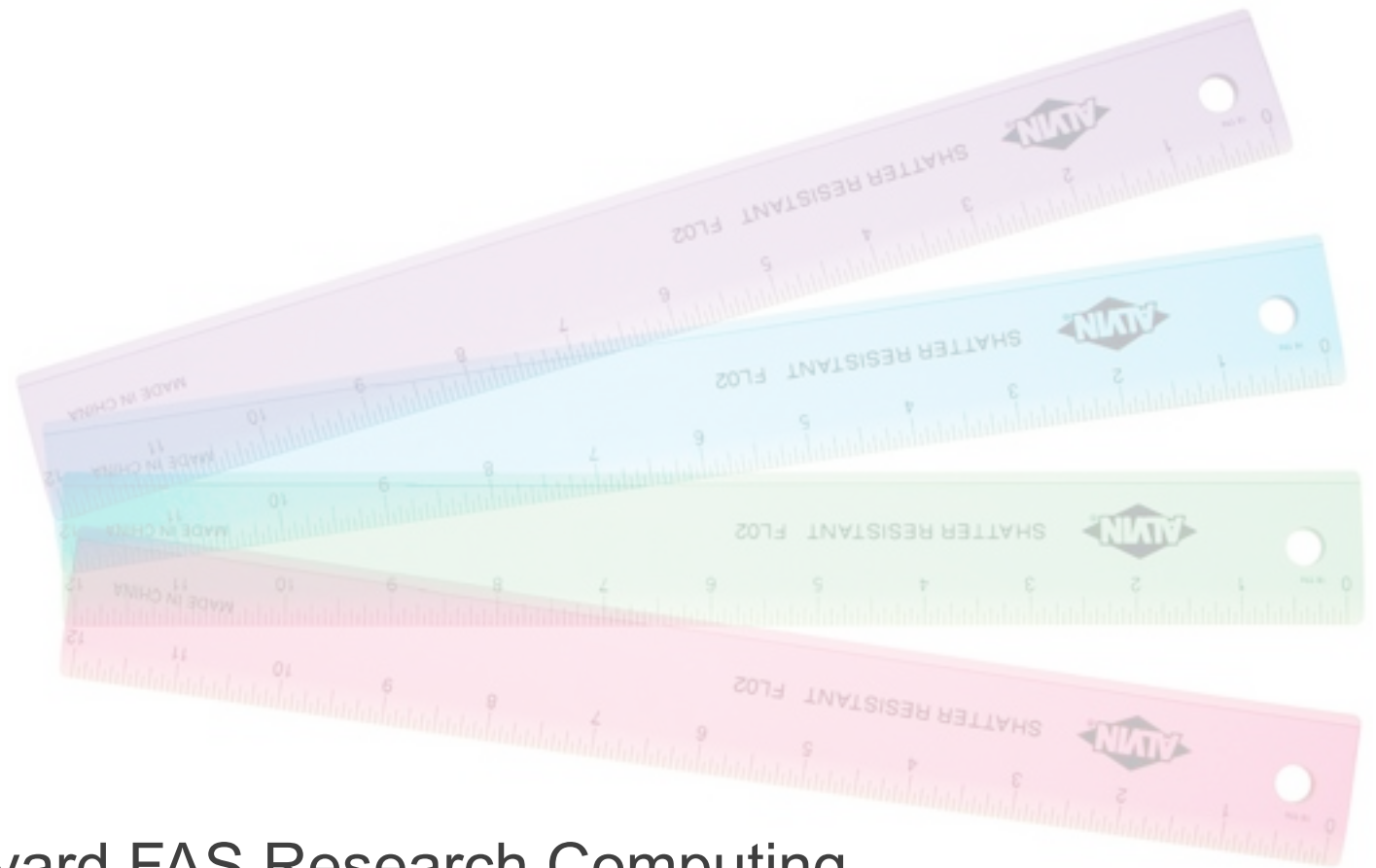
Eric Mazur
Todd Zickler
Rachel Scherr (Seattle Pacific University)

Carolann Koleci
Marinna Madrid
Kelly Miller
Youngjin Seo
Sumalee Waiyarod
Fauzy Wan
Helan Wu
Junehee Yoo

Beth Altringer
NB Team

Mazur Group
John Brunelle and Harvard FAS Research Computing

NSF
Hauser Initiative for Learning and Teaching (HILT)
Harvard School of Engineering and Applied Sciences



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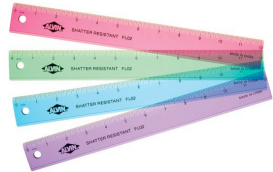
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Sue Borchardt, 2011



http://www.youtube.com/watch?v=Wzs2zXI_aZc&noredirect=1

Helan Wu and Eric Mazur, 2013



<http://www.jerrysartarama.com/images/products/drafting/alvin/0v09319000000-st-01-alvin-shatter-resistant-plastic-ruler.jpg>



<http://oeit-dev.mit.edu/sites/default/files/cscl1.pdf>

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